A SURVEY OF ENVIRONMENTAL ¹⁴C LEVELS IN HONG KONG

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ABSTRACT. As an industrialized city, Hong Kong annually consumes a large amount of fossil fuel. In addition, the Daya Bay Nuclear Power Plant in Shenzhen, Guangdong Province, has just begun operation 20 km from Hong Kong. These factors suggest that it may be appropriate and significant to examine the variation of atmospheric ¹⁴C levels in Hong Kong. We have collected and tested a variety of samples from different parts of Hong Kong: terrestrial annual grasses, marine plants and atmospheric CO₂. We measured their ¹⁴C activity and compared it with that of cassia oil samples from Guangxi Province, China. The values obtained indicate that environmental ¹⁴C levels in the Hong Kong region agree with those found in Guangxi, both of which are significantly higher than the levels predicted by Povinec, Chudý and Šivo (1986).

INTRODUCTION

Three ongoing anthropogenic activities are known to alter environmental ¹⁴C levels:

- Since the industrial revolution in the 18th century, much CO₂ has been produced by combustion of fossil fuels, which have no measurable ¹⁴C content. This, the "Suess effect", has decreased the average concentration of atmospheric ¹⁴C, as measured by its activity, by *ca.* 3% over the period 1850–1950 (Freundlich 1979; Levin, Münnich and Weiss 1980; Levin *et al.* 1989).
- 2. Nuclear testing since 1945, particularly nuclear weapons tests, has increased atmospheric ¹⁴C; indeed, by 1963 ¹⁴C from this source amounted to *ca*. 3% of the ¹⁴C content in all global reservoirs; the atmospheric ¹⁴C concentration in the northern hemisphere approached a peak value of double the natural atmospheric ¹⁴C concentration. Since then, the excess atmospheric ¹⁴C has diffused into other reservoirs (mainly to the oceans) and, following the Limited Test Ban Treaty of 1963, the excess atmospheric ¹⁴C has decreased exponentially with a decay half-life of 7 yr (Nydal, Lövseth and Gulliksen 1979; Povinec, Chudý and Šivo 1986).
- 3. Nuclear power has become an important world energy source. Although the ¹⁴C produced by nuclear reactors is small compared to that from weapons testing, the number of nuclear power installations is increasing rapidly, so that ¹⁴C released from this source is not negligible (Obelić et al. 1986; Hertelendi, Uchrin and Ormai 1989; Loosli and Oeschger 1989).

These three factors are all relatively recent and have produced fluctuations over a comparatively short time scale. However, although ¹⁴C concentration has been decreasing since 1968 and the effect of proliferating nuclear installations is counteracted by the combustion of fossil fuel, the current ¹⁴C concentration in the atmosphere is still *ca*. 15% above the natural ¹⁴C level. Further, given projected future development of the nuclear power industry, it is difficult to foresee continued decrease in the atmospheric ¹⁴C level.

This increased level of ¹⁴C may represent a health hazard. ¹⁴C and ³H are incorporated into human tissue through the food chain and are combined during metabolic activity. Their nuclides may then be transferred into genetic material as part of DNA or RNA molecules. In this event, the genetic molecules might be harmed by radioactivity or changed to anomalous molecules by ¹⁴C decay, in

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either case producing a variable genetic factor. Although this possibility is small, any increase in the environmental ¹⁴C level increases the health risk (Stenhouse and Baxter 1979).

The purpose of this study is to investigate environmental ¹⁴C in Hong Kong, one of the most developed cities in southeast Asia. Currently, two major factors impact ¹⁴C in the Hong Kong region: the large quantity of fossil fuel consumption there, and the new Daya Bay Nuclear Power Plant in Shenzhen, Guangdong Province, just 20 km from Hong Kong.

METHODS AND RESULTS

We collected a variety of samples from different parts of Hong Kong, including terrestrial annual grasses and marine plants. We also collected atmospheric CO_2 samples using static absorption of CO_2 in NaOH solution over a period of 5–7 days. Following traditional procedures for ¹⁴C dating, the samples were then prepared and converted to liquid benzene samples (Qiu, Chen and Cai 1990; Radiocarbon Laboratory 1978, 1983). We measured ¹⁴C activity after adding scintillator material to the liquid benzene, using a low-background scintillation counter and Chinese sucrose charcoal (modern carbon standard) as well (Qiu *et al.* 1983).

We used a Model DYS-2 low-background scintillation counter (manufactured by the Biophysics Institute of CASS) with a background of *ca*. 4 cpm per 7 ml benzene and a counting efficiency of *ca*. 70%. All measurements were made in the ¹⁴C laboratory of the Archaeology Institute, CASS, Beijing. Results are listed in Table 1. For comparison, we also analyzed cassia oil samples from Guangxi (Table 2). Figure 1 plots the results of sample analyses for both groups against predicted future ¹⁴C concentrations.

Sample no.	Site	Collection date	Material	Δ ¹⁴ C (‰)
1	Fanling	December 1993	Grass	160 ± 6
2	Kowloon Tong	November 1993	Grass	145 ± 6
2	Chai Wan	December 1993	Grass	148 ± 6
4	Kowloon Tong	January 1994	Atmospheric CO ₂	158 ± 12
5	Stanley	March 1994	Seaweed*	247 ± 20
6	Shek O	March 1994	Seaweed	169 ± 6
0 7	Kowloon Tong	March 1994	Atmospheric CO ₂ †	222 ± 20

TABLE 1. Results of Samples Collected from Hong Kong

*This sample contains too little carbon and too much silicified plant material, which may be some years old; thus, it exhibits unusually high ¹⁴C activity.

†Sample collected after Daya Bay Nuclear Power Plant commenced operation.

Guangxi, China*						
Sample no.	Collection date	Δ ¹⁴ C (‰)				
9001115	1990	200 ± 20				
910809	1991	180 ± 20				
920117	1992	140 ± 20				
920727-8	1992	160 ± 20				
920818-6	1992	150 ± 20				
941072	1994	160 ± 20				
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TABLE 2. Results of Cassia Oil Samples Collected in Guangyi China*

*Cassia oil samples were collected by Chinese Shanghai Essence Ltd. from Guangxi. Only a partial listing of the results is included here.



Fig. 1. Recorded (\frown) and predicted (\frown) atmospheric concentration of ¹⁴C (from Povinec 1986: Fig. 2). Lower curves show the boundary ¹⁴C level for predicted growth of the nuclear industry. \times = cassia oil sample; \otimes = Hong Kong sample; \boxtimes = sugar carbon; O = wheat grains, Beijing (Shen *et al.* 1990)

DISCUSSION

As Hong Kong is situated at a low latitude, far from sites of nuclear weapons testing, one might expect that the environmental ¹⁴C concentration would not fluctuate significantly and would match the global average level. Although the data presented cover a short period and cannot show variation over time, the values obtained indicate that environmental ¹⁴C levels in the Hong Kong area agree with those found in Guangxi province. However, both sets of values are significantly higher than the levels predicted by Povinec, Chudý and Šivo (1986).

The atmospheric ¹⁴C level approached a peak value in 1963 and then began to decrease as an exponential function. Provided there is no further anthropogenic disturbance, the global ¹⁴C level would be expected to decrease to the original natural ¹⁴C level in time (Povinec, Chudý and Šivo 1986). However, the data obtained in Hong Kong and Guangxi (shown in Fig. 1 alongside Povinec's model) suggest the possibility of additional anthropogenic effects in operation. Whether these involve continued weapons testing, effects from the Chernobyl Nuclear Power Station accident, or the steady increase of global nuclear power plant operations, cannot be determined given the data available. Alternately, the decrease in the environmental ¹⁴C level caused by fossil-fuel combustion may be less than expected.

Although more data need to be collected over a longer period, the present data can serve as a base for monitoring any effects from the Daya Bay Nuclear Power Plant, which began operating on 1 February 1994. Normally, the influence of ¹⁴C released from nuclear installations may be negligible at a distance beyond 10 km (McCartney *et al.* 1986). Because Hong Kong is 20 km from Daya Bay, its fluctuations in ¹⁴C level are apparently due mainly to the influence of nuclear testing.

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