## **Granite and Marble**

Two types of stone, granite and marble, have been used since ancient times for building, decoration, and sculpture. Granite is used particularly in applications that require great resistance to weathering, such as in monuments, tombstones, road curbings, building facings, and foundations.

Granite is the most common igneous rock (formed from solidified magma), and is found in masses that may extend for tens or hundreds of miles. Granite varies in color and is usually light, with gray predominating, but is also found in pink, yellow, red, and green. The coloration is determined by impurities in granite's most dominant mineral constituents, feldspar and quartz. Other components of granite include muscovite, biotite, pyroxene, and hornblende.

In addition to its pleasing colors, granite has several other advantages: it possesses great strength and durability, has a homogeneous texture, and is not too difficult to work with. Granite is the hardest of all building stones. Disadvantages include its rough surface (which takes on a dirty appearance) and its tendency to spall, or crumble, when directly exposed to fire. Prior to World War II, the widespread use of macadam and concrete for roads, along with the occurrence of the Great Depression of the 1930s (which brought construction to a virtual standstill), caused a downturn in granite production that lasted until the post-war building boom.

Marble, a metamorphic rock usually derived from limestone (in which the calcite has been recrystallized), is a favorite stone of architects because of its beauty and the ease with which it can be worked, as well as the warmth of its polished surfaces. The calcite grains in marble reflect light penetrating through minute cracks in the surface, producing a luminescence. Marble's resistance to abrasion makes it popular for stair treads and flooring. The hardness of marble is 3 on the Mohs scale. Marble has been used in constructing such famous landmarks as the Parthenon in Athens, the Taj Mahal in India, the Lincoln Memorial in Washington D.C., and St. Peter's Basilica in Rome.

Because granite and marble have been extensively used since ancient times, quarrying of the stone was a major industrial activity. Beginning in ancient Egypt, granite or marble quarries were

70

the sites for large construction projects. From these sites, the Egyptians could move blocks weighing up to a million kilograms, hauling them to distant building sites. Most of these quarries were open-faced, though workers could extend tunnels several hundred meters into cliffs to reach inclusions of highquality granite or marble.

After stripping away the top layers of dirt and weathered stone, rows of holes were drilled into the exposed rock with metal bow drills. Wooden wedges were driven tightly into the holes, then doused with water; the wood absorbed the water, which expanded the wedges enough to crack the rock along the line of holes. Since granite is much harder than marble, workers used not only picks but also pounding balls of dolerite (a very hard, coarse basalt) weighing up to 5 kg. Wherever possible, workers took advantage of natural joints in the rock to facilitate easy splitting.

In modern granite-quarrying operations, similar methods are used to split the blocks. Holes, about 2 cm, are drilled 8–13 cm into the rock; inside each hole is placed a set of steel "feathers" (halfround shims) and a wedge plug. The plug is successively driven between the feathers; the force is eventually sufficient to split the rock.

Marble blocks are quarried with the same technique, though the softer marble requires the drilling of deeper holes. Wire saws may also be used—a singlestrand or three-strand wire runs as a belt over the rock, cutting by abrasion as sand and water are fed into it. Explosives are almost never used for marble quarrying because of the danger of shattering the rock.

Masons finished rough marble stones to the proper dimensions at the building site itself, using metal chisels and mallets, squares, plumb bobs, and straightedges. Such tools remained standard until the 1800s, when power machinery and explosives began to be widely used. Mill sawing of marble into slabs is currently done with parallel iron blades fed by sand and water. For finishing, marble may be machined with lathes and carburundum wheels.

In quarrying, at least one half of the marble is wasted because of breakage. The waste marble is crushed or powdered for use in stucco or mosaics. Small spheres made of waste marble chips called "marbles"—were used in 17th-century childrens' games; toy marbles are now usually made out of glass.

Since granite is the hardest and most durable of typical building stones, historically it was the most widely used material for foundations and structural work. The Egyptians used both granite and marble, as well as quarried sandstone and limestone, for their massive constructions. The Romans imported marbles and granites of different colors from various parts of the Empire. They treated marble mainly as a decorative rock, setting it into cement or applying it in slabs to cover brick walls. The Romans also used marble for pavement that was set in geometric patterns or mosaics.

Marble is perhaps most famous, however, for its prevalent use in fine sculpture. Granite, with its salt-and-pepper coloring and irregular glinting from embedded mica and quartz crystals, is much less favored than the fine, nearly translucent white marbles. The practice of sculpting in marble started in ancient Greece and continued through the Renaissance to modern times. Michelangelo, Donatello, and Antonio Canova used marble, as did their ancient predecessors Phidias and Praxiteles.

Typically, a sculptor starts with a block of stone and a pointed chisel, knocking off large chunks that will not be included in the finished piece. Holding the chisel at an acute angle, the sculptor roughs out the approximate shape of the statue, paying close attention to the grain or stratification of the marble, since a misdirected blow could disastrously split the piece. Carving around the stone, the sculptor crisscrosses the direction of each set of trimming. The detail work is completed with successively finer chisels, rasps, files, and abrasives, until the finished sculpture is polished.

Today, unfortunately, the widespread use of fossil fuels in our industrialized society poses a theat—in the form of "acid rain" and other pollution-to marble sculpture and fine stonework displayed outside. Sulfur dioxide in the air transforms calcite on the surface of the marble into gypsum, and the process of recrystallization causes severe damage to the stone. Migrating salts, such as nitrates or chlorides, also harm the internal structure of marble as they travel into deep pores and expand during recrystallization. Many efforts are currently underway to protect and restore classic works of architecture and sculpture.

KEVIN J. ANDERSON

Papers published in the most recent issue of .....

#### **NEW BOOKS FROM** The Materials Research Society

The following books have just been published from the 1992 MRS Fall Meeting in Boston, Massachusetts.

#### Nanophase and Nanocomposite Materials

Editors: S. Komarneni, J.C. Parker,

G.J. Thomas

Covers the synthesis, processing and properties of nanophase and nanocomposite materials, documenting the first symposium of its kind in this rapidly growing field. Topics: nanophase ceramic materials; nanophase metals and alloys; sol-gel nanocomposites; biological and polymer nanocomposites; nanocomposites from layered and cage structures; nanocomposites through multilayers; nanocomposites for various applications. 1993, hardcover, 66 papers, 457 pages.

ISBN: 1-55899-181-6 Code: 286-B \$55.00 MRS Members \$63.00 U.S. List \$72.00 Foreign

#### **Silicon Nitride Ceramics**

Scientific and Technological Advances

Editors: I-W. Chen, P.F. Becher, M. Mitomo, G. Petzow, T-S. Yen

In the past decade, silicon nitride ceramics have been developed for a number of applications, including those with guite stringent reliability and economic requirements (e.g., automobile turbocharger rotors). These applications serve as graphic examples of the real advances that have been accomplished. The progression of these materials has been based on advances in the fundamental understanding of processing-phase equilibria-microstructure-mechanical properties relationships coupled with the development of the manufacturing and engineering practice capable of producing highly reliable components. These advances are reflected in this proceedings volume. 1993, hardcover, 67 papers, 549 pages. ISBN; 1-55899-182-4 Code: 287-B \$55.00 MRS Members \$63.00 U.S. List \$72.,00 Foreign

#### **ORDER FROM:**

Materials Research Society 9800 McKnight Road Pittsburgh, PA 15237 Phone (412) 367-3012 FAX (412) 367-4373

In Europe, Africa, and the Middle East contact:

Clarke Associates-Europe, Ltd, 13a Small Street, Bristol BS1 1DE, England; phone 0272 268864, FAX 0272 226437



#### Contents of Volume 16, number 5, May 1993

Mirror-translational (011) twin boundary in modulated Bi<sub>2</sub>Sr<sub>2</sub> CaCu<sub>2</sub>O<sub>8+δ</sub> X.C. Wu, W.S. Wei and T.S. Shi

Formation of Li<sub>x</sub>Ni<sub>1-x</sub>O solid solution from Ni/Li<sub>2</sub>CO<sub>3</sub> mixtures E. Antolini

Structural and electrical properties of PZT (La, K) ceramics K.L. Yadav and R.N.P. Choudhary

Fractal morphology of cement foams G.N. Karam and T.D. Tonyan

Structural analysis and space model of the glassy alloy Cu<sub>0.10</sub>As<sub>0.45</sub>Te<sub>0.45</sub> by X-ray diffraction

C. Wagner, J. Vázquez, P. Villares and R. Jiménez-Garay

Vibrational spectral/structural changes from the hydrolysis/polycondensation of methylmodified silicates. III. IR spectral comparisons for condensates from the binary methyltrimethoxysilane/tetramethoxysilane system

C.A. Capozzi, R.A. Condrate Sr. and L.D. Pye

Time and composition dependence of the spectral shape of the X-ray Ce-L<sub>III</sub> absorption in the  $R_{2-x}$  Ce<sub>x</sub>CuO<sub>4- $\delta$ </sub> superconductors

L.D. Finkelstein, N.N. Efremova, A.V. Postnikov, E.Z. Kurmaev, N.A. Babushkina and G. Balakrishnan

Training and aging effects on shape memory behavior in a two-phase NiAlFe alloy J.H. Yang and C.M. Wayman

Diffuse phase transition in solid solutions of Pb(Mg  $_{1}$  Zn  $_{1}$ Nb  $_{1}$ )O<sub>3- $\delta$ </sub> and PbTiO<sub>3</sub> S. Sharma, R. Sati, R.N.P. Choudhary and T.P. Sinha

Colloidal processing of silicon carbide Y. Hirata, S. Yamada and Y. Fukushige

An in situ TEM observation on the thermal stability of a nanocrystalline FeBSi alloy H.Y. Tong, B.Z. Ding, H.G. Jiang, Z.Q. Hu, L. Dong and Q. Zhou

Optical and electrical properties of Sc<sub>0.2</sub>Yb<sub>0.8</sub>As and Sc<sub>0.7</sub> Er<sub>0.8</sub>As semimetallic compounds lattice matched to GaAs

D. Neveux, J.C. Simon, Y. Ballini and A. Guivarc'h

The effect of a plasma-sprayed hydroxyapatite coating on the fatigue properties of Ti-6Al-4V

S.L. Evans and P.J. Gregson

For a free sample copy of this issue contact:

#### **ELSEVIER SCIENCE PUBLISHERS,**

PO Box 103, 1000 AC Amsterdam, The Netherlands Tel: 31 20 5862819, Fax: 31 20 5862580

For information on a personal subscription contact the MRS by using the Reader Service Card.

Circle No. 21 on Reader Service Card.

# CALL FOR PAPERS



**ABSTRACT DEADLINE: JUNE 20,1993** 

### **SYMPOSIA**

#### AA: ATOMIC SCALE IMPERFECTIONS IN MATERIALS - R.W. BALLUFFI FEST

David N. Seidman, Northwestern University, (708) 491-4391, Fax (708) 491-7820; Richard W. Siegel, Argonne National Laboratory, (708) 252-4963, Fax (708) 252-4798; Paul D. Bristowe, Massachusetts Institute of Technology, (617) 253-3326, Fax (617) 258-7874

#### A: MATERIALS SYNTHESIS AND PROCESSING USING ION BEAMS

**Robert J. Culbertson**, Arizona State University, (602) 965-0945, Fax (602) 965-7954; **O.W. Holland**, Oak Ridge National Laboratory, (615) 576-2502, Fax (615) 576-8135; **Kevin S. Jones**, University of Florida, (904) 392-9872, Fax (904) 392-6359; **Karen Maex**, IMEC, Belgium, (32) 16-281-358, Fax (32) 16-281-214

#### **B: MECHANISMS OF THIN FILM EVOLUTION**

Steven M. Yalisove, University of Michigan, (313) 764-4346, Fax (313) 763-4788, SMY@CAEN.ENGIN.UMICH.EDU; Carl V. Thompson, Massachusetts Institute of Technology, (617) 253-7652, Fax (617) 258-6749, CTHOMP@MTL.MIT.EDU; David J. Eaglesham, AT&T Bell Laboratories, (908) 582-3768, Fax (908) 582-4228, DAVE@PHYSICS.ATT.COM

#### CA: INTERFACE CONTROL OF ELECTRICAL, CHEMICAL, AND MECHANICAL PROPERTIES

S.P. Murarka, Rensselaer Polytechnic Institute, (518) 276-2978, Fax (518) 276-8761; K. Rose, Rensselaer Polytechnic Institute, (518) 276-2981, Fax (518) 276-8761; T. Ohmi, Tohoku University, Japan, (81) 22-224-2649, Fax (81) 22-224-2549; T. Seidel, SEMATECH, (512) 356-3350, Fax (512) 356-3618

#### **CB: DEFECT-INTERFACE INTERACTIONS**

E.P. Kvam, Purdue University, (317) 494-4097, Fax (317) 494-1204; M.J. Mills, Sandia National Laboratory, (510) 294-3018, Fax (510) 294-3231; Alexander H. King, SUNY at Stony Brook, (516) 632-8499, Fax (516) 632-8052; T.D. Sands, Bellcore, (908) 758 -2945, Fax (908) 758-4372; V. Vitek, University of Pennsylvania, (215) 898-7883, Fax (215) 898-8296

#### D: SILICIDES, GERMANIDES, AND THEIR INTERFACES

**R.W. Fathauer**, California Institute of Technology, (818) 354-4962, Fax (818) 393-4540; Internet: FATHAUER%VAXEB@JPL.NASA.GOV; **Siegfried Mantl**, Institute of Thin Film and Ion Technology (ISI), Germany, (49) 2461-613643, Fax (49) 2461-614673; **Leo Schowalter**, Rennselaer Polytechnic Institute, (518) 276-6435, Fax (518) 276-6680, Internet: SCHOWALT@UNIX.CIE.RPI.EDU; **K.N. Tu**, IBM T.J. Watson Research Center, (914) 945-1602, Fax (914) 945-4407, Bitnet: KNTU@YKTVMV

#### E: CRYSTALLIZATION AND RELATED PHENOM-ENA IN AMORPHOUS MATERIALS - CERAMICS, METALS, POLYMERS, AND SEMICONDUCTORS

Matthew Libera, Stevens Institute of Technology, (201) 216-5259, Fax (201) 216-8306, PTF-MLIBERA@VAXC.STEVENS-TECH.EDU; Peggy Cebe, Massachusetts Institute of Technology, (617) 253-8638, Fax (617) 258-7874; Tony E. Haynes, Oak Ridge National Laboratory, (615) 576-2858, Fax (615) 576-6720; Jim Dickinson, Corning Incorporated, (607) 974-3659, Fax (607) 974-3675

#### F: HIGH-TEMPERATURE SILICIDES AND REFRACTORY ALLOYS

B.P. Bewlay, General Electric Company, (518) 387-6121, Fax (518) 387-7495; C.L. Briant, General Electric Company, (518) 387-5765, Fax (518) 387-7495; H.A. Lipsitt, Wright State University, (513) 873-5095, Fax (513) 873-5009; J.J. Petrovic, Los Alamos National Laboratory, (505) 667-0125, Fax (505) 665-3363; A.K. Vasudevan, Office of Naval Research, (703) 696-4715, Fax (703) 696-0308

#### G: FULLERENES AND RELATED MATERIALS

**Robert N. Compton**, Oak Ridge National Laboratory, (615) 574-6233, Fax (615) 576-4407; **A.F. Hebard**, AT&T Bell Laboratories, (908) 582-4944, Fax (908) 582-2451; **H. Nori Shinohara**, Mi'e University, Japan, (81) 592-31-2252; John H. Weaver, University of Minnesota, (612) 625-6548, Fax (612) 625-6043

#### H: SUPERCONDUCTIVITY - MATERIALS AND PROPERTIES

Kenneth W. Lay, General Electric Company, (518) 387-6147, Fax (518) 387-7495; Martin P. Maley, Los Alamos National Laboratory, (505) 665-0189, Fax (505) 665-3164; H. Richard Kerchner, Oak Ridge National Laboratory, (615) 574-6270, Fax (615) 574-6263; Xiao Xing Xi, University of Maryland, (301) 405-6175, Fax (301) 314-9541

#### I: DEVELOPING MATERIALS PROCESSES FOR FACTORIES

B.K. Keramati, General Electric Company, (518) 387-5456, Fax (518) 387-6232; J. Malas, USAF Wright Laboratories, (513) 255-8787, Fax (513) 476-7995; Court Skinner, National Semiconductor, (408) 721-7420, Fax (408) 736-8503; Henry J. Rack, Clemson University, (803) 656-5636, Fax (803) 656-4435

#### J: ELECTRONIC PACKAGING MATERIALS SCIENCE VII

Roger Pollak, IBM T.J. Watson Research Center, (914) 945-1446, Fax (914) 945-4002, POLLAK@WATSON.IBM.COM; Hisao Yamada, Cerone, Inc., (602) 451-0453, Fax (602) 496-5060; Peter Børgesen, Cornell University, (607) 255-7218/ 255-4045, Fax (607) 255-6575; Klavs F. Jensen, Massachusetts Institute of Technology, (617) 253-4589, Fax (617) 258-8224, KFJENSEN@ATHENA.MIT.EDU

#### K: DIAGNOSTIC TECHNIQUES FOR SEMICON-DUCTOR MATERIALS PROCESSING

Orest J. Glembocki, U.S. Naval Research Laboratory, (202) 767-2799, Fax (202) 767-4290, glembocki@estd.nrl.navy.mil; Stella W. Pang, University of Michigan, (313) 936-2962, Fax (313) 747-1781, pang@eecs.umich.edu; Fred H. Pollak, Brooklyn College, (718) 951-5356, Fax (718) 951-4871; Graydon Larrabee, (214) 239-0008, Fax (214) 980-7224; Gabriel M. Crean, University College, Ireland, (353) 21-276871, Ext. 2721, Fax (353) 21-270271

#### L: DEFECTS IN ADVANCED SEMICONDUCTORS -PHYSICS AND APPLICATIONS

M.O. Manasreh, Wright Laboratory, (513) 255-4736, Fax (513) 255-3374; H.J. von Bardeleben, Groupe de Physique des Solides, France; (33) 1-4427-7948, Fax (33) 1-4427-7948; G.S. Pomrenke, Air Force Office of Scientific Research, (202) 767-4931, Fax (202) 767-4986; M. Lannoo, Institut Supériure d'Electronique du Nord, France, (33) 20-306220, Fax (33) 20-545666; Evelyn L. Hu, University of California, Santa Barbara, (805) 893-2368, Fax (805) 893-3262; D.N. Talwar, Indiana University of Pennsylvania, (412) 357-2370, Fax (412) 357-5700

#### M: GROWTH, PROCESSING AND CHARACTERIZA-TION OF SEMICONDUCTOR HETERO-STRUCTURES

Godfrey Gumbs, City University of New York, (212) 650-3935, Fax (212) 772-5390, GOGHC@CUNYVM.BITNET; Bernard Weiss, University of Surrey, U.K., (44) 483-509128, Fax (44) 483-34139, B.WEISS@EE.SURREY.AC.UK; Serge Luryi, AT&T Bell Laboratories, (908) 582-6614, Fax (908) 582-2043, SERGE@ BARKEEP.ATT.COM; Gary W. Wicks, University of Rochester, (716) 275-4867, Fax (716) 244-4936, WICKS@OPTICS.ROCHESTER.EDU

#### N: COVALENT CERAMICS II: NON-OXIDES

Andrew R. Barron, Harvard University, (617) 495-5008, Fax (617) 496-7402; Michael A. Fury, (914) 894-3248, Fax (914) 892-6399; Gary S. Fischman, Alfred University, ( 607) 871-2284, Fax (607) 871-2317; Aloysius F. Hepp, NASA Lewis Research Center, (216) 433-3835, Fax (216) 433-6106

#### O: COMPLEX FLUIDS

Scott Milner, Exxon Research and Engineering Company, (908) 730-2309, Fax (908) 730-3042, Internet: STMILNE@ERENJ.COM; Eric Kaler, University of Delaware, (302) 831-3553; Fax (302) 831-1048; Mark Robbins, Johns Hopkins University, (410) 516-7204, Fax (410) 516-7239

#### P: DISORDERED MATERIALS - FRACTALS, SCALING, AND DYNAMICS

Gary S. Grest, Exxon Research and Engineering Company, (908) 730-2881, Fax (908) 730-3042, GSGREST@ERENJ.COM; David G. Grier, The University of Chicago, (312) 702-9176, Fax (312) 702-5863, GRIER@FAFNIR.UCHICAGO.EDU; Po-zen Wong, University of Massachusetts, (413) 545-3288, Fax (413) 545-1691, PZWONG@PHAST.UMASS.EDU

#### Q: ELECTRICAL, OPTICAL, AND MAGNETIC PROPERTIES OF ORGANIC SOLID STATE MATERIALS

Anthony F. Garito, University of Pennsylvania, (215) 898-5810, Fax (215) 898-2010; Larry R. Dalton, University of Southern California, (213) 740-8768; Fax (213) 740-2701 ; Alex K-Y. Jen, EniChem America, Inc., (908) 422-3709, Fax (908) 422-3783; Charles Lee, AFOSR, (202) 767-4963, Fax (202) 767-4961

#### R: NEW MATERIALS FOR ADVANCED SOLID STATE LASERS

https://doi.org/10.1557/S0883769400047199 Published online by Cambridge University Press

**Bruce H.T. Chai**, University of Central Florida, (407) 658-6847, Fax (407) 658-6880; **Steven A. Payne**, Lawrence Livermore National Laboratory, (510) 423-0570, Fax (510) 423-6506; **Tso Yee Fan**, MIT Lincoln Laboratory, (617) 276-6701, Fax (617) 276-6721; **A. Cassanho**, Massachusetts Institute of Technology, (617) 253-3696, Fax (617) 253-2112; **Toomas H. Allik**, SAIC, (703) 704-3265, Fax (703) 704-1752



#### S: BIOMOLECULAR MATERIALS BY DESIGN

Hagan Bayley, Worcester Foundation for Experimental Biology, (508) 842-9146, Fax (508) 842-9632, BAYLEY@WFEB2; David Kaplan, U.S. Army Natick RD&E Center, (508) 651-5525, Fax (508) 651-5521, DKAPLAN@NATICK-EMH1.ARMY.MIL; Manuel Navia, Vertex Pharmaceuticals, (617) 576-3111, Fax (617) 576-2109, NAVIA@VPHARM.COM

#### T: BIOMATERIALS FOR DRUG AND CELL DELIVERY

Antonios G. Mikos, Rice University, (713) 285-5355, Fax (713) 285-5353; Howard Bernstein, Enzytech, Inc., (617) 252-0001, Fax (617) 252-0916; Regina Murphy, University of Wisconsin, (608) 262-1587, Fax (608) 262-5434; Nicholas A. Peppas, Purdue University, (317) 494-7944, Fax (317) 494-0805

#### U: DETERMINING NANOSCALE PHYSICAL PROPERTIES OF MATERIALS BY MICROSCOPY AND SPECTROSCOPY

Mehmet Sarikaya, University of Washington, (206) 543-0724, Fax (206) 543-6381, SARIKAYA@CARSON.U.WASHINGTON.EDU; Michael Isaacson, Cornell University, (607) 255-4302, Fax (607) 255-7658, ISACRNLMSC2.BITNET; H. Kumar Wickramasighe, IBM T.J. Watson Research Center, (914) 945-3794, Fax (914) 945-2141, WICKRAM@IBM.COM

#### V: SCIENTIFIC BASIS FOR NUCLEAR WASTE MANAGEMENT XVII

Aaron Barkatt, The Catholic University of America, (202) 319-5522, Fax (202) 319-4469; Rich Van Konynenburg, Lawrence Livermore National Laboratory, (510) 422-0456, Fax (510) 422-6892

#### W: GAS-PHASE AND SURFACE CHEMISTRY IN ELECTRONIC MATERIALS PROCESSING

T.J. Mountziaris, SUNY at Buffalo, (716) 645-2109, Fax (716) 645-3822, CMELAKIS@UBVMS.CC.BUFFALO.EDU; Frank T.J. Smith, Loral Infrared Imaging Systems, (617) 863-4079, Fax (617) 863-3638; Phillip R. Westmoreland, National Institute of Standards and Technology, (301) 975-2602, Fax (301) 869-5924, WESTM@ECS.UMASS.EDU; Gustavo R. Paz-Pujalt, Eastman Kodak Company, (716) 477-6803, Fax (716) 477-4947

#### X: FRONTIERS OF MATERIALS RESEARCH

Philippe M. Fauchet, University of Rochester, (716) 275-1487, Fax (716) 275-2073; David B. Poker, Oak Ridge National Laboratory, (615) 576-8827, Fax (615) 576-6720; Alan I. Taub, General Electric Company, (518) 387-6234, Fax (518) 387-6232; Rustum Roy, Pennsylvania State University, (814) 865-3421, Fax (814) 865-2326

#### Y: METAL-ORGANIC CHEMICAL VAPOR DEPOSITION OF ELECTRONIC CERAMICS

Seshu B. Desu, Virginia Polytechnic Institute, (703) 231-6820, Fax (703) 231-8919; Bruce W. Wessels, Northwestern University, (708) 491-3219, Fax (708) 491-7820; David B. Beach, IBM T.J. Watson Research Center, (914) 945-3956, Fax (914) 945-2141; Suleyman Gokoglu, NASA Lewis Research Center, (216) 433-5499, Fax (216) 433-8660

For general meeting information, contact: MATERIALS RESEARCH SOCIETY 9800 McKnight Road, Pittsburgh, PA 15237, FAX (412) 367-4373

### **MEETING CHAIRS**

Philippe M. Fauchet, Department of Electrical Engineering, University of Rochester, Computer Studies Building, Room 514, Rochester, NY 14627, (716) 275-1487, Fax (716) 275-2073

David B. Poker, Oak Ridge National Laboratory, P.O. Box 2008, MS 6048, Oak Ridge , TN 37831, (615) 576-8827, Fax (615) 576-6720

Alan I. Taub, Corporate Research and Development, General Electric Company, P.O. Box 8, K-1 203MB, Schenectady, NY 12301, (518) 387-6234, Fax (518) 387-6232