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## Aims and Scope

The *British Journal of Nutrition* is an international, peer-reviewed journal publishing original papers, review articles, short communications and technical notes on human and clinical nutrition, general nutrition, and animal nutrition. Correspondence is encouraged in a nutrition discussion forum. The Journal recognizes the multidisciplinary nature of nutritional science and encourages the submission of material from all of the specialities involved in research and clinical practice. The Journal also publishes supplements on topics of particular interest.

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# THE BRITISH JOURNAL OF NUTRITION

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(Revised January 1997)

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**Units.** Results should be presented in metric units according to the International System of Units (see *Quantities, Units, and Symbols*. London: The Royal Society, 1971, and *Metric Units, Conversion Factors and Nomenclature in Nutritional and Food Sciences*. London: The Royal Society, 1972—as reproduced in *Proceedings of the Nutrition Society* (1972) **31**, 239–247).

Energy measurements should be expressed in joules.

For substances of known molecular weight, e.g. glucose, urea, Ca, Na, Fe, K, P, values should be expressed as mol/l; for substances of indeterminate molecular weights, e.g. phospholipids, proteins, and for trace elements, e.g. Cu, Zn, then g/l should be used.

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Data from individual replicates should not be given for large experiments, but may be given for small studies. The methods of statistical analysis used should be de-

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Comparisons between means can be made by using either confidence intervals or significance tests; both are based on basic measures of variability. The most appropriate of such measures is usually the standard error of a difference between means (SED), or the standard errors of the means (SE or SEM) when these vary between means. The standard deviation (SD) is more useful only when there is specific interest in the variability of individual values. The degrees of freedom associated with SED, SEM or SD should also be stated. When presenting results, forms such as 'mean 3.51 (SE 0.67)  $\mu\text{mol}$ ' should be adopted, rather than the cryptic use of the notation ' $\pm$ '. The number of decimal places quoted should be sufficient but not excessive.

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Where the method of analysis is unusual, or if the experimental design is at all complex, further details (e.g. experimental plan, raw data, confirmation of assumptions, analysis of variance tables, etc.) should be submitted in an accompanying document. Analyses which frequently cause difficulties are change-over designs and other studies with sequential observations on the same experimental unit, studies where variation differs substantially between treatment groups, and studies with a factorial treatment structure or with increasing levels (e.g. doses) of the same treatment factor.

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**Nomenclature of Vitamins.** Most of the names for vitamins and related compounds that are accepted by the Editors are those recommended by the IUNS Committee on Nomenclature. See *Nutrition Abstracts and Reviews A* (1978), **48**, 831-835.

Acceptable Name	Other names*
<i>Vitamin A</i>	
Retinol	Vitamin A <sub>1</sub>
Retinaldehyde, retinal	Retinene
Retinoic acid (all-trans or 13-cis)	Vitamin A <sub>1</sub> acid
3-Dehydroretinol	Vitamin A <sub>2</sub>
<i>Vitamin D</i>	
Ergocalciferol, ercalciol	Vitamin D <sub>2</sub> calciferol
Cholecalciferol, calciol	Vitamin D <sub>3</sub>
<i>Vitamin E</i>	
α-, β- and γ-tocopherols plus tocotrienols	
<i>Vitamin K</i>	
Phylloquinone	Vitamin K <sub>1</sub>
Menaquinone-n (MK-n)†	Vitamin K <sub>2</sub>
Menadione	Vitamin K <sub>3</sub> , menaquinone, menaphthone
<i>Vitamin B<sub>1</sub></i>	
Thiamin	Aneurin(e), thiamine
<i>Vitamin B<sub>2</sub></i>	
Riboflavin	Vitamin G, riboflavin, lactoflavin
Nicotinic acid or niacin	Nicotinic acid
<i>Niacin</i>	
Nicotinamide	Vitamin PP
Nicotinic acid	
<i>Folate or Folic Acid</i>	
Pteroyl(mono)glutamic acid	Folacin, vitamin B <sub>C</sub> or M
<i>Vitamin B<sub>6</sub></i>	
Pyridoxine	Pyridoxol
Pyridoxal	
Pyridoxamine	
<i>Vitamin B<sub>12</sub></i>	
Cyanocobalamin	
Hydroxocobalamin	Vitamin B <sub>12a</sub> or B <sub>12b</sub>



Aquocobalamin	
Methylcobalamin	
Adenosylcobalamin	
Inositol	
Myoinositol	Meso-inositol
Choline	
Panthenic acid	
Biotin	Vitamin H
Vitamin C	
Ascorbic acid	
Dehydroascorbic acid	

\* Including some names which are still in use elsewhere, but are not used by the *British Journal of Nutrition*.

† Details of the nomenclature for these and other naturally occurring quinones should follow the Tentative Rules of the IUPAC-IUB Commission on Biochemical Nomenclature (see *European Journal of Biochemistry* (1975), 53, 15–18).

**Generic descriptors.** The terms **vitamin A**, **vitamin C** and **vitamin D** may still be used where appropriate for example in phrases such as 'vitamin A deficiency', 'vitamin D activity'.

**Vitamin E.** The term **vitamin E** should be used as the descriptor for all tocol and tocotrienol derivatives exhibiting qualitatively the biological activity of  $\alpha$ -tocopherol. The term **tocopherols** should be used as the generic descriptor for all methyl tocols. Thus, the term tocopherol is not synonymous with the term **Vitamin E**.

**Vitamin K.** The term **vitamin K** should be used as the generic descriptor for 2-methyl-1,4-naphthoquinone (menaphthone) and all derivatives exhibiting qualitatively the biological activity of phyloquinone (phytylmenaquinone).

**Niacin.** The term **niacin** should be used as the generic descriptor for pyridine 3-carboxylic acid and derivatives exhibiting qualitatively the biological activity of nicotinamide.

**Vitamin B<sub>6</sub>.** The term **vitamin B<sub>6</sub>** should be used as the generic descriptor for all 2-methylpyridine derivatives exhibiting qualitatively the biological activity of pyridoxine.

**Folate.** Due to the wide range of carbon-substituted, unsubstituted, oxidized, reduced and mono- or poly-glutamyl side-chain derivatives of pteroylmonoglutamic acid which exist in nature, it is not possible to provide a complete list. Authors are encouraged to use either the generic name, or the correct specific name(s) of the derivatives, as appropriate for each circumstance.

**Vitamin B<sub>12</sub>.** The term **vitamin B<sub>12</sub>** should be used as the generic descriptor for all corrinoids exhibiting qualitatively the biological activity of cyanocobalamin. The term **corrinoids** should be used as the generic descriptor for all compounds containing the corrin nucleus and thus chemically related to cyanocobalamin. The term **corrinoid** is not synonymous with the term **vitamin B<sub>12</sub>**.

**Vitamin C.** The terms **ascorbic acid** and **dehydroascorbic acid** will normally be taken as referring to the naturally occurring L-forms. If the subject matter includes other optical isomers, authors are encouraged to

include the L- or D- prefixes, as appropriate. The same is true for all those vitamins which can exist in both natural and alternative isomeric forms.

**Amounts of vitamins and summation.** Weight units are acceptable for the amounts of vitamins in foods and diets. For concentrations in biological tissues, SI units should be used; however, the authors may, if they wish, also include other units, such as weights or international units, in parentheses.

See *Metric Units, Conversion Factors and Nomenclature in Nutritional and Food Sciences*. London: The Royal Society, 1972 (paras. 8 and 14–20).

**Nomenclature of Fatty Acids and Lipids.** In the description of results obtained for the analysis of fatty acids by conventional gas-liquid chromatography, the shorthand designation proposed by Farquhar, J. W., Insull, W., Rosen, P., Stoffel, W. & Ahrens, E. H. (*Nutrition Reviews* (1959), 17, Suppl.) for individual fatty acids should be used in the text, tables and figures. Thus 18:1 should be used to represent a fatty acid with eighteen carbon atoms and one double bond; if the position and configuration of the double bond is unknown, this fatty acid should not be referred to as oleic acid. The shorthand designation should also be used in the synopsis but sentences should be constructed so that it is clear to the non-specialist reader that 18:1 refers to a fatty acid; for example, '... resulted in an increase in the concentrations of the fatty acid 18:1 in the liver triacylglycerols...'. If the positions and configurations of the double bonds are known, and these are important to the discussion, then a fatty acid such as linoleic acid may be referred to as *cis*-9,*cis*-12-18:2 (positions of double bonds related to the carboxyl carbon atom 1). However, to illustrate metabolic relations between different unsaturated fatty acid families, it is sometimes more helpful to number the double bonds in relation to the terminal methyl carbon atom, *n*. The preferred nomenclature is then: 18:3<sub>n-3</sub> and 18:3<sub>n-6</sub> for  $\alpha$ -linolenic and  $\gamma$ -linolenic acids respectively; 18:2<sub>n-6</sub> and 20:4<sub>n-6</sub> for linoleic and arachidonic acids respectively and 18:1<sub>n-9</sub> for oleic acid. Positional isomers such as  $\alpha$ - and  $\gamma$ -linolenic acid should always be clearly distinguished. It is assumed that the double bonds are methylene-interrupted and are of the *cis*-configuration (see Holman, R. T. in *Progress in the Chemistry of Fats and Other Lipids*, vol. 8, part I, p. 3. Oxford: Pergamon Press, 1966). Groups of fatty acids that have a common chain length but vary in their double bond content or double bond position should be referred to, for example, as C<sub>20</sub> fatty acids or C<sub>20</sub> polyunsaturated fatty acids. The modern nomenclature for glycerol esters should be used, i.e. triacylglycerol, diacylglycerol, monoacylglycerol *not* triglyceride, diglyceride, monoglyceride. The form of fatty acids used in diets should be clearly stated, i.e. whether ethyl esters, natural or refined fats or oils. The composition of the fatty acids in the dietary fat and tissue fats should be stated clearly, expressed as mol/100 mol or g/100 g total fatty acids.

**Nomenclature of Enzymes.** The nomenclature should be that of the Recommendations of the Nomenclature Committee of the International Union of Biochemistry

(*Enzyme Nomenclature*. London: Academic Press, 1992). Relevant EC numbers should be given.

**Nomenclature of Micro-organisms.** The correct name of the organism, conforming with international rules of nomenclature, must be used: if desired, synonyms may be added in parentheses when the name is first mentioned. Names of bacteria must conform with the current Bacteriological Code and the opinions issued by the International Committee on Systematic Bacteriology. Names of algae and fungi must conform with the current International Code of Botanical Nomenclature. Names of protozoa must conform with the current International Code of Zoological Nomenclature.

The following books may be found useful:

Holt, J. G., Krieg, N. R., Sneath, P. H. A., Staley, J. T. & Williams, S. T. (1994). *Bergey's Manual of Determinative Bacteriology*, 9th ed. Baltimore: The Williams and Wilkins Co.

N. J. W. Kreger-van Rij (editor) (1984). *The yeasts, a taxonomic study*. Amsterdam: Elsevier Science Publishers.

*Ainsworth and Bisby's Dictionary of the fungi*, (1983), 7th ed. Kew: Commonwealth Mycological Institute.

**Nomenclature of Plants.** For plant species where a common name is used that may not be universally intelligible, the Latin name in italics should follow the first mention of the common name. The cultivar should be given where appropriate.

**Other Nomenclature, Symbols and Abbreviations.** Authors should follow current numbers of the *British Journal of Nutrition* in this connection. The IUPAC rules on chemical nomenclature should be followed, and the Recommendations of the IUPAC-IUB Commission on Biochemical Nomenclature (see *Biochemical Journal* (1978) 169, 11–14). The symbols and abbreviations, other than units, are essentially those listed in *British Standard 5775* (1979–1982). *Specifications for Quantities, Units and Symbols*, parts 0–13. Day should be abbreviated to d, for example 7 d; except for example, 'each day', '7th day' and 'day 1'.

Elements and simple chemicals (e.g. Fe and CO<sub>2</sub>) can be referred to by their chemical symbol or formula from the first mention in the text; titles can be taken as an exception. Well-known abbreviations for chemical substances may be used without explanation, thus: RNA for ribonucleic acid and DNA for deoxyribonucleic acid. Other substances that are mentioned frequently may also be abbreviated, the abbreviation being placed in parentheses at the first mention, thus: free fatty acids (FFA) after that, FFA. Terms such as 'bioavailability' or 'available' may be used providing that the use of the term is adequately defined.

Spectrophotometric terms and symbols are those proposed in *IUPAC Manual of Symbols and Terminology for Physicochemical Quantities and Units* (1979) (London: Butterworths). The attention of authors is particularly drawn to the following symbols: m (= milli) = 10<sup>-3</sup>, μ

(micro)10<sup>-6</sup>, n (nano)10<sup>-9</sup> and p (pico)10<sup>-12</sup>. Note also that ml (millilitre) should be used instead of cc, μm (micrometre) instead of μ (micron) and μg (microgram) instead of γ.

Numbers. Figures should be used with units, for example 10 g, 7 d, 4 years (except when beginning a sentence, thus 'Four years ago...'); otherwise, words (except when 100 or more), thus: one man, ten ewes, ninety-nine flasks, three times (but with decimal 2.5, times), 100 patients, 120 cows, 136 samples.

**Ethics of Human Experimentation.** The notice of contributors is drawn to the guide-lines in the Declaration of Helsinki (1964) (*British Medical Journal* (1964) ii, 177–178), the Report of ELSE as printed in *British Journal of Nutrition* (1973) 29, 149, the *Guidelines on the Practice of Ethics Committees Involved in Medical Research Involving Human Subjects*, (1990) (London: The Royal College of Physicians) and to the *Guidelines for the Ethical Conduct of Medical Research Involving Children*, published in 1992 by the British Paediatric Association, 5 St Andrew's Place, Regents Park, London NW1 4LB. A paper describing any experimental work on human subjects should include a statement that the Ethical Committee in the Institution in which the work was performed has approved it and the authors should submit a copy of the letter of approval with the paper. A paragraph headed *Ethical considerations* in which the experiments are discussed and justified from an ethical standpoint should form the last paragraph of the Experimental section.

**Animal Experimentations.** The Editors will reject papers reporting work carried out using inhumane procedures. The criteria that they will adopt are set out in *Guidelines on the Use of Living Animals in Scientific Investigations*, published in 1987 by the Biological Council, Institute of Biology, 20 Queensbury Place, London SW7 2DZ, which are based on the UK legislation embodied in the Animals (Scientific Procedures) Act 1986. Where the Editors have doubts about the ethics of the procedures used they will normally seek guidance from the UK Home Office.

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