THE

BRITISH JOURNAL OF NUTRITION

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Form of Papers Submitted for Publication. The onus of preparing a paper in a form suitable for sending to press lies in the first place with the author. Authors should consult a current issue in order to make themselves familiar with the practice of the *British Journal of Nutrition* concerning typographical and other conventions, use of crossheadings, lay-out of tables, etc. Attention to these and other details (mentioned below) in the preparation of the

typescript before this is sent to the Editors will shorten the time required for publication. The need for undue amounts of editorial revision caused by badly prepared typescript will lead to delay in publication for which the Editors cannot accept responsibility. Papers on specialized aspects of the subject should be presented in such a way as to make them intelligible, without undue difficulty, to the ordinary reader of the Journal. In any case sufficient information should be made available to permit repetition of the published work by any competent reader of the Journal.

Papers intended for publication should be in doublespaced typing on one side of sheets of uniform size with adequate margins. Top copies only should be submitted, packed flat. The paper should be written in the English language, the spelling being that of the Oxford English Dictionary, and should, in general, be divided into the following parts: (a) Introductory paragraph, containing the reasons for publication of the work; (b) Experimental methods adopted: with chemical papers the experimental part will normally appear towards the end, but with other types of publication Methods should appear after Introduction; (c) Results: these should be given as concisely as possible, with the help of figures or tables; (d) Discussion: it is desirable that the presentation of the results and the discussion of their significance should be considered separately; (e) Summary: this should be a brief narrative (not more than 5% of the length of the paper) in the past tense of what was done and of results and conclusions; the paragraphs of the summary should be numbered; (f) References: these should be given in the text thus: Barnett & Robinson (1942), (Culbertson & Thomas, 1933); where a paper to be cited has more than two authors, the names of all the authors should be given when reference is first made, e.g. (Osborne, Mendel & Ferry, 1919); subsequent citations should appear thus: (Osborne et al. 1919). Where more than one paper by the same authors has appeared in one year the reference should be given as follows: Osborne & Mendel (1914a); Osborne & Mendel (1914b); or Osborne & Mendel (1914a, b); (Osborne & Mendel, 1914a, 1916; Barnett & Robinson, 1942).

References. At the end of the paper references should be given in alphabetical order according to the name of the first author of the publication quoted, and should include the authors' initials; the title of the paper should not be included. Titles of journals should be abbreviated in accordance with the system used in the World List of Scientific Periodicals (1934: 2nd ed. Oxford University Press). Examples of such abbreviations will be found in the current numbers of the British Journal of Nutrition and a useful list has recently been published in the Journal of Physiology (1945, 104, 232). References to books and monographs should include the town of publication and the name of the publisher, as well as the date of publication and the number of the edition to which reference is made. Thus:

Barnett, J. W. & Robinson, F. A. (1942). Biochem. J. 36,

Culbertson, C. C. & Thomas, B. H. (1934). Rep. Ia agric. Exp. Sta. 1933-4, p. 51.

Doisy, E. A., Somogyi, M. & Shaffer, P. A. (1923). J. biol. Chem. 55, Proc. xxxi.

Fairley, N. H. (1938). Nature, Lond., 142, 1156.

Hennessy, D. J. (1941). *Industr. Engng Chem.* (Anal. ed.), 13, 216.

King, H. (1941). J. chem. Soc. p. 338.

Osborne, T. B. & Mendel, L. B. (1914a). J. biol. Chem. 17, 325.

Osborne, T. B. & Mendel, L. B. (1914b). J. biol. Chem. 18, 1.
Osborne, T. B. & Mendel, L. B. (1916). Biochem. J. 10, 534.
Osborne, T. B., Mendel, L. B. & Ferry, E. L. (1919). J. biol. Chem. 37, 233.

Starling, E. H. (1915). Principles of Human Physiology, 2nd ed. London: Churchill.

Statistical Treatment of Data. In general the publication is not necessary of all the individual results of a number of replicated tests. A statement of the number of individual results, their mean value, and some appropriate measure of their variability, is usually sufficient.

The methods of analysis followed should be indicated, but statistical details, such as an analysis of variance tables, need not be given unless they are relevant to the discussion. A statement that the difference between the mean values of two groups of data is statistically significant should be accompanied by an indication of the level of significance attained.

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Description of Solutions. Solutions of common acids, bases and salts are preferably defined in terms of normality (N) or molarity (M), e.g. N-HCl; 0·1 M-NaH₂PO₄. The term '%' must be used in its correct sense, i.e. g./100 g. of solution. 10% HCl means 10 g. of hydrogen chloride in 100 g. of aqueous solution, and should never be used to indicate a ten-fold dilution of laboratory concentrated hydrochloric acid. For 'per cent by volume', i.e. ml./100 ml., the term '% (v/v)' may be employed. To indicate that a given weight of substance is contained in 100 ml. of solution, the term '% (w/v)' (weight per volume) may be used.

Symbols and Abbreviations. Authors should refer to current numbers of the British Journal of Nutrition for information in this connexion. The chemical nomenclature adopted is that followed by the Chemical Society (see J. chem. Soc. 1936, p. 1067). With a few exceptions the symbols and abbreviations are those adopted by a committee of the Chemical, Faraday and Physical Societies in 1937 (see J. chem. Soc. 1944, p. 717). Spectrophotometric terms and symbols are those proposed by the Society of Public Analysts and other Analytical Chemists (see The Analyst, 1943, 67, 164). For mathematical notation and numerals the rules laid down in Proc. roy. Soc. A, 1909, 82, 14, should be followed. The attention of authors is particularly drawn to the following symbols: $m = (milli) = 10^{-3}$ and $\mu = (\text{micro}) = 10^{-6}$. Note also that ml. (millilitres) should be employed instead of c.c., and µg. (micrograms) instead of y.

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