

SECTION II

PHYSICAL PROCESSES IN PLANETARY NEBULAE

RECENT ADVANCES IN ATOMIC CALCULATIONS AND EXPERIMENTS OF INTEREST IN THE STUDY OF PLANETARY NEBULAE

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ABSTRACT

Recent advances in the calculation and measurement of transition probabilities, electron excitation rate coefficients and photoionization cross sections relevant to the study of planetary nebulae are discussed. A compilation of these parameters is also presented.

1 INTRODUCTION

Planetary nebulae are, in a sense, rich and challenging atomic physics laboratories. The low electron temperatures and densities of the nebular envelope give rise to a variety of atomic processes which are difficult to reproduce and measure in more accessible laboratories. The atomic data required to interpret these phenomena are therefore mainly obtained by calculation, where a clear understanding is necessary followed by an enormous computational effort. In the last decade several sophisticated computer packages have been developed to calculate atomic data with these points in mind. Advances in computer technology, such as the introduction of the CRAY-1 vector processor, have enabled a high degree of accuracy to be reached for the simpler ions, the possibility to treat consistently ions of the second row of the periodic table, and to make firm introductory attempts in the study of the larger ions such as those of iron.

In the present paper we review briefly recent developments made in the calculation and measurement of transition probabilities, electron excitation rate coefficients and photoionization cross sections. A compilation of these parameters is given in the Appendix.

2 TRANSITION PROBABILITIES

The calculation of accurate transition probabilities for forbidden and semi-forbidden lines, such as those found in nebular spectra, must

include the relativistic interaction and electron correlation effects. For light ions, where relativistic effects are small, the relativistic interaction can be treated as a perturbation by considering the Hamiltonian

$$H = H_{NR} + H_{BP} \quad (1)$$

where H_{NR} is the non-relativistic Hamiltonian and H_{BP} is the Breit-Pauli relativistic correction¹. The relativistic wave function $\psi_i(R)$ can then be expanded in terms of the non-relativistic wave functions $\psi_j(NR)$ thus

$$\psi_i(R) = \psi_i(NR) + \sum_{j \neq i} \psi_j(NR) \frac{\langle \psi_j(NR) | H_{BP} | \psi_i(NR) \rangle}{E_i(NR) - E_j(NR)} \dots \quad (2)$$

Electron correlation effects are usually included by the method of configuration interaction (CI). The NR wave functions are linearly expanded in a configuration basis of the form

$$\psi(NR) = \sum_k c_k \phi_k \quad (3)$$

where the configuration functions ϕ_k are built from one-electron orbitals. The convergence of the CI expansion can be slow and the transition probabilities of forbidden and semi-forbidden transitions are usually sensitive to the number of configurations in the CI expansion. In most cases configurations containing one-electron pseudo-orbitals, i.e. artificial orbitals adjusted variationally to improve term energy separations or fine-structure energy splittings, must be considered. Furthermore, if a high degree of accuracy is desired small semi-empirical adjustments cannot be avoided. For instance, Zeippen *et al.*² introduce semi-empirical term energy corrections such that $\psi(R)$ are calculated with the "exact" NR energies in equation (2), and calculate the A-values with the experimental energy level separations.

In the past these interactions were mainly treated by semi-empirical methods with a fair amount of success. Work such as that by Garstang³, Czyzak and Krueger⁴ and McKim-Malville and Berger⁵ on forbidden lines, and Laughlin and Victor⁶ on intercombination lines have provided reliable results for many years. The present trend is to carry out large CI calculations where such interactions are included more rigorously.

Perhaps the most significant recent theoretical development made in connection with transition probabilities relevant to planetary nebulae is that by Eissner and Zeippen⁷ on the transition probabilities for forbidden lines within the np³ configuration of the nitrogen and phosphorus isolectronic sequences. They have shown that higher-order relativistic corrections to the magnetic dipole operator⁸ (up to order $\alpha^2 Z^2$), which are usually

negligible, must be taken into account in order to obtain accurate A-values for the low members of the sequence, and thereby ending a long-standing discrepancy between the observed and theoretical values for the density sensitive line intensity ratio

$$r(N_e) = I(^2D_5^0 - ^4S_3^0)/I(^2D_3^0 - ^4S_3^0) \quad (4)$$

in the high density limit. For $N_e \rightarrow \infty$

$$r(\infty) = \frac{3}{2} A(^2D_5^0 - ^4S_3^0)/A(^2D_3^0 - ^4S_3^0), \quad (5)$$

where $A(i-j)$ is the transition probability, and thus provides a useful comparison between observation and theory. In table 1 we compare $r(\infty)$ for N^0 , O^+ and S^+ calculated in this way (MZ) with previous theoretical results (G) and observations (OBS).

ION	OBS	MZ	G
N^0	≤ 0.51	0.54	0.65
O^+	0.35	0.35	0.42
S^+	0.45	0.44	0.39

TABLE 1. Observed and theoretical values for $r(\infty)$ (see text). OBS(ref. 9, 10), MZ(ref. 11, 12) and G(ref. 3).

Also in connection with forbidden lines, Nussbaumer and Rusca¹³, Zeippen¹¹ and Mendoza and Zeippen¹² have carried out extensive calculations of A-values for transitions within the np^q configurations ($n=2-3$; $q=2-4$; $Z=6-28$), examining the importance of CI, relativistic corrections and semi-empirical adjustments. They have improved substantially on the earlier semi-empirical results compiled by Garstang³ and McKim-Malville and Berger⁵. Nussbaumer and Storey¹⁴ are systematically studying the ions of Fe.

There have been a considerable number of detailed calculations on the intercombination lines of astrophysical interest. The work of Glass and Hibbert¹⁵ and Nussbaumer and Storey¹⁶ on the Be-like ions may be chosen to illustrate the main features of these calculations. They demonstrate the high sensitivity of the transition probabilities to the values of small coefficients c_k of the CI expansion, and stress the need to use large configuration bases containing pseudo-orbitals and to make semi-empirical term energy corrections. Although different numerical methods

are used the agreement is better than 10%, in marked contrast with recent work by Cowan *et al.*¹⁷ on f-values for astrophysical intercombination lines where limited configuration bases are used and an accuracy of only 50% is claimed.

Even though a perturbative treatment of the relativistic interaction for light ions can be regarded as satisfactory, there are also numerical methods based on the more formal solution of the Dirac equation. Hata and Grant¹⁸ have used the multi-configuration Dirac-Fock method, which allows for correlation effects, to compute the lifetime of the astrophysically important 2^3S state of He⁰. They obtain a value of 8.8×10^3 sec in close agreement with the experimental result¹⁹ of $9 \pm 3 \times 10^3$ sec. This method has also been used recently²⁰ to compute a large number of E1, E2 and M1 transition probabilities within the n=2 complex with moderate accuracy for low Z.

The experimental techniques employed to measure lifetimes for allowed transitions are in general not suited for forbidden transitions and, consequently, there have been few experimental results. Corney and Williams²¹ have measured the radiative decay constant of the 1^1S_0 metastable state of O⁰ by the pulsed afterglow method. They obtain $\Gamma_{\text{expt}} = 1.31 \pm 0.05$ sec⁻¹ in excellent agreement with preliminary results by Mendoza and Zeippen of $\Gamma_{\text{th}} = 1.29$ sec⁻¹ using the CI method. There is also a new interesting technique being developed by Nightingale²² to measure transition probabilities for forbidden lines of singly-ionized species. He is trying to measure the A-value for the $1^1S_0 - 1^3D_0$ ($\lambda 5754$) transition of N⁺ by the long-path absorption spectroscopy method. Due to the small ionic absorption coefficients optical paths greater than 1 Km are required which he obtains by using a plasma source ten metres long and a multiple-pass optical laser system.

3 ELECTRON EXCITATION RATE COEFFICIENTS

Some of the problems encountered in the calculation of electron impact excitation cross sections of positive ions of astrophysical interest have been discussed by Mendoza²³. The electron excitation rate coefficient for a transition between levels $j \rightarrow i$ ($i > j$) at an electron temperature T_e can be expressed in terms of the effective collision strength $T_{ij}(T_e)$ (see Appendix), where

$$T_{ij}(T_e) = \int_0^\infty \Omega(i,j) \exp(-\epsilon_i/kT_e) d(\epsilon_i/kT_e), \quad (6)$$

ϵ_i is the energy of the electron with respect to the i th level, k is the Boltzmann constant and $\Omega(i,j)$ is the collision strength. For the low temperatures found in nebulae the cross section (i.e. $\Omega(i,j)$) must be calculated at low energies ($\epsilon_i \lesssim 1$ Ryd) where it is dominated by threshold effects such as resonances and the opening of new reaction channels. To be able to reproduce these effects detailed descriptions of the target

ion and the collision process are needed. Moreover, for light ions relativistic effects are usually neglected and LS coupling assumed; cross sections between fine-structure levels are then calculated by algebraic transformations of the reactance matrix²⁴.

The total wave function of the N-electron target ion + incoming electron is expanded in terms of the target eigenfunctions by the close-coupling (CC) expansion

$$\Psi(\text{LS}) = \sum_i A_{ij} \chi_i \theta_i + \sum_j c_j \phi_j. \quad (7)$$

Each term in the first summation gives rise to an interaction channel; χ_i are the target eigenfunctions, θ_i the incident-electron functions and A is an antisymmetrising operator. ϕ_j take the form of bound-state configurations for the (N+1)-electron system and account for short-range correlations effects. The variational principle leads to a set of coupled integro-differential equations of the general form

$$\left\{ \frac{d^2}{dr^2} - \frac{\ell_i(\ell_i+1)}{r^2} + \frac{2Z}{r} + \epsilon_i \right\} F_i(r) + \sum_{i'} W_{ii'} F_{i'}(r) = 0 \quad (8)$$

where $F_i(r)$ is the i th channel electron radial functions. The inter-channel coupling potentials $W_{ii'}$ give rise to the threshold effects mentioned above, and sufficient channels must be included in the CC expansion to account for the effects present in the region of interest. In earlier methods such as the Coulomb-Born and distorted wave the radial functions $F_i(r)$ were calculated neglecting this coupling, and in the exact resonance approximation²⁶ the important quadrupole p-wave interactions were introduced perturbatively. At present the coupled equations are solved numerically by different methods such as the Linear Algebraic (IMPACT)²⁷, the R-matrix²⁸ and the Non-iterative Integral Equation (NIEM)²⁹ methods. The Linear Algebraic method, in particular, can also be used to calculate bound-state energies for the (N+1)-electron system which leads to an indirect estimate of the accuracy of the cross sections.

Advances in this field have been boosted by the introduction of super-computers. It is now possible to study ions of the first and second rows of the periodic table with accurate CI target wave functions and sufficient target states in the CC expansion to ensure convergence. There have also been attempts to treat the ions of Fe with some consistency¹⁴. The carbon and silicon sequences have been studied extensively; in figure 1 we plot $z^2 T_{ij}(10000)$ (which tends to a constant limit as $z \rightarrow \infty$) as a function of z , where z is the effective charge of the ion, for the $3p - ^1D$ and $^1D - ^1S$ forbidden transitions of the np^2 configurations. It can be seen that the older results obtained with the exact resonance and distorted wave approximations can show large differences from recent close-coupling results.

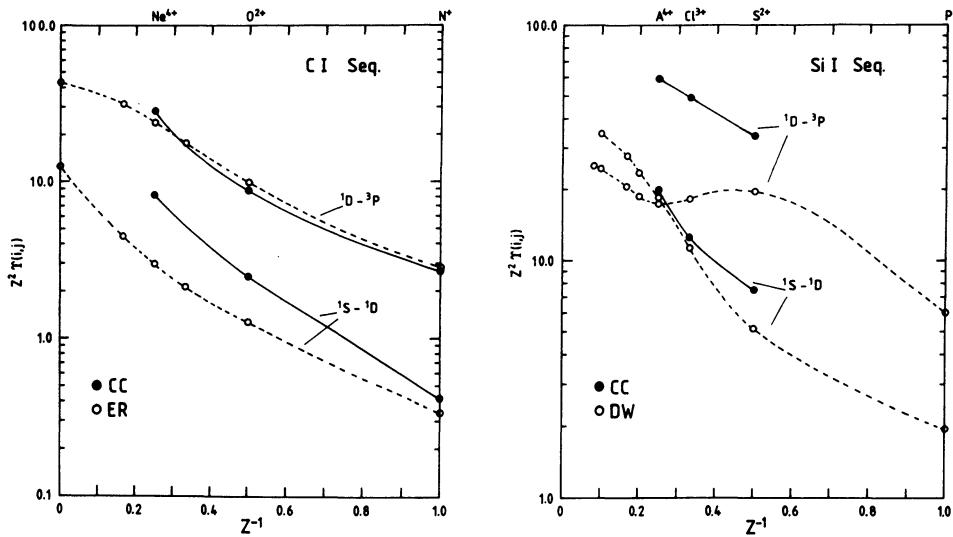


FIGURE 1. $z^2 T(10000)$ as a function of z (the effective charge of the ion) for the $^3\text{P}-^1\text{D}$ and $^1\text{D}-^1\text{S}$ forbidden transitions of the carbon and silicon sequences. CC - recent close-coupling results; ER - exact resonance approximation (ref.26); DW - distorted wave results (ref.35).

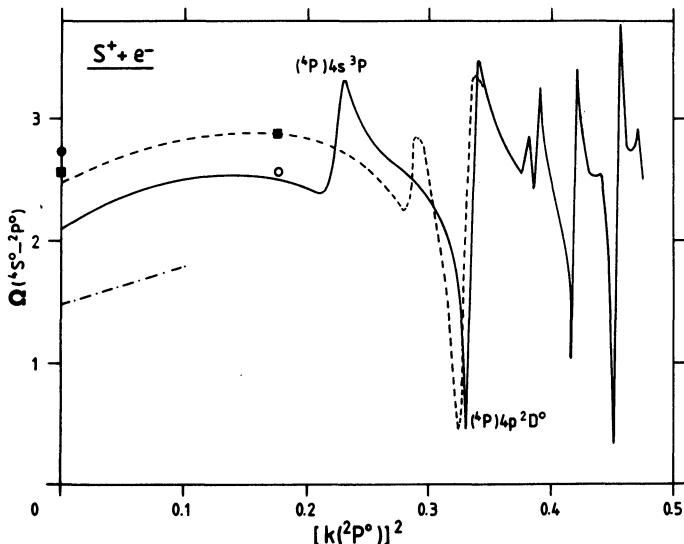


FIGURE 2. Total collision strength for the $^4\text{S}^0-^2\text{P}^0$ transition of S^+ . Work by Pradhan³⁶ in 3CC+CI (filled circles) and DW (filled squares) approximations; DW results by Krueger and Czyzak³⁵ (dot-dashed curve); recent results by Mendoza³⁴ in a 6CC+CI approximation using the R-matrix method (solid curve) and Linear Algebraic method (dashed curve).

For some cases, such as the forbidden transitions of O^{2+} (ref. 30, 31), Ne^{4+} (ref. 32, 33) and S^+ (ref. 34), there are comparisons using different close-coupling methods which gives a more realistic indication of the accuracy of the collision rates. In general the agreement is very good, but some transitions show discrepancies greater than the error margin assigned by the authors ($\lesssim 10\%$). This is mainly due to shifts in the resonance positions as illustrated in figure 2 for the $4S^0 - 2P^0$ transition of S^+ . Mendoza³⁴ has carried out comparative calculations for this transition using the Linear Algebraic and R-matrix methods. Although the same approximation was used in both calculations (6CC+CI) the target orbitals employed by these two packages are calculated by different methods: the former uses orbitals calculated in a Thomas-Fermi statistical model potential whereas the latter uses Hartree-Fock orbitals. The earlier results by Krueger and Czyzak³⁵ and Pradhan³⁶ are also shown.

In recent years there has also been progress in the relativistic treatment of the scattering problem. For intermediate-weight atoms the relativistic effects can be included by using the Breit-Pauli Hamiltonian; Scott and Burke³⁷ have combined this approach with the R-matrix method to calculate excitation cross sections for Fe XXIII finding large discrepancies with the non-relativistic case. For heavier ions it becomes necessary to base the scattering problem on the Dirac equation; Norrington and Grant³⁸ have included the Dirac Hamiltonian in the R-matrix formalism and tested their code with the electron scattering from Ne^+ . They find excellent agreement with previous non-relativistic calculations in support of the approximation mentioned above.

The most outstanding experimental work on electron impact excitation of multiply-ionized ions is that by Dunn and co-workers on the resonance transition of the Li, Na and K isoelectronic sequences using the crossed beams technique^{39,40,41}. They obtain absolute cross sections for C^{3+} and N^{4+} which agree with close-coupling calculations to better than 10%. This method has been refined recently to pick up the resonance structure found in the theoretical excitation cross section of Mg^+ (ref. 42), and preliminary results show satisfying agreement.

4 PHOTOIONIZATION CROSS SECTIONS

Experiments show that in most cases the photoionization continuum of light atoms is dominated by special features such as resonances, Cooper minima and delayed maxima. A similar situation is found by theory for the lowly-ionized species.

At present there are two tendencies in the calculation of photoionization cross sections for the large number of atoms and ions needed in astrophysics. On the one hand, there are detailed calculations which include electron correlation effects and aim at bringing out the complicated resonance structure and features of the cross section. Methods such as the close-coupling (described in the previous section) and Many-Body Perturbation Theory⁴³ are used. On the other hand, there are calcu-

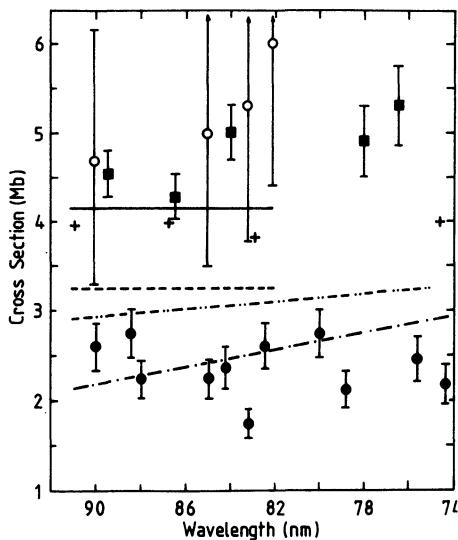


FIGURE 3. Present state of knowledge of the photoionization cross section of the ground state of O^0 in the threshold region. Experiments by Cairns and Samson⁴⁷ (open circles); Comes *et al.*⁴⁸ (filled circles); Kohl *et al.*⁴⁹ (filled squares). Theoretical results by Henry⁵⁰ in the length (dot-dash-dotted curve) and velocity (dot-dashed curve) formulations; close-coupling results by Taylor and Burke⁵¹ in the length (solid curve) and velocity (dashed curve) formulations; close-coupling results by Pradhan and Saraph⁵² in the length formulation (crosses). Figure originally by Kohl *et al.*

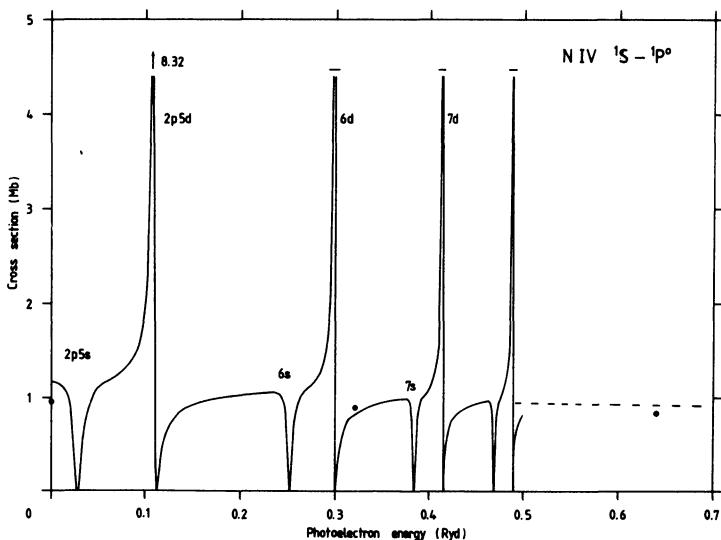


FIGURE 4. Photoionization cross section of the ground state of N^{3+} . Solid curve - close-coupling results by Butler *et al.* (to be published); dashed curve - average over resonances; filled circles - quantum defect results.

lations based on central potential approximations^{44, 45} or semi-empirical methods such as the one-channel Quantum Defect Theory⁴⁶ which only attempt to obtain the general behaviour of the cross section over a broad energy range, and these results are widely used in astrophysics.

In figure 3 we summarize the present state of knowledge of the photoionization cross section of the ground state of O^0 in the threshold region. It has taken almost a decade for experiment and theory to converge to a common result. The central potential calculation of Reilman and Manson⁴⁴ (not shown) grossly over-estimates the cross section in this region. In figure 4 the photoionization cross section of the ground state of N^{3+} is shown. It can be seen that correlation effects (resonances) are still conspicuous even for a fairly stripped ion, but the background of the cross section is obtained satisfactorily by the Quantum Defect method. Reilman and Manson⁴⁴ also obtain the background accurately in this region.

There is good agreement between theory and experiment for the rare gases, C^0 , N^0 , O^0 and Al^0 , but there are long-standing discrepancies for the alkalis. For ions there are no absolute experimental results, but it is possible for some numerical methods, such as the Linear Algebraic, to compute f-values at the same time. These can be compared with experiment to obtain an indirect estimate of the absolute value of the cross section.

There is also recent work on the relativistic effects of the photoionization cross section. Taylor and Scott⁵³ have calculated the resonance structure between the $J=3/2$ and $1/2$ levels of the A^+ ground state obtaining good agreement with experiment. Chang⁵⁴ has included the Dirac Hamiltonian in the R-matrix formalism and calculated the photoionization cross section of the ground state of Ne^0 . He finds close agreement with previous non-relativistic results.

5 FUTURE WORK

There remains a lot of work to be done in this interesting field. We can summarize the priority areas as follows.

- (i) Radiative and collision rates for transitions to the $n=3$ states of He^0 .
- (ii) Fine-structure collision rates for the intercombination lines of ions of the B isoelectronic sequence.
- (iii) Transition probabilities and fine-structure collision rates for ions of the Mg and Al sequences.
- (iv) Comparative CI calculations of transition probabilities for forbidden lines of the np^1 configurations, particularly of neutrals, using different types of orbitals. At the moment there is only a detailed comparison for O^{2+} (ref. 55).
- (v) Fine-structure collision rates for the IR forbidden lines of ions of the first and second row.
- (vi) Collision rates for ions of the second tow such as Si, Cl and A.
- (vii) Radiative and collision rates for the larger ions.

(viii) Detailed photoionization cross sections of atoms and ions, particularly from excited states and inner shells. Presentation of the data is also an important factor here.

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APPENDIX

COMPILED OF TRANSITION PROBABILITIES, ELECTRON EXCITATION RATE COEFFICIENTS AND PHOTOIONIZATION CROSS SECTIONS

We include in this section a selected and critically evaluated compilation of transition probabilities, electron excitation rate coefficients and photoionization cross sections for use in the study of planetary nebulae. We have attempted to present the data in a clear, practical and self-explanatory manner, but find it necessary to discuss briefly the general arrangements of the tables.

1 INDICES

Table 1 is an index of the ions and transitions for which A-values and electron excitation rate coefficients are included in this compilation. It also contains the numbers of the relevant tables (tables 2 - 9), the methods used, estimates of accuracy and the reference sources (referenced at the end of the Appendix). Similarly, table 10 contains a selected bibliography of photoionization cross sections; for each ion, the initial stages for which cross sections have been obtained, the states of the final ion, the method, accuracy and the reference source are given.

2 METHODS

The following abbreviations are used:

CI: Configuration interaction method.

nCC: n-state close-coupling approximation.

MV: Matrix variational method.

ER: Exact resonance approximation.

DW: Distorted wave approximation.

CBII: Unitarised Coulomb-Born approximation.

MP: Polarization model potential calculations.

CP: Central potential calculations.

MBPT: Many-body perturbation theory.

TDHF: Time-dependent Hartree-Fock method.

SE: Semi-empirical calculations.

TABLES: Values obtained from NBS tables.

INT(EXT): Interpolated (Extrapolated).

COMP: Compilation of several methods.

EXPT: Experiment.

3 ACCURACY

The following scheme is used:

A: Uncertainties within 10%

B: " " 20%

C: Uncertainties within 30%
D: " " 40%

A + sign means "much better than". Accuracy ratings given in this work should only be treated as a rough guide as it is very difficult to critically evaluate these data due to the general lack of experimental results, theoretical comparisons and even error estimates provided by the authors themselves. Furthermore, it is not practical within the present context to give estimates of uncertainties for every transition within a multiplet or configuration, and in some cases they can be significantly different. For instance, for the $ns^2 - nsnp$ transitions of the Be and Mg sequences, it is the transition probability for the intercombination line $^1S_0 - ^3P^o$ that shows the greater uncertainties whereas the A-values for the other transitions are probably correct to 5%; for the forbidden transitions within the np^q configurations the accuracy of the M1 transition probabilities is appreciably greater than that of the E2 type; also the accuracy of the photoionization cross section of the ground state of an ion can differ from that of the excited states, and the error in the cross sections can also vary with the energy region particularly where resonances are present.

4 TRANSITION PROBABILITIES AND EXCITATION RATE COEFFICIENTS

Tables 2 - 9 are arranged in isoelectronic sequences. For each transition they list: the observed energy level separation $\Delta E_{ij}(\text{cm}^{-1})$, the transition probability $A_{ij}(\text{sec}^{-1})$, and the effective collision strength $T_{ij}(T_e)$ as a function of electron temperature. The i th level always corresponds to the upper level unless ΔE appears with a negative sign (the A and T values are always for the downward transition). The electron de-excitation rate coefficient, q_{ij} , can be obtained from the relation

$$q_{ij}(T_e) = \frac{8.63 \times 10^{-6} T_{ij}(T_e)}{\omega_i T_e^{\frac{1}{2}}} \quad (\text{cm}^3 \text{sec}^{-1})$$

where ω_i is the statistical weight of the i th level and T_e is the electron temperature in $^{\circ}\text{K}$. The excitation rate coefficient is given by

$$q_{ji} = (\omega_i / \omega_j) q_{ij} \exp(-\Delta E_{ij}/kT_e) \quad (i > j)$$

where k is the Boltzmann constant ($1/k = 1.43883 \text{ cm K}$).

When only one value of T is given it is assumed to be temperature independent to within the accuracy of the calculation. A value of the effective collision strength bracketed by arrows ($\leftarrow T \rightarrow$) corresponds to the value for the whole multiplet (LS coupling); in this case the T 's for the fine-structure transitions can be obtained from the ratios of

the statistical weights of the multiplet levels. For the $^2P_0 - ^4P_J'$, transition of the B and Al isoelectronic sequences both the total T for the multiplet and that for each fine-structure transition are given, as in some cases they have been computed by different methods (eg N^{2+}) or, as explained by the author (eg O^{3+}), the sum of the fine-structure components does not add up to the total LS value given.

Finally, the notation $a \pm b$ signifies $a \times 10^{\pm b}$.

NOTE: The author kindly requests users to quote the original sources whenever data from this compilation are referenced.

ION	TRANSITION(S)	TABLE	A_{ij}			T_{ij}		
			METHOD	ACC	SOURCE	METHOD	ACC	SOURCE
H ⁰	n=2	2	TABLES	A+	I	CC	B+	126
He ⁰	n=2	2	COMP	A	1,2,3,4,5	5CC	C+	6
C ⁰	2p ² 1s, 1D, 3P	6	CI	A	7	CC+CI	C	8
C ⁰	2s2p ³ 5S ₂ ⁰ - 2s ² 2p ² 3P _J	6	CI	B	39	MV		9
C ⁺	2s2p ² 4P _J - 2s ² 2p 2P _J ⁰	5	CI	B+	10	8CC+CI	B	11
C ⁺	2s2p ² 4P - 2s ² 2p 2P ⁰	5				5CC+CI	B+	12
C ²⁺	2s2p 1P ₁ ⁰ , 3P _J ⁰ , 2s ² 1S ₀	4	CI	A	13	6CC+CI	A	14
C ³⁺	2p ² 2P _J ⁰ - 2s ² S ₁ ²	3	TABLES	A+	I	EXPT;5CC	A+	15;16
N ⁰	2p ³ 2P ⁰ , 2D ⁰ , 4S ⁰	7	CI	A	17	8CC+CI	A	18
N ⁰	2P _J ⁰ - 2D _J ⁰	7	CI	A	17	CC	C	19
N ⁺	2p ² 1s, 1D, 3P	6	CI	A+	7	CC+CI	A	20
N ⁺	2p ² 3P _J ⁰ - 3P _J ¹	6	CI	A+	7	ER	C	21
N ⁺	2s2p ³ 5S ₂ ⁰ - 2s ² 2p ² 3P _J	6	CI	B+	22	7CC+CI	B+	12
N ²⁺	2s2p ² 4P _J - 2s ² 2p 2P _J ⁰	5	CI	B+	23	DW	C	23
N ²⁺	2s2p ² 4P - 2s ² 2p 2P ⁰	5				6CC+CI	A	24
N ³⁺	2s2p 1P ₁ ⁰ , 3P _J ⁰ , 2s ² 1S ₀	4	CI	A	25	INT	B+	
N ⁴⁺	2p ² 2P _J ⁰ - 2s ² S ₁ ²	3	TABLES	A+	I	2CC	A+	26,27
O ⁰	2p ⁴ 1s, 1D, 3P	8	CI	A	28	6CC+CI	A	18
O ⁰	2p ⁴ 3P _J ⁰ - 3P _J ¹	8	CI	A	28	MV	B+	29
O ⁺	2P ³ 2P _J ⁰ , 2D _J ⁰ , 4S ₁ ⁰	7	CI	A+	17	5CC+CI	A	30
O ²⁺	2s2p ³ 5S ⁰ , 2s ² 2p ² 1S, 1D, 3P	6	CI	A+	31	12CC+CI	A	32
O ²⁺	2p ² 3P _J ⁰ - 3P _J ¹	6	CI	A+	31	12CC+CI	A	33
O ³⁺	2s2p ³ 4P _J - 2s ² 2p 2P _J ⁰	5	CI	B	34	7CC+CI	A	34
O ⁴⁺	2s2p 1P ₁ ⁰ , 3P _J ⁰ , 2s ² 1S ₀	4	CI	A	25	6CC+CI	A	14
O ⁵⁺	2p ² 2P _J ⁰ - 2s ² S ₁ ²	3	TABLES	A+	I	CBII	A	26

TABLE 1. Index of transition probabilities, A_{ij} , and effective collision strengths, T_{ij} , for transitions in ions of interest in the study of planetary nebulae. For each ion the number of the relevant table (tables 2 - 9), the method used, an estimate of the accuracy and the reference source (referenced at the end of the appendix) is given.

ION	TRANSITION(S)	TABLE	A _{I,J}			T _{I,J}		
			METHOD	ACC	SOURCE	METHOD	ACC	SOURCE
Ne ⁺	2p ⁵ 2P _{1/2} ⁰ - 2P _{3/2} ⁰	9	CI	A+	28	ER(adj.)	A	20
Ne ²⁺	2p ⁴ 1S, 1D, 3P	8	CI	A+	28	4CC+CI	A	35
Ne ²⁺	2p ⁴ 3P _J - 3P _{J'}	8	CI	A+	28	ER	C	21
Ne ³⁺	2p ³ 2P _J ⁰ , 2D _J ⁰ , 4S _{1/2} ⁰	7	CI	A+	17	9CC+CI	A	36
Ne ⁴⁺	2p ² 1S, 1D, 3P	6	CI	A+	7	12CC+CI	A	37
Ne ⁴⁺	2s2p ³ 5S ₂ ⁰ - 2s ² 2p ² 3P _J	6	CI	A	39	12CC+CI	A	37
Ne ⁴⁺	2p ² 3P _J - 3P _{J'}	6	CI	A+	7	ER	C	21
Ne ⁵⁺	2s ² 2p 2P _{1/2} ⁰ - 2P _{3/2} ⁰	5	TABLES	A+	1	ER	C	21
Ne ⁶⁺	2s2p 1P ₁ ⁰ , 3P _J ⁰ , 2s ² 1S ₀	4	CI	A	25	6CC+CI	A	38
Na ²⁺	2p ⁵ 2P _{1/2} ⁰ - 2P _{3/2} ⁰	9	CI	A+	28	ER	C	21
Na ³⁺	2p ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A+	28	ER	C	21
Na ⁴⁺	2p ³ 2P _J ⁰ , 2D _J ⁰ , 4S _{3/2} ⁰	7	CI	A+	17	ER	C	21
Mg ⁰	3s3p 3P ₁ ⁰ , 3P _J ⁰ , 3s ² 1S ₀	4	CI,MP	B	39,40	DW,CC		41,42
Mg ⁺	3p 2P _J ⁰ - 3s ² S _{1/2}	3	MP	A+	43	4CC	A	44
Mg ³⁺	2p ⁵ 2P _{1/2} ⁰ - 2P _{3/2} ⁰	9	CI	A+	28	ER	C	21
Mg ⁴⁺	2P ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A+	28	ER	C	21
Mg ⁵⁺	2p ³ 2P _J ⁰ , 2D _J ⁰ , 4S _{3/2} ⁰		CI	A+	17	ER	C	21
Si ⁰	3p ² 1S ₀ , 1D ₂ , 3P _J	6	CI	A	45	DW		46
Si ⁺	3s3p ² 4P - 3s ² 3p 2P ⁰	5	CI	B	47	7CC+CI	B	48
Si ²⁺	3s3p 1P ₁ ⁰ , 3P _J ⁰ , 3s ² 1S ₀	4	CI,MP	B	39,40	12CC+CI	B+	49
Si ³⁺	3p 2P _J ⁰ - 3s ² S _{1/2}	3	ICC+MP	A+	28	DW	B	26,50
Si ⁵⁺	2p ⁵ 2P _{1/2} ⁰ - 2P _{3/2} ⁰	9	CI	A+	28	ER	C	21
S ⁰	3p ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A	51			
S ⁺	3p ³ 2P _J ⁰ , 2D _J ⁰ , 4S _{3/2} ⁰	7	CI	A+	52	6CC+CI	A	53
S ²⁺	3p ² 1S, 1D, 3P	6	CI	A+	45	12CC+CI	B+	53
S ²⁺	3p ² 3P _J ⁰ - 3P _{J'}	6	CI	A+	45	7CC+CI	B+	53
S ³⁺	3s3p ² 4P _J - 3s ² 3p 2P _J ⁰	5	CI	B+	54	6CC+CI	B+	54
S ⁴⁺	3s3p 1P ₁ ⁰ , 3P _J ⁰ , 3s ² 1S	4	CI,MP	B	39,40	5CC+CI	B+	56
S ⁴⁺	3s3p 3P _J ⁰ - 3P _{J'}	4	CI	A+	39	DW(EXT)		55
S ⁵⁺	3p 2P _J ⁰ - 3s ² S _{1/2}	3	ICC+MP	A+	28	DW	B	26,50

TABLE 1. (continued)

ION	TRANSITION(S)	TABLE	A _{IJ}			T _{IJ}		
			METHOD	ACC	SOURCE	METHOD	ACC	SOURCE
Cl ⁰	3p ⁵ 2P _{1/2} ^o - 2P _{3/2} ^o	9	CI	A+	28			
Cl ⁺	3p ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A+	51	DW	D	57
Cl ²⁺	3p ³ 2P _J ^o , 2D _J ^o , 4S _{3/2} ^o	7	CI	A+	52	DW	D	57
Cl ³⁺	3p ² 1S, 1D, 3P	6	CI	A+	45	I2CC+CI	B	53
Cl ³⁺	3p ² 3P _J - 3P _J ^o	6	CI	A+	45	DW	D	57
Cl ⁴⁺	3s ² 3p 2P _{3/2} ^o - 2P _{1/2} ^o	5	CI	A+	28	DW	D	57
A ⁺	3p ⁵ 2P _{1/2} ^o - 2P _{3/2} ^o	9	CI	A+	28	DW	D	57
A ²⁺	3p ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A+	51	DW	D	57
A ³⁺	3p ³ 2P _J ^o , 2D _J ^o , 4S _{3/2} ^o	7	CI	A+	52	DW	D	57
A ⁴⁺	3p ² 1S, 1D, 3P	6	CI	A+	45	I2CC+CI	B	53
A ⁴⁺	3p ² 3P _J - 3P _J ^o	6	CI	A+	45	DW	D	57
A ⁵⁺	3s ² 3p 2P _{3/2} ^o - 2P _{1/2} ^o	5	CI	A+	28	DW	D	57
K ²⁺	3p ⁵ 2P _{1/2} ^o - 2P _{3/2} ^o	9	CI	A+	28	DW	D	57
K ³⁺	3p ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A+	51	DW	D	57
K ⁴⁺	3p ³ 2P _J ^o , 2D _J ^o , 4S _{3/2} ^o	7	CI	A+	52	DW	D	57
Ca ⁺	4p 2P _J ^o - 4s 2S _{1/2}	3	MP	A	43	EXPT	A	26,58
Ca ³⁺	3p ⁵ 2P _{1/2} ^o - 2P _{3/2} ^o	9	CI	A+	28	DW	D	57
Ca ⁴⁺	3p ⁴ 1S ₀ , 1D ₂ , 3P _J	8	CI	A+	51	DW	D	57
Fe ⁰	Forbidden transitions	CI			59			
Fe ⁺	Forbidden transitions	CI		D	60	4CC+CI	D	60
Fe ²⁺	Forbidden transitions	SE			61	SCC		62
Fe ³⁺	Forbidden transitions	SE			63,64			
Fe ⁴⁺	Forbidden transitions	SE			61			
Fe ⁵⁺	Forbidden transitions	CI		C	65	DW	C	65
Fe ⁶⁺	Forbidden transitions	CI		B	66	DW	C	66

TABLE 1. (continued)

PARM	$T_e (10^4 \text{ K})$	2s-1s	2p-1s		
ΔE	82258.9	82259.2		ΔE	159850.3
A	0.244	4.699+8	0.403	A	1.13-4
T	0.5	0.280	0.485	T	0.5
1.0	0.300	0.563	0.75		0.0686
1.5	0.311	0.631	1.0		0.0719
2.0			0.0736		0.0393
			2.0		0.0435
			3.0		0.0612
				(i)	0.0792
				(ii)	0.0817
					0.0526
					0.0543

PARM	$T_e (10^4 \text{ K})$	2 ¹ s-2 ³ s	2 ³ p-2 ³ s	2 ¹ p-2 ³ s	2 ³ p-2 ¹ s	2 ¹ p-2 ¹ s	2 ¹ p-2 ³ p
ΔE	6421.4	9230.9	11278.8	2809.6	4857.5	2047.9	
A	1.51-7	1.022+7	1.29	2.7-2	1.976+6		
T	0.1	1.61	3.08	0.312	0.967	1.07	2.47
0.2	1.97	5.47	0.464	1.15	2.87		2.83
0.5	2.39	14.0	0.750	1.41	9.35		3.42
0.75	2.49	21.9	0.906	1.55	14.5		3.75
1.0	2.52	29.5	1.03	1.66	19.1		4.05
2.0	2.45	57.0	1.28	1.89	33.2		4.80
3.0	2.32	80.0	1.36	1.94	44.1		5.08
						(iii)	

TABLE 2. Energy level separations $\Delta E (\text{cm}^{-1})$, transition probabilities A (sec^{-1}) and effective collision strengths $T(T_e)$ for: (i) 1 \rightarrow 2 transitions of H^0 ; (ii) 1 \rightarrow 2 transitions of He^0 ; (iii) transitions within n=2 of He^0 .

ION	PARM	$T_e (10^4 K)$	$2P_1^0 - S_1^0$	$2P_1^0 - S_1^0$	ION	PARM	$T_e (10^4 K)$	$2P_1^0 - S_1^0$	$2P_1^0 - S_1^0$	ION	PARM	$T_e (10^4 K)$	$2P_1^0 - S_1^0$	$2P_1^0 - S_1^0$	
C^{3+}	ΔE	66591.7	66484.0	Mg ⁺			35761.0	35669.4	Ca ⁺			25414.4	25191.5		
	A	2.65+8	2.63+8	A	ΔE	T	2.55+8	2.54+8	A	ΔE	T	1.48+8	1.45+8		
	T	0.5	1.0	8.88	→		15.6	16.5	1.0	15.6	0.5	1.0	17.5	15.6 →	
		1.5	2.0	8.95			1.0	1.5	1.5	1.0		1.5	19.2		
							1.5	17.2	17.7	1.5		2.0	20.8		
N^{4+}	ΔE	80721.9	80463.2	Si ³⁺			71748.6	71287.5							
	A	3.38+8	3.36+8	A	ΔE	T	9.26+8	9.15+8							
	T	0.5	1.0	6.61	→		16.9	17.0							
		1.5	2.0	6.65			1.0	1.5	1.5	1.0		1.5	17.0		
				6.69			1.5	2.0	2.0	1.5		2.0	17.1		
O^{5+}	ΔE	96907.5	96375.0	S ⁵⁺			107137	105874							
	A	4.09+8	4.02+8	A	ΔE	T	1.75+9	1.70+9							
	T	0.5	1.0	4.98	→		11.8	11.9							
		1.5	2.0	5.00			1.0	1.5	1.5	1.0		1.5	11.9		
				5.03			1.5	2.0	2.0	1.5		2.0	11.9		
				5.05											

TABLE 3. Energy level separations ΔE (cm⁻¹), transition probabilities A (sec⁻¹) and effective collision strengths $T(T_e)$ for the ns - np transition of the Li (n=2), Na (n=3) and K (n=4) isoelectronic sequences.

ION	PARM	T_e (10^4 K)	$^3P_2^o - ^1S_0$	$^3P_1^o - ^1S_0$	$^3P_0^o - ^1S_0$	$^1P_1^o - ^1S_0$	$^3P_1^o - ^3P_0^o$	$^3P_2^o - ^3P_0^o$	$^3P_2^o - ^3P_1^o$
C^{2+}	ΔE		52447.1	52390.8	52367.1	102352.0	23.7	80.1	56.4
	A		5.19-3	9.59+1		1.79+9	2.39-7		2.41-6
	T	0.5	←→	1.12	←→	3.85	0.848	0.579	2.36
		1.0		1.01		4.34	0.911	0.677	2.66
		1.5		0.990		4.56	0.975	0.776	2.97
N^{3+}	ΔE		67416.3	67272.3	67209.2	130693.9	63.1	207.1	144.0
	A		1.15-2	5.77+2		2.38+9	4.53-6		4.03-5
	T	0.5	←→	0.904	←→	3.20			
		1.0		0.852		3.46			
		1.5		0.817		3.58			
O^{4+}	ΔE		82385.3	82078.6	81942.5	158797.7	136.1	442.8	306.7
	A		2.16-2	2.25+3		2.92+9	4.54-5		3.89-4
	T	0.5	←→	0.733	←→	2.66			
		1.0		0.721		2.76			
		1.5		0.674		2.82			
Ne^{6+}	ΔE		112711.5	111717	111264.9	214951.6	452	1446.6	995
	A		5.78-2	1.98+4		4.08+9	1.69-3		1.32-2
	T	0.5	←→	0.129	←→	1.39			
		1.0		0.172		1.56			
		1.5		0.205		1.63			
Mg^o	ΔE		21911.2	21870.5	21850.4	35051.3	20.1	60.8	40.7
	A		4.13-4	1.80+2		4.93+8	1.45-7	4.08-12	9.10-7
	T								
Si^{2+}	ΔE		53115.0	52853.3	52724.7	82884.4	128.6	390.3	261.7
	A		1.20-2	1.26+4		2.60+9	3.82-5	3.20-9	2.42-4
	T	0.5	←→	6.90	←→	5.48			
		1.0		5.43		5.82			
		1.5		4.80		6.21			
S^{4+}	ΔE		84155.2	83393.5	83024.0	127150.7	369.5	1131.2	761.7
	A		6.59-2	1.26+5		5.13+9	9.07-4	1.63-7	5.96-3
	T	0.5	←→	0.911	←→	7.30	0.272	0.400	1.24
		1.0		0.910		7.30			
		1.5		0.914		7.29			
		2.0		0.905		7.27			

TABLE 4. Energy level separations ΔE (cm^{-1}), transition probabilities A (sec^{-1}) and effective collision strengths $T(T_e)$ for the $ns^2 - nsnp$ transitions of the Be ($n=2$) and Mg ($n=3$) isoelectronic sequences.

ION	PARM	T _e (10 ⁴ K)	⁴ P - ² P ^o	² P _{3/2} ^o - ² P _{1/2}	⁴ P _{1/2} - ² P _{1/2} ^o	⁴ P _{3/2} - ² P _{1/2} ^o	⁴ P _{3/2} - ⁴ P _{1/2}	⁴ P _{3/2} - ² P _{1/2}	⁴ P _{3/2} - ² P _{1/2} ^o	⁴ P _{3/2} - ⁴ P _{1/2}	
C ⁺	ΔE										
	A	0.4	3.25								
	T	1.0	3.17	1.25							
		1.5	3.09								
N ²⁺	ΔE										
	A	0.4	2.00	0.701	174.4 4.77-5	57012.7 3.39+2	57246.8 3.64+2	59.7 5.90+1	57327.9 0.175	57153.5 0.695	140.8 2.51+2
	T	1.0	2.03	0.0952	0.0616	0.139	0.174	0.0800	0.390	43990.2 4.32+1	50.3 3.49-14
		1.4	2.07							28.3 3.67-7	
O ³⁺	ΔE										
	A	1.0	1.27	2.33	386.3 5.20-4	71440.0 1.22+3	71053.7 1.24+3	71185.1 3.24+1	71755.9 5.07-5	71369.6 0.122	184.5 9.37+2
	T	1.5	1.39		0.113	0.087	0.174	0.234	0.989	0.506	1.02-4 1.82
		2.0	1.48		2.40	0.128	0.106	0.197	0.274	0.568	1.97
Ne ⁵⁺	ΔE										
	A				1306.6 2.02-2						
	T				0.433						
					287.3 2.17-4	42824.4 4.55+3	42537.0 3.00+3	42932.7 1.32+1	42645.4 1.62+3	108.3 1.62+3	43108.0 1.62+3
Si ⁺	ΔE										
	A	0.5	5.28								
	T	1.0	5.14								
		1.5	5.05								
S ³⁺	ΔE										
	A	2.0	4.97								
	T				950.2 7.73-3	71180 5.50+4	70230 3.39+4	71524 1.40+2	70574 1.95+4	344 1.47	72071 3.04
					6.42 6.41	0.51 0.47	0.66 0.62	0.87 0.82	2.82 1.38	7.01 0.55	
Cl ⁴⁺	ΔE										
	A										
	T				6.38 6.38	0.44 0.44	0.59 0.77	1.30 2.59	0.85 0.85	5.47 2.40	
					1490.8 2.98-2	1.052				6.49 2.40	
Ar	ΔE										
	A										
	T				2207.5 9.66-2	0.798				5.85	
S ⁺	ΔE										
	A										
	T										

TABLE 5. Energy level separations ΔE (cm⁻¹), transition probabilities A (sec⁻¹), and effective collision strengths T(E) for the ns² np - nnp² transitions of the B (n=2) and Al (n=3) isoelectronic sequences.

ION	PARM	T_e ($10^4 K$)	$^1D_2 - ^3P_0$	$^1D_2 - ^3P_1$	$^1D_2 - ^3P_2$	$^1S_0 - ^3P_1$	$^1S_0 - ^3P_2$	$^1S_0 - ^1D_2$	$^3P_1 - ^3P_0$	$^3P_2 - ^3P_0$	$^3P_2 - ^3P_1$	$^3S_2 - ^3P_1$	$^5S_2 - ^3P_2$
C ⁰	ΔE		10192.6	10176.2	10149.2	21631.6	21604.6	11455.4	16.4	43.4	27.0	33718.8	33691.8
	A		7.77-8	8.21-5	2.44-4	2.71-3	2.00-5	5.28-1	7.93-8	1.71-14	2.65-7	6.94	1.56+1
	T	0.05	$\xleftarrow{\quad}$	$\xrightarrow{0.0625}$	$\xleftarrow{\quad}$	$\xrightarrow{0.0172}$	$\xrightarrow{\quad}$	$\xrightarrow{0.0620}$	$\xrightarrow{\quad}$	$\xrightarrow{0.0877}$	$\xrightarrow{\quad}$	$\xrightarrow{0.150}$	$\xrightarrow{\quad}$
	0.1		0.125			0.0339		0.196				0.212	
	0.5		0.603		0.149		0.196					0.475	
	1.0		1.14		0.252		0.277					0.671	
	1.5		1.60		0.320		0.340					0.822	
	2.0		1.96		0.365		0.392					0.950	
N ⁺	ΔE		15316.2	15267.5	15185.4	32640.1	32558.0	17372.6	48.7	130.8	82.1	46735.9	46653.8
	A		5.35-7	1.01-3	2.99-3	3.38-2	1.51-4	1.12	2.08-6	1.16-12	7.46-6	4.8+1	1.07+2
	T	0.5	$\xleftarrow{\quad}$	$\xrightarrow{2.64}$	$\xrightarrow{\quad}$	$\xrightarrow{\quad}$	$\xrightarrow{0.352}$	$\xrightarrow{\quad}$	$\xrightarrow{0.405}$	$\xrightarrow{0.401}$	$\xrightarrow{0.279}$	$\xrightarrow{1.13}$	$\xrightarrow{\quad}$
	1.0		2.68		0.352		0.411					1.28	
	1.5		2.72		0.359		0.418					1.29	
	2.0		2.73		0.365		0.423					1.27	
O ²⁺	ΔE		20273.3	20160.1	19967.1	43072.5	42879.5	22912.4	113.2	306.2	193.0	60211.8	60018.8
	A		2.74-6	6.74-3	1.96-2	2.23-1	7.85-4	1.78	2.62-5	3.02-11	9.76-5	2.12+2	5.22+2
	T	0.5	$\xleftarrow{\quad}$	$\xrightarrow{2.02}$	$\xrightarrow{\quad}$	$\xrightarrow{\quad}$	$\xrightarrow{0.248}$	$\xrightarrow{\quad}$	$\xrightarrow{0.516}$	$\xrightarrow{0.517}$	$\xrightarrow{0.257}$	$\xrightarrow{1.22}$	$\xrightarrow{\quad}$
	1.0		2.17		0.276		0.617		0.542	0.271	1.29		1.18
	1.5		2.30		0.299		0.638		0.553	0.281	1.32		1.22
	2.0		2.39		0.314		0.634		0.556	0.288	1.34		1.24
Ne ⁴⁺	ΔE		30291.5	29879.1	29181.4	63501.2	62803.5	33622.1	412.4	1110.1	697.7	87950.7	87253.0
	A		2.37-5	1.31-1	3.65-1	4.21	6.69-3	2.85	1.28-3	5.08-9	4.59-3	2.37+3	6.06+3
	T	0.5	$\xleftarrow{\quad}$	$\xrightarrow{1.70}$	$\xrightarrow{\quad}$	$\xrightarrow{\quad}$	$\xrightarrow{0.284}$	$\xrightarrow{\quad}$	$\xrightarrow{0.581}$	$\xrightarrow{0.244}$	$\xrightarrow{0.122}$	$\xrightarrow{0.578}$	$\xrightarrow{\quad}$
	1.0		1.78		0.248		0.518					1.51	
	1.5		1.85		0.240		0.550					1.53	
	2.0		1.92		0.238		0.602					1.51	
Si ⁰	ΔE		6298.9	6221.7	6075.7	15317.3	15171.2	9095.5	77.1	223.2	146.0	33248.9	33102.9
	A		4.70-7	7.93-4	2.25-3	3.13-2	9.02-4	1.14	8.25-6		4.21-5		
	T												
S ²⁺	ΔE		11320	11023	10488	26866	26331	15843	297.2	832.5	535.3	59401	58866
	A		5.82-6	2.21-2	5.76-2	7.96-1	1.05-2	2.22	6.72-4	4.61-8	2.07-3		
	T	0.5	$\xleftarrow{\quad}$	$\xrightarrow{9.07}$	$\xrightarrow{\quad}$	$\xrightarrow{\quad}$	$\xrightarrow{1.16}$	$\xrightarrow{\quad}$	$\xrightarrow{1.42}$	$\xrightarrow{2.64}$	$\xrightarrow{1.11}$	$\xrightarrow{5.79}$	
	1.0		8.39			1.19		1.88		2.59	1.15	5.81	
	1.5		8.29			1.21		2.02		2.38	1.15	5.56	
	2.0		8.20			1.24		2.08		2.20	1.14	5.32	
Cl ³⁺	ΔE		13767.6	13275.6	12425.7	32055.8	31205.9	18780.2	492.0	1341.9	849.9		
	A		1.54-5	7.23-2	1.79-1	2.47	2.62-2	2.80	2.14-3	2.70-7	8.25-3		
	T	0.5	$\xleftarrow{\quad}$	$\xrightarrow{5.10}$	$\xrightarrow{\quad}$	$\xrightarrow{\quad}$	$\xrightarrow{2.04}$	$\xrightarrow{\quad}$	$\xrightarrow{0.935}$	$\xrightarrow{0.475}$	$\xrightarrow{0.400}$	$\xrightarrow{1.50}$	
	1.0		5.42			2.27		1.39					
	1.5		5.88			2.32		1.73					
	2.0		6.19			2.30		1.92					
Ar ⁴⁺	ΔE		16299.4	15535.5	14270.2	37148.6	35883.3	21613.1	763.9	2029.2	1265.3		
	A		3.50-5	2.04-1	4.76-1	6.55	5.69-2	3.29	7.99-3	1.24-6	2.72-2		
	T	0.5	$\xleftarrow{\quad}$	$\xrightarrow{4.37}$	$\xrightarrow{\quad}$	$\xrightarrow{\quad}$	$\xrightarrow{1.17}$	$\xrightarrow{\quad}$	$\xrightarrow{1.26}$	$\xrightarrow{0.257}$	$\xrightarrow{0.320}$	$\xrightarrow{1.04}$	
	1.0		3.72			1.18		1.25					
	1.5		3.52			1.11		1.24					
	2.0		3.42			1.03		1.23					

TABLE 6. Energy level separations ΔE (cm^{-1}), transition probabilities A (sec^{-1}) and effective collision strengths T(T_e) for the $ns^2 np^2$ and $nspn^3$ transitions of the C (n=2) and Si (n=3) isoelectronic sequences.

ION	PARM	T_e ($10^4 K$)	$2_{D_2^-}^0 4_{S_2^0}$	$2_{D_2^-}^0 4_{S_2^0}$	$2_{P_2^-}^0 4_{S_2^0}$	$2_{P_2^-}^0 4_{S_2^0}$	$2_{D_2^-}^0 2_{D_2^0}$	$2_{P_2^-}^0 2_{P_2^0}$	$2_{P_2^-}^0 2_{P_2^0}$	$2_{F_2^-}^0 2_{D_2^0}$	$2_{P_2^-}^0 2_{D_2^0}$	$2_{P_2^-}^0 2_{D_2^0}$
N ⁰	ΔE	19224.5	19233.2	28839.3	28838.9	-8.71	0.386	9614.8	9606.1	9614.5	9605.7	9605.7
	A	7.27-6	2.02-5	6.58-3	2.71-3	1.27-8		6.14-2	2.76-2	3.45-2		5.29-2
	T	0.05	9.18-3	6.12-3	3.23-3	1.6-3						
		0.1	2.30-2	1.53-2	8.53-3	4.27-3						
		0.5	0.155	0.103	0.0597	0.0298	0.128	0.0329	0.162	0.0856	0.0626	0.0601
		1.0	0.290	0.194	0.113	0.0567	0.269	0.071	0.266	0.147	0.109	0.097
		2.0	0.476	0.318	0.189	0.0947	0.465	0.153	0.438	0.252	0.190	0.157
O ⁺	ΔE	26810.7	26830.2	40467.5	40468.6	-19.5	-1.1	13656.8	13637.3	13657.9	13638.4	13638.4
	A	3.82-5	1.65-4	5.64-2	2.32-2	1.20-7	2.08-11	1.17-1	6.14-2	6.15-2	1.03-1	1.03-1
	T	0.5	0.795	0.530	0.265	0.133	1.22	0.280	0.718	0.401	0.290	0.270
		1.0	0.801	0.534	0.270	0.135	1.17	0.287	0.730	0.408	0.295	0.275
		1.4	0.808	0.538	0.274	0.137	1.14	0.292	0.740	0.413	0.299	0.279
		2.0	0.818	0.545	0.280	0.140	1.11	0.300	0.755	0.422	0.305	0.284
Ne ³⁺	ΔE	41234.6	41279.5	62441.3	62434.6	-44.9	6.7	21206.7	21161.8	21200.0	21155.1	21155.1
	A	4.84-4	5.54-3	1.27	5.21-1	1.48-6	2.68-9	4.00-1	1.15-1	3.93-1	0.347	0.327
	T	0.6	0.843	0.562	0.308	0.154	1.37	0.323	0.867	0.482	0.368	0.336
		1.0	0.838	0.559	0.313	0.156	1.36	0.343	0.900	0.509	0.373	0.339
		1.4	0.834	0.556	0.312	0.156	1.36	0.355	0.908	0.515	0.374	0.339
		2.0	0.824	0.550	0.309	0.155	1.33	0.370	0.909	0.516		
Na ⁴⁺	ΔE	48313.5	48359.3	73236.4	73201.9	-45.8	34.5	24922.9	24877.1	24888.4	24842.6	24842.6
	A	1.46-3	2.69-2	4.27	1.76	1.55-6	3.04-7	9.18-1	1.29	1.41-1	9.55-1	9.55-1
	T	0.551	0.368	0.239	0.120	0.696	0.438	0.502	0.279	0.201	0.190	0.190
S ⁺	ΔE	14884.8	14853.0	24571.8	24524.9	31.8	46.9	9687.0	9718.8	9640.1	9671.9	9671.9
	A	2.60-4	8.82-4	0.225	9.06-2	3.35-7	1.03-6	0.179	0.133	7.79-2	1.63-1	1.63-1
	T	0.5	4.38	2.92	1.44	0.722	8.15	2.39	4.74	3.36	2.55	1.50
		1.0	4.19	2.79	1.52	0.759	7.59	2.38	4.79	3.38	2.56	1.52
		1.5	4.00	2.67	1.54	0.772	7.11	2.33	4.74	3.32	2.51	1.51
		2.0	3.84	2.56	1.54	0.769	6.71	2.27	4.62	3.21	2.43	1.48
Cl ²⁺	ΔE	18118.6	18053	29907	29812	66	95	11788	11854	11693	11759	11759
	A	7.04-4	4.83-3	0.754	0.305	3.22-6	7.65-6	0.316	0.323	0.100	0.303	0.303
	T	1.88	1.26	1.26	0.627	3.19	1.34	3.33	1.91	1.38	1.24	1.24
A ³⁺	ΔE	21219.3	21090.4	35032.6	34855.5	128.9	177.1	13813.3	13942.2	13636.2	13765.1	13765.1
	A	1.77-3	2.23-2	2.11	0.862	2.30-5	4.94-5	0.598	0.789	0.119	0.603	0.603
	T	0.854	0.570	0.423	0.212	1.35	0.901	2.50	1.24	0.865	1.01	1.01
K ⁴⁺	ΔE	24249.6	24012.5	40080.2	39758.1	237.1	322.1	15830.6	16067.7	15508.5	15745.6	15745.6
	A	4.59-3	8.84-2	5.19	2.16	1.42-4	2.96-4	1.21	1.86	0.141	1.25	1.25
	T	0.455	0.303	0.173	0.086	0.971	0.281	2.12	0.974	0.665	0.884	0.884

TABLE 7. Energy level separations ΔE (cm^{-1}), transition probabilities A (sec^{-1}) and effective collision strengths $T(\tau_e)$ for the np^3 transitions of the N (n=2) and P (n=3) isolectronic sequences.

ION	FARM	T_e (10 ⁴ K)	1D ₂ -3P ₀	1D ₂ -3P ₁	1D ₂ -3P ₂	1S ₀ -3P ₁	1S ₀ -3P ₂	1S ₀ -1D ₂	3P ₀ -3P ₁	3P ₀ -3P ₂	3P ₁ -3P ₂
O ⁺	ΔE	15640.9 7.23-7	15709.6 2.11-3	15867.9 6.34-3	33634.3 7.32-2	33792.6 2.88-4	17924.7 1.22	68.7 1.74-5	227.0 0.0221	158.3 8.92-5	
	A	0.05 0.1	0.0058 0.0151	0.0151 0.124	0.0153 0.266	0.00184 0.0732	0.0310 0.0112	0.0066 0.0222	0.0066 0.0022	0.0027 0.0076	
	T	0.5 0.5 0.5 1.0 1.0 1.5 2.0	0.0221 0.018 0.0732 0.0112 0.0265 0.0292 0.0536	0.0018 0.0732 0.0112 0.0265 0.0292 0.0536 0.207	0.0066 0.0222 0.0022 0.0066 0.0222 0.0292 0.207						
N _e ²⁺	ΔE	24920.3 8.51-6	25197.9 5.42-2	25840.8 1.71-1	55107.7 2.00	55750.6 3.94-3	29909.8 2.71	277.6 1.15-3	920.5 0.18-8	662.9 0.131	
	A	2.24-5 →	1.35 1.34	1.33 1.32	1.32 1.32	0.152 0.151	0.220 0.236	0.185 0.131	0.185 0.527	5.97-3 0.527	
	T	0.5 1.0 1.5 2.0	1.17 1.17	1.17 1.17	1.17 1.17	0.157 0.157	0.284 0.284	0.177 0.177	0.111 0.111	3.04-2 0.471	
Na ³⁺	ΔE	29264 2.24-5 →	29734 1.86-1	30840 6.10-1	65390 7.10	66496 1.05-2	35656 3.46	469.3 5.37-3	1575.6 1.67-7	1106.3 0.111	
	A	5.20-5 →	34149 5.41-1	35932 1.85	75503 2.14+1	77286 2.45-2	41354 4.23	738.7 2.17-2	2521.8 0.156	1783.1 0.0908	
	T	1.02 →	1.02 1.02	1.02 1.02	1.02 1.02	0.146 0.146	0.129 0.129	0.111 0.111	0.111 0.400	1.27-1 0.400	
S ⁰	ΔE	8665.0 3.84-6	8842.6 8.16-3	9238.6 2.78-2	21783.9 3.50-1	22180.0 8.23-3	12941.3 1.53	177.6 3.02-4	573.7 6.71-8	396.1 1.39-3	
	A	3.84-6 →	8.16-3 8.16-3	8.16-3 8.16-3	8.16-3 8.16-3	8.16-3 8.16-3	8.16-3 8.16-3	8.16-3 8.16-3	8.16-3 8.16-3	1.39-3 2.17	
Cl ⁺	ΔE	10657.1 9.82-6 →	10957.6 2.92-2	11653.6 1.04-1	27182.0 1.31	27878.0 1.97-2	16224.4 2.06	300.5 1.46-3	996.5 0.933	696.0 4.57-7	
	A	3.86 →	3.86 3.86	3.86 3.86	3.86 3.86	0.456 0.456	1.15 1.15	0.933 0.933	0.443 0.443	7.57-3 2.17	
	T	4.74 →	4.74 4.74	4.74 4.74	4.74 4.74	0.680 0.680	0.823 0.823	1.18 1.18	0.531 0.531	2.24 2.24	
A ²⁺	ΔE	12439.8 2.21-5 →	12897.9 8.23-2	14010.0 3.14-1	32153.6 3.91	33265.7 4.17-2	19255.7 2.59	458.1 5.17-3	1570.2 2.37-6	1112.1 3.08-2	
	A	4.74 →	4.74 4.74	4.74 4.74	4.74 4.74	0.680 0.680	0.823 0.823	1.18 1.18	0.531 0.531	2.24 2.24	
K ³⁺	ΔE	14062.9 4.34-5 →	14712.7 1.98-1	16384.1 8.14-1	36874.9 1.00+1	38546.3 8.17-2	22162.2 3.18	649.8 1.48-2	2321.2 0.421	1671.4 1.04-1	
	A	1.90 →	1.90 1.90	1.90 1.90	1.90 1.90	0.292 0.292	0.798 0.798	0.421 0.421	0.290 0.290	1.04-1 1.16	
Ca ⁴⁺	ΔE	15554.7 8.42-5 →	16425.6 4.26-1	18830.3 1.90	41431.8 2.31+1	43836.5 1.45-1	25006.2 3.73	870.9 3.54-2	3275.6 0.202	2404.7 3.67-5	
	A	0.904 →	0.904 0.904	0.904 0.904	0.904 0.904	0.116 0.116	0.793 0.793	0.224 0.224	0.224 0.224	3.10-1 0.760	

TABLE 8. Energy level separations ΔE (cm⁻¹), transition probabilities A (sec⁻¹) and effective collision strengths T(T_e) for the np⁴ transitions of the O (n=2) and S (n=3) isoelectronic sequences.

ION	PARM	$T_e (10^4 K)$	$2P_1^o - 2P_2^o$	ION	PARM	$T_e (10^4 K)$	$2P_1^o - 2P_2^o$
Ne^+	ΔE		780.4	$C\lambda^o$	ΔE		882.4
	A	0.5	8.55-3		A		1.24-2
	T	1.0	0.362		T		
		1.5	0.368				
Na^{2+}	ΔE	0.375	A^+	ΔE		1431.6	
	A	2.0	0.381	A		5.27-2	
	T			T		0.635	
Mg^{3+}	ΔE	1366.7	K^{2+}	ΔE		2166.1	
	A	4.59-2		A		1.83-1	
	T	0.300		T		1.78	
Si^{5+}	ΔE	2229.5	Ca^{3+}	ΔE		3117.9	
	A	1.99-1		A		5.45-1	
	T	0.300		T		1.06	

TABLE 9. Energy level separations ΔE (cm^{-1}), transition probabilities A (sec^{-1}) and effective collision strengths $T(T_e)$ for the np^5 transitions of the F ($n=2$) and $C\lambda$ ($n=3$) isolectronic sequences.

PARENT(STATES)	FINAL ION(STATES)	METHOD	ACC	SOURCE
H ⁰ (nl)	H ⁺	Hydrogenic	A+	67
He ⁰ (1S,2S,3S,1P ^o ,3P ^o)	He ⁺ (1s,2s,2P)	CMB;EXPT	A	68-71;72-74
He ^{+(nl)}	He ²⁺	Hydrogenic	A+	67
C ⁰ (3P,1D,1S)	C ⁺ (2P ^o ,4P,2D,2S,P)	BCC+CI,SCC+CI;EXPT	B+,C	75,105;76
C ^{+(2P^o,4P)}	C ²⁺	6CC+CI;CC	D	77;85
C ²⁺ (1S,3P ^o)	C ³⁺	SCC;TDHF	A;	78;79
C ³⁺ (2S)	C ⁴⁺	SCC	A	78,80
N ⁰ (4S,2D,2P ^o)	N ⁺ (3P,1D,1S,5S ^o ,3D ^o ,3P ^o ,1D ^o ,3S ^o ,1P ^o)	SCC+CI,8CC+CI;EXPT	B+	81,82;83
N ^{+(3P,1D,1S)}	N ²⁺ (2P ^o)	6CC+CI;CC	D	84;85
N ²⁺ (ground+excited)	N ³⁺	6CC+CI	A	86
N ³⁺ (1S,2P ^o)	N ⁴⁺	SCC;TDHF	A;	87;79
N ⁴⁺ (2S)	N ⁵⁺	SCC	A	80
O ⁰ (3P,1D,1S)	O ⁺ (4S ^o ,2D ^o ,2P ^o)	CC+CI;CC;EXPT	B+,D;A	88,125;85;89
O ⁰ (4S,2D,2P ^o)	O ²⁺ (3P,1D,S)	CC	D	85
O ²⁺ (3P,D,S,5S ^o)	O ³⁺ (2P ^o ,4P)	8CC+CI;CC	A;D	90;85
O ³⁺ (ground+excited)	O ⁴⁺	6CC+CI	A	91
O ⁴⁺ (1S)	O ⁵⁺	CP;TDHF	B;	92;79
O ⁵⁺ (2S)	O ⁶⁺	SCC	A	80
Ne ⁰ (1S)	Ne ⁺	2CC+CI;MBPT;EXPT	A	93;94;73,95
Ne ^{+(2P)}	Ne ²⁺ (3P,1D,1S,3P ^o ,1P ^o)	5CC+CI	B+	96
Ne ²⁺ (3P)	Ne ³⁺ (4S ^o ,2D ^o ,2P ^o ,4P,2D ^o ,2S,P)	7CC+CI;CC	B+,D	96;85
Ne ³⁺ (4S,2D,2P ^o)	Ne ⁴⁺ (3P,1D,1S,5S ^o ,3D ^o ,3P ^o ,1D ^o ,3S ^o ,1P ^o)	9CC+CI;CC	B+,D	96;85
Ne ⁴⁺ (3P,1D,1S)	Ne ⁵⁺ (2P ^o)	CC	D	85
Ne ⁵⁺ (2P ^o)	Ne ⁶⁺ (1S)	CC	D	85
Na ⁰ (ground+excited)	Na ⁺	IP;EXPT	D	97-99;100,101
Na ⁺ (1S)	Na ²⁺	MBPT	B+	102,103
Na ²⁺ -Na ⁴⁺	Na ³⁺ -Na ⁵⁺	CP		92
Mg ⁰ (1S,P ^o ,1P ^o ,3S)	Mg ⁺	4CC	A	104
Mg ^{+(2S)}	Mg ²⁺	IP	A	43
Mg ²⁺ (1S)	Mg ³⁺	MBPT	B+	102,103
Mg ³⁺ -Mg ⁴⁺	Mg ⁴⁺ -Mg ⁵⁺	CP		92

TABLE 10. Selected bibliography of photoionization cross sections for ions of interest in planetary nebulae studies.

PARENT(STATES)	FINAL ION(STATES)	METHOD	ACC	SOURCE
Si ^o (³ P, ¹ D, ¹ S)	Si ⁺ (² P ^o)	7CC+CI;CC	B+;D	104;106
Si ⁺ (² P ^o)	Si ²⁺	12CC+CI;MBPT;CC		107;108;106
Si ²⁺ (¹ S)	Si ³⁺	5CC;CP	:B	109;92
Si ³⁺ (² S)	Si ⁴⁺	1CC+MP;MP	:B	109;110
Si ⁴⁺ (¹ S)	Si ⁵⁺	MBPT	B+	103
Si ^o (³ P, ¹ D, ¹ S)	S ⁺ (⁴ S ^o , ² D ^o , ² P ^o)	6CC+CI;CC	B+; D	104;111
S ⁺ (⁴ S ^o , ² D ^o , ² P ^o)	S ⁺ (³ S, ¹ D, ¹ S)	10CC+CI;CC	B+; D	104;111
S ²⁺ (³ P, ¹ D, ¹ S)	S ³⁺ (² P ^o)	7CC+CI;CC	B+; D	104;111
S ³⁺ (² P ^o)	S ⁴⁺ (¹ S)	CC	D	111
S ⁴⁺ (¹ S)	S ⁵⁺	5CC;CP	:B	109;92
S ⁵⁺ (² S)	S ⁶⁺	1CC+MP;CP	:B	109;92
Cl ^o (² P ^o)	Cl ⁺	MBPT	B+	112
Cl ⁺ - Cl ⁴⁺	Cl ²⁺ - Cl ⁵⁺	CP	92	
A ^o (¹ S)	A ⁺	COMP;EXPT	A	93,94,113;73,114
A ⁺⁽² P ^o)	A ²⁺ (³ P, ¹ D, ¹ S)	CC	D	106
A ²⁺ (³ P, ¹ D, ¹ S)	A ³⁺ (⁴ S ^o , ² D ^o , ² P ^o)	CC	D	106
A ³⁺ (⁴ S ^o , ² D ^o , ² P ^o)	A ⁴⁺ (³ S, ¹ D, ¹ S)	CC	D	106
A ⁴⁺ (³ P, ¹ D, ¹ S)	A ⁵⁺ (² P ^o)	CC	D	106
A ⁵⁺ (² P ^o)	A ⁶⁺ (¹ S)	CC	D	106
K ^o (² S)	K ⁺	MP;EXPT	D	115;116,117
K ⁺ (¹ S)	K ²⁺	MBPT	B+	102,118
K ²⁺ (² P ^o)	K ³⁺	CC	B+	119
K ³⁺ - K ⁴⁺	K ⁴⁺ - K ⁵⁺	CP	92	
Ca ^o (¹ S)	Ca ⁺	4CC;EXPT	C	104;120,121
Ca ⁺⁽² S)	Ca ²⁺	MP	B	43
Ca ²⁺ (¹ S)	Ca ³⁺	MBPT	B+	102,118
Ca ³⁺ - Ca ⁴⁺	Ca ⁴⁺ - Ca ⁵⁺	CP	92	
Fe ^o	Fe ⁺	MBPT;CP;EXPT		122;123;124
Fe ⁺ - Fe ⁵⁺	Fe ²⁺ - Fe ⁶⁺	CP		123

TABLE 10. (continued)

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