## TEM and EELS Study of Dislocation Loops in Reduced BaTiO<sub>3</sub>

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To avoid the oxidation of inner electrodes, Ni-BaTiO<sub>3</sub> multilayer ceramic capacitors (MLCCs) are fired in reducing atmosphere. Consequently, the BaTiO<sub>3</sub> becomes oxygen deficient. Structural defects, such as dislocation loops, are often observed in the reduced BaTiO<sub>3</sub> perovskite lattice. A few TEM studies have shown that the dislocation loops (DL) mostly lie on {100} planes with Burgers vectors of type <100>.<sup>1-2</sup> In this work, transmission electron microscopy (TEM) coupled with electron energy-loss spectroscopy (EELS) have been used to study chemical characteristics of the dislocation loops.

The Ni-BaTiO<sub>3</sub> MLCCs were fabricated by co-firing Ni internal electrodes with BaTiO<sub>3</sub> dielectric layers at ~ 1300 °C in a PO<sub>2</sub> of ~  $10^{-12}$  atm. TEM observations of thinned specimens were performed using a JEOL 2010F transmission electron microscope equipped with a field-emission gun operated at 200 kV. Electron energy-loss spectra (EELS) were acquired with a Gatan Enfina parallel electron energy-loss spectrometer attached to the JEOL 2010F microscope. EEL spectra were recorded in TEM-diffraction mode with a collection angle of 14 mrad, and analyzed with Gatan Digital Micrograph software.

Figure 1 (a) shows a typical bright-field TEM image obtained from co-fired BaTiO<sub>3</sub> grains. A high density of dislocation loops is observed in BaTiO<sub>3</sub>. Since image contrast of the dislocation loops is highly sensitive to crystal orientation, the dislocation loops are observed in some grains although the dislocation loops are equally present in other grains. Figure 1(b) shows the EEL spectra obtained from the dislocation loop and just away from the defects, as noted by the circles in figure 1 (b). The fine structures of the Ti L<sub>2,3</sub> and O K edges are shown in figure 1 (b). In the EEL spectrum obtained from the regions away from the dislocation loops, a clear split in the Ti L<sub>2,3</sub> edges (t<sub>2g</sub> and e<sub>g</sub>), indicative of Ti<sup>4+</sup>, is observed. The fine structures of O K edges are noted by capital letters A, B, C, and D. In the spectrum taken from the dislocation loop, characteristics of the fine structures of the Ti L<sub>2,3</sub> and O K edges become faint. Such evolutions in Ti L<sub>2,3</sub> and O K edges is associated with local oxygen deficiency.<sup>3-5</sup>

Shown in figure 2 is the EEL spectrum obtained with energy dispersion of 0.5 eV/channel. Atomic ratio of Ti and Ba were calculated through quantification of the EEL spectra obtained from the dislocation loops and away from the dislocation loops with the integration window with of 50 eV. The k-factor measured from a standard BaTiO<sub>3</sub> sample was used. More than 20 pairs of EEL spectra were analyzed and the ratios,  $C_{Ti}/C_{Ba}$ , can be classified as into 3 groups:  $(C_{Ti}/C_{Ba})_{DL} < (C_{Ti}/C_{Ba})_{away}$  DL,  $(C_{Ti}/C_{Ba})_{DL} \approx (C_{Ti}/C_{Ba})_{away}$  DL,  $(C_{Ti}/C_{Ba})_{DL} \approx (C_{Ti}/C_{Ba})_{away}$  DL, and  $(C_{Ti}/C_{Ba})_{DL} > (C_{Ti}/C_{Ba})_{away}$  DL, implying that the dislocation loops in reduced BaTiO<sub>3</sub> may be generated through three possible Schottky reactions: complete Schottky reaction (null  $\rightarrow V_{Ba}^{"} + V_{Ti}^{"'} + 3 V_{O}^{"}$ ) and partial Schottky reactions (null  $\rightarrow V_{Ba}^{"} + V_{O}^{"}$ , null  $\rightarrow V_{Ti}^{"'} + 2V_{O}^{"}$ ).<sup>6</sup> It is suggested that the formation of dislocation loops in reduced BaTiO<sub>3</sub> should be associated with local clustering of oxygen vacancies and/or metallic vacancies.

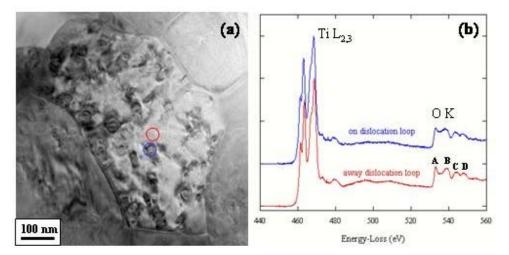


Figure 1. (a) BF TEM image of dislocation loops in reduced BaTiO<sub>3</sub> grains, (b) EEL spectra obtained from dislocation loops and just away from the dislocation loops.

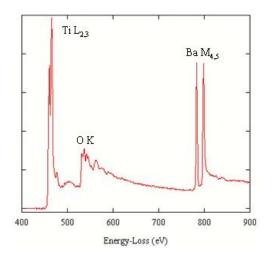


Figure 2. EEL spectrum obtained with energy dispersion of 0.5 eV/channel showing the Ti  $L_{2,3}$ , O K and Ba  $M_{4,5}$  edges. The integration window width of 50 eV was used to quantify Ti and Ba atomic ratio.

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