Microstructure of a Spinodal Fe-Ni-Mn-Al Alloy

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The crystal structures, chemical compositions, and morphology of the phases in a new quaternary alloy, Fe30Ni20Mn25Al25, were investigated using transmission electron microscopy (TEM), electron and x-ray diffraction, and x-ray energy dispersive spectroscopy (EDS). This alloy is of interest for its very high strength and hardness, and as an example of a class of quaternary transition metal alloys in which both spinodal decomposition behavior and ordering reactions are found. The alloy is also somewhat magnetic.

TEM of an arc-cast Fe30Ni20Mn25Al25 ingot revealed a periodic microstructure consisting of alternating ~45 nm and ~55 nm wide rods with coherent {100} interfaces, see Figure 1. Convergent beam electron diffraction showed that the crystal structures of these phases were b.c.c. and B2, respectively. The lattice parameters of the two phases were indistinguishable by x-ray diffraction, with both in the range 2.913 ± 0.007 Å. However, EDS analysis showed that the compositions of the two phases differed markedly, with much of the Al and nearly all the Ni segregated into the B2 phase, while most of the Fe and Mn were in the b.c.c. phase, see Table 1.

TEM of Fe30Ni20Mn25Al25 samples annealed at 550ºC for 16 h revealed the presence of incoherent third-phase precipitates; these had grown to micron-scale after annealing for 115 h at the same temperature, see Figure 2. The composition of these precipitates is rich in Fe and Mn, see Table 1, and diffraction patterns from them indicate a β-Mn structure.

The initial, metastable microstructure is strongly suggestive of formation by a spinodal decomposition mechanism. Observation of electron diffraction patterns in a hot straining stage, and analysis of samples quenched from high temperatures, indicated that the parent phase was B2-ordered.

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Figure 1: Bright-field images of as-cast Fe$_{30}$Ni$_{20}$Mn$_{25}$Al$_{25}$ viewed along [100].

Table 1: Chemical composition (at.%) of phases in Fe$_{30}$Ni$_{20}$Mn$_{25}$Al$_{25}$, as determined by EDS.

<table>
<thead>
<tr>
<th>Phase</th>
<th>Fe</th>
<th>Ni</th>
<th>Mn</th>
<th>Al</th>
</tr>
</thead>
<tbody>
<tr>
<td>BCC (as-cast)</td>
<td>49.1 ± 1.0</td>
<td>1.6 ± 0.15</td>
<td>30.0 ± 1.0</td>
<td>19.3 ± 1.4</td>
</tr>
<tr>
<td>B2 (as-cast)</td>
<td>12.7 ± 0.5</td>
<td>34.3 ± 0.8</td>
<td>13.9 ± 0.5</td>
<td>38.9 ± 0.9</td>
</tr>
<tr>
<td>Third phase (115 h at 550°C)</td>
<td>45.4 ± 0.9</td>
<td>&lt; 1</td>
<td>42 ± 1.2</td>
<td>12.3 ± 0.2</td>
</tr>
</tbody>
</table>

Figure 2: Dark field image of Fe$_{30}$Ni$_{20}$Mn$_{25}$Al$_{25}$ after 115 h at 550°C, taken along <100> using a {010} superlattice reflection, showing remnants of the spinodal microstructure between coarsened third-phase “β-Mn” particles.