

Distribution and Isotopic Compositions of *n*-Alkanes and *n*-Alkenes in Cryoconites from the Glaciers of Tibetan Plateau

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Abstract

In the current study, cryoconite samples were collected from six glaciers on the Tibetan Plateau to analyze the *n*-alkanes and *n*-alkenes. The findings revealed that the concentrations of *n*-alkanes and *n*-alkenes varied from 40.1 to 496.1 $\mu\text{g g}^{-1}$ and 4.6 to 13.8 $\mu\text{g g}^{-1}$, respectively. The CPI of the long-chain *n*-alkanes ranged from 3.3 to 8.4, and the ACL ranged from 28.7 to 29.3. Moreover, the $\delta^{13}\text{C}$ of the *n*-alkanes in cryoconites were within the range of C_3 plants, demonstrating that the *n*-alkanes in cryoconites were only derived from vascular plants. However, the $\delta\text{D}_{\text{mean}}$ were more negative than that of C_3 plants, which could be caused by dry and humid condition of glacier. Unlike *n*-alkanes, the *n*-alkenes ranged from $\text{C}_{17:1}$ to $\text{C}_{30:1}$ showed a weak even-over-odd carbon number preference in the DKMD, YZF, LHG and TS glaciers, but a weak odd carbon preference in the QY glacier. The *n*-alkenes in the YL Snow Mountains showed an obvious odd-over-even carbon number predominance from $\text{C}_{17:1}$ to $\text{C}_{22:1}$ with C_{max} at $\text{C}_{19:1}$, and the even-over-odd carbon number preference from $\text{C}_{23:1}$ to $\text{C}_{30:1}$ with C_{max} at $\text{C}_{28:1}$. This demonstrated that the *n*-alkenes of cryoconite may be mainly derived from *in-situ* production in glacier.

Keywords: Tibetan Plateau; Glacier; Cryoconite; *n*-alkane; *n*-alkene

1. Introduction

Cryoconites are dark-colored, spherical granules on the glacier surface (Langford et al., 2014; Cook et al., 2016; Takeuchi et al., 2018). Cryoconite granules are often composed of filamentous cyanobacteria entangled with minerals and organic particles (Takeuchi et al., 2014). Due to their darker tone, cryoconites can absorb more solar radiation than the surrounding ice, which promote the melting of the ice beneath the granules and form cryoconite holes (Chandler et al., 2015; Tedesco et al., 2016). Cryoconite holes are regarded as unique glacier ecosystems, in which liquid water is available on the glacier surface during the ablation season. Therefore, cryoconite holes could act as biologically active habitats for a diverse community of microorganisms, including algae, cyanobacteria, bacteria, archaea, viruses, rotifers, tardigrades, and insects (Bagshaw et al., 2016; Zhou et al., 2019; Rozwalak et al., 2022). Generally, the algae and cyanobacteria usually grow at the bottom of cryoconite holes, and they produce organic matter through photosynthesis, which supports the proliferation of heterotrophs (Takeuchi et al., 2018).

The organic matter accumulated on cryoconites usually has both autochthonous and allochthonous sources (Hood et al., 2009; Singer et al., 2012; Stibal et al., 2012). The autochthonous organic matter in cryoconites is produced *in situ* via heterotrophic or autotrophic microbial activities (Anesio et al., 2009; Smith et al., 2017), whereas allochthonous organic matter (such as levoglucosan) may be derived from biomass burning (Li et al., 2019). However, the particles deposited on the glacier as well as the autochthonous and allochthonous organic matter are darker, so they reduce the albedo of the glacier surface and accelerate glacier melting (Takeuchi et al., 2001, 2002; Rozwalak et al., 2022). Up to now, there have been only sporadic reports on organic matters in cryoconites. Xu et al. (2010) found normal alkanes in cryoconites from Athabasca glacier of Mount Rocky, Canada were derived from mossy and vascular plant origin. They also detected higher concentrations of phospholipid fatty acid, indicating that the glacier surface were dominated by gram positive and gram negative bacteria, as well as cyanobacteria. Pautler et al. (2013) also found that Antarctic cryoconites contained microbial proteins, peptides and phospholipid fatty acids. Above results suggested these organic matters in cryoconites from Arctic and Antarctic were derived

from the combined contributions of both higher plants and *in situ* microorganism's activities.

Normal alkanes (*n*-alkanes), which are stable, long-lived and non-polar saturated hydrocarbon molecules that originate from epicuticular waxes of vascular higher plants (Eglinton and Hamilton, 1967; Bush and McInerney, 2013; Diefendorf and Freimuth, 2017). Short chain *n*-alkanes in the range of C₁₄–C₂₂ are produced mainly by bacteria, algae and microbial organisms (Han and Calvin, 1969; Grimalt and Albaiges, 1987). Mid-chain *n*-alkanes (C₂₃–C₂₅) have been mainly detected in aquatic macrophytes (Eglinton and Calvin, 1967; Ficken et al., 2000). *n*-Alkanes are strongly recalcitrant due to their hydrophobic nature and excellent chemical stability during transportation, deposition, and burial (Schwark et al., 2002). Moreover, the stable carbon isotope compositions ($\delta^{13}\text{C}$) of the *n*-alkanes in different matrix have been widely employed to understand their source regions because they are sensitive to the plant types (C₃ or C₄) they are derived from (Chikaraishi and Naraoka, 2003; Schefuß et al., 2005; Hockun et al., 2016). In addition, the hydrogen isotope ratios (δD) of higher plant waxes, primarily reflect the δD of the precipitation during photosynthesis (Sachse et al., 2006, 2012). Hence, the combination of distribution and dual-isotope ratios ($\delta^{13}\text{C}$ and δD) of *n*-alkanes provides a powerful tool for accessing their source regions (Diefendorf and Freimuth, 2017; Zhang et al., 2023). Compared with the *n*-alkanes, the *n*-alkenes are geologically unstable due to the sensitivity of the double bond to oxidation. The long chain *n*-alkanes and *n*-alkenes have been detected previously in Antarctic soils (Matsumoto et al., 1990), peat (Xie et al., 2004) and sediments (Zhang et al., 2015; Pu et al., 2017). However, investigations for these compounds in the glacier is still a fairly novel field. Therefore, to fill this information gap, the cryoconite samples were collected from the six glaciers of the Tibetan Plateau (TP) to probe the molecular distribution characteristics of *n*-alkanes and *n*-alkenes in cryoconites and discussed their potential sources based on the $\delta^{13}\text{C}$ and δD . For the first time, this paper provides an account of the unusual distribution of *n*-alkenes in cryoconite from the TP and discusses their possible origins.

2. Materials and Methods

2.1. Study Site

During July and August 2014, cryoconite samples were collected from six glaciers which are presented in Fig. 1 and the sampling informations are listed in Table 1. The Urumqi (TS) glacier No. 1 is located on the eastern side of the Tianshan Mountains (Fig. 1). The Laohugou (LHG) glacier

No. 12 and the Qiyi (QY) glacier are located on the northern slope of the Qilian Mountains on the northeastern TP. The Yuzhufeng (YZF) glacier is located in the eastern Kunlun Mountains on the northern TP. These four glaciers are affected by the westerlies throughout the year. The Dongkemadi (DKMD) glacier is located on the northern slope of the Tanggula Mountains on the central TP, which is the transitional region of the South Asian monsoon and westerly winds. The Baishui glacier No.1 (YL Snow Mountains) is located on the southeastern TP, which has a typical Indian monsoon climate.

Figure 1. Near hear

Table 1. Near hear

2.2. Sample Collection

The cryoconite samples were collected from the ablation areas of each glacier using a sterile stainless-steel scoop. All of the samples were sealed and immediately labeled and stored in $-20\text{ }^{\circ}\text{C}$ freezer until further analysis.

2.3. Extraction and Measurement

The cryoconite samples were lyophilized at $-70\text{ }^{\circ}\text{C}$ and ground into powder. Cryoconite powder (20 g) was extracted using a Soxhlet extractor with dichloromethane and methanol (9:1 v/v) for 72h. The aliphatic hydrocarbon fractions were loaded onto precleaned silica gel columns and were eluted with hexane. The aliphatic hydrocarbons fractions were determined by a gas chromatography–mass spectrometry (GC–MS) using an Agilent 7890A gas chromatograph coupled with an Agilent 5975C mass spectrometer. The chromatographic column was a DB-1 MS capillary column (30 m \times 0.25 mm \times 0.25 μm). The GC oven temperature was programmed from 70 to 320 $^{\circ}\text{C}$ by 4 $^{\circ}\text{C}/\text{min}$ and held for 30 min. Samples were injected in splitless mode (1 μL) and high purity helium (99.999%) was used as the carrier gas at a flow rate of 1.0 mL/min. The ion source temperature was 250 $^{\circ}\text{C}$, and ionization energy was 70 eV. The full scanning mode was used to select the characteristic ions (m/z 85) for the detection of *n*-alkanes. The identification and quantification of *n*-alkanes were achieved by comparing peak areas with external *n*-alkane standards (mixture of C₂₁, C₂₅, C₂₇, C₂₉, C₃₁, and C₃₃) of known concentration.

Compound-specific carbon and hydrogen isotope were measured using an HP 6890 GC, interfaced with a Finnigan MAT Delta plus XL isotope ratio mass spectrometry (IRMS) instrument by a high temperature pyrolysis reactor. In all analyses, an Agilent HP-1 MS column (30 m \times 0.32 mm

i.d., 0.25 μm film thickness) capillary column was used, and helium was used as the carrier gas. $\delta^{13}\text{C}$ and δD values are reported in per mil (‰) relative to Vienna Pee Dee Belemnite (VPDB) and Vienna Standard Mean Ocean Water (VSMOW), respectively. Each sample was independently measured three times. Isotopic measurements were calibrated by coinjected 4–6 *n*-alkane standards. Detailed experiment procedure of compound-specific carbon and hydrogen isotope can be found in Bi et al. (2005). Achieved precision, expressed as the average standard deviation, was 4‰ for the standard mixtures.

2.4. *n*-Alkane Indices

The carbon preference index (CPI) and the average chain length (ACL) were used as *n*-alkane indices. The CPI measures the odd-over-even carbon number predominance of the *n*-alkane, which is used to characterize the sources of *n*-alkane (Eglinton and Hamilton, 1967). The ACL of *n*-alkanes from the higher plant is predominantly a function of vegetation type (Li et al., 2013). The CPI and ACL were calculated as follows:

$$CPI_{n\text{-alkanes}} = \frac{1}{2} \times \left[\frac{\sum_{\text{odd}} (C_{25} - C_{33})}{\sum_{\text{even}} (C_{24} - C_{32})} + \frac{\sum_{\text{odd}} (C_{25} - C_{33})}{\sum_{\text{even}} (C_{26} - C_{34})} \right] \quad (1)$$

$$ACL_{n\text{-alkanes}} = \frac{27 \times C_{27} + 29 \times C_{29} + 31 \times C_{31} + 33 \times C_{33} + 35 \times C_{35}}{C_{27} + C_{29} + C_{31} + C_{33} + C_{35}} \quad (2)$$

$$CPI_{18:1-24:1} = \frac{1}{2} \times \left[\frac{\sum_{\text{odd}} (C_{19:1} - C_{23:1})}{\sum_{\text{even}} (C_{18:1} - C_{22:1})} + \frac{\sum_{\text{odd}} (C_{19:1} - C_{23:1})}{\sum_{\text{even}} (C_{20:1} - C_{24:1})} \right] \quad (3)$$

$$CPI_{23:1-30:1} = \frac{1}{2} \times \left[\frac{\sum_{\text{odd}} (C_{23:1} - C_{29:1})}{\sum_{\text{even}} (C_{22:1} - C_{28:1})} + \frac{\sum_{\text{odd}} (C_{23:1} - C_{29:1})}{\sum_{\text{even}} (C_{24:1} - C_{30:1})} \right] \quad (4)$$

Odd-over-even predominance (OEP), similar to CPI, is another *n*-alkane ratio proxy for the predominance of odd over even. The OEP of *n*-alkanes and *n*-alkenes were calculated as follows:

$$OEP_{n\text{-alkanes}} = \frac{C_{27} + C_{29} + C_{31}}{C_{26} + C_{28} + C_{30}} \quad (5)$$

$$OEP_{17:1-22:1} = \frac{C_{17} + C_{19} + C_{21}}{C_{18} + C_{20} + C_{22}} \quad (6)$$

$$OEP_{23:1-30:1} = \frac{C_{23:1} + C_{25:1} + C_{27:1} + C_{29:1}}{C_{24:1} + C_{26:1} + C_{28:1} + C_{30:1}} \quad (7)$$

3. Results and Discussion

3.1. Concentrations of the Aliphatic Hydrocarbons in Cryoconites

The concentrations of the *n*-alkanes and *n*-alkenes in cryoconite samples of different glaciers varied from 36.5 to 496.1 $\mu\text{g g}^{-1}$ and from 4.8 to 17.0 $\mu\text{g g}^{-1}$, respectively (Fig. 2). The average concentration of *n*-alkanes was highest in the YL Snow Mountains ($496.1 \pm 68.0 \mu\text{g g}^{-1}$) and the lowest in the DKMD glacier ($36.5 \pm 3.9 \mu\text{g g}^{-1}$). The average concentrations of *n*-alkanes were $118.9 \pm 5.7 \mu\text{g g}^{-1}$ in the TS glacier, $76.0 \pm 2.7 \mu\text{g g}^{-1}$ in the LHG glacier, $70.2 \pm 8.2 \mu\text{g g}^{-1}$ in the QY glacier, and $49.8 \pm 6.4 \mu\text{g g}^{-1}$ in the YZF glacier. Similarly, the highest total concentration of *n*-alkenes was measured in the YL Snow Mountains ($17.0 \pm 5.0 \mu\text{g g}^{-1}$), while the lowest concentration was measured in the YZF glacier ($4.8 \pm 2.1 \mu\text{g g}^{-1}$) rather than the DKMD glacier ($7.2 \pm 0.7 \mu\text{g g}^{-1}$). Overall, *n*-alkanes are 4.6 to 29.3 times larger than the corresponding carbon number *n*-alkenes in six glaciers. Moreover, the concentrations of *n*-alkane in the cryoconites were much higher than those previously reported in Antarctic soil (0.013 to 2.2 $\mu\text{g g}^{-1}$ from *n*-C₁₄ to *n*-C₃₅) (Matsumoto et al., 1990), but lower than those in the cryoconites reported in Western Canada (290 to 2990 $\mu\text{g g}^{-1}$ from *n*-C₁₈ to *n*-C₃₃) (Pautler et al., 2013) and Antarctica (800 to 13400 $\mu\text{g g}^{-1}$ from *n*-C₁₈ to *n*-C₃₃) (Xu et al., 2010).

Fig. 2. near here

3.2. Distributions Characteristic of *n*-Alkanes

Vascular plants typically show a strong odd-over-even carbon number predominant distribution, with a maximum abundance at *n*-C₂₇, *n*-C₂₉, or *n*-C₃₁ (Eglinton and Hamilton, 1967; Collister et al., 1994). The CPI values of *n*-alkanes from higher plants are generally > 5 (Eglinton and Hamilton, 1967), whereas the *n*-alkanes from lower organisms, such as bacteria and algae, as well as fossil fuels, have low CPI values of close to 1 (El Nemr et al., 2016). The average relative abundance of the *n*-alkanes and *n*-alkene in the cryoconites of different glaciers are shown in Fig. 3. The cryoconite samples in the DKMD, YZF, QY, LHG, and TS glaciers contain a suite of *n*-alkanes ranging from *n*-C₁₄ to *n*-C₃₅, with C_{max} values at *n*-C₂₉ or *n*-C₃₁. This finding is comparable with the distribution of *n*-alkanes in 154 lacustrine surface sediment on the TP (Xia et al., 2008). The *n*-alkanes in the YL Snow Mountains show the wider distribution ranged from *n*-C₁₃ to *n*-C₃₇ compared with the other five glaciers. However, the *n*-alkanes of *n*-C₁₈, *n*-C₁₉ and *n*-C₂₀ in the DKMD, YZF,

QY, LHG, TS glaciers and $n\text{-C}_{19}$ in the YL Snow Mountains are below detection limit. The GC-MS figure has been showed in the [Supplementary Information 1](#).

[Fig. 3. near here](#)

As shown in Table 2, the average CPI values of the $n\text{-C}_{25}$ to $n\text{-C}_{33}$ n -alkanes in the cryoconites varied from 3.3 in the YL Snow Mountains to 8.4 in the QY glacier ([Table 2](#)), similar to the Japan Sea Sediments (Yamada and Ishiwatari, 1999), showing an obvious odd-over-even carbon number predominance, which indicates that the n -alkanes in cryoconites derived from higher plants. The CPI value is the lowest in the YL Snow Mountains, because the higher CPI values usually correlate with cold and dry environments, whereas the smaller CPI values usually correlate with warm and humid environments (Ankit et al., 2017). In addition, CPI values were also related to local vegetation type (Bai et al., 2019). The OEP value of the n -alkane in the polluted substrates is in the range of 1.0 to 1.2 (Snedaker et al., 1995), so OEP can determine whether the sedimentary area is polluted by petroleum hydrocarbons (Snedaker et al., 1995). The OEP of cryoconites ranged from 2.7 to 8.6, indicating that cryoconites on the TP glaciers were not polluted by petroleum and its derivatives. However, previous studies showed that unpolluted environments had stable ACL values (Sikes et al., 1993). ACL of greater than 27 represents an input of terrigenous higher plants (Ankit et al., 2017). In these six glaciers, the ACL of the cryoconite sample varied from 28.8 to 29.3, it was basically stable at about 29, indicating that the main source of n -alkanes in cryoconites was terrestrial higher plants.

For n -alkanes, the $n\text{-C}_{27}$ and $n\text{-C}_{29}$ are diagnostic of woody plants while the $n\text{-C}_{31}$ is diagnostic of herbaceous plants (Meyers et al., 2003). Therefore, the $n\text{-C}_{27}/n\text{-C}_{31}$ ratio of n -alkanes is usually used to evaluate the relative contribution of herbaceous and woody plants (Meyers et al., 2003; Bush and Mcinerney, 2013). The ratio of $n\text{-C}_{27}/n\text{-C}_{31} < 1$ indicates an input increase of herbaceous plant, and the ratio of $n\text{-C}_{27}/n\text{-C}_{31} > 1$ indicates an input increase of woody plants. The ratio of $n\text{-C}_{27}/n\text{-C}_{31}$ in cryoconites ranged from 0.50 and 0.89, which indicates n -alkanes in TP glacier could be mainly derived from herbaceous plants. The $\Sigma\text{C}_{21}^-/\Sigma\text{C}_{22}^+$ ratio of n -alkanes reflect the relative abundance changes of lower bacterial algae organisms and higher plants (Meyers et al., 2003). When $\Sigma\text{C}_{21}^-/\Sigma\text{C}_{22}^+ < 1$, indicating that the soils are in the early development stage and greatly influenced by the input of terrigenous higher plants. The $\Sigma\text{C}_{21}^-/\Sigma\text{C}_{22}^+$ of n -alkanes is between 0.09 to 0.16,

indicating that *n*-alkanes in cryoconites mainly come from terrigenous higher plants, and cryoconite has not been formed into soil.

Table 2. near hear

3.3. Distributions and Source of *n*-Alkenes

The *n*-alkenes ranged from C_{17:1} to C_{30:1} with a C_{max} value of *n*-C_{19:1} were detected in all cryoconite samples (Fig. 3). The relatively *n*-alkene abundance of C_{17:1}–C_{20:1} in DKMD, YZF, QY, LHG, and TS glaciers were higher than that of C_{21:1}–C_{30:1}. Moreover, *n*-alkenes with C_{17:1}–C_{30:1} were also detected in lower concentration than the corresponding *n*-alkane. The *n*-alkenes in the YL Snow Mountains showed a bimodal distribution ranging from C_{17:1} to C_{33:1}. The main peak was C_{19:1} and the second peak was C_{28:1}, which was in line with the occurrence of the *n*-alkenes in lake sediments (Pu et al., 2017) and Antarctic soils (Matsumoto et al., 1990). The CPI_{17:1-23:1} and OEP_{17:1-22:1} of *n*-alkenes were 4.40 and 6.32, respectively. Meanwhile, the CPI_{23:1-30:1} and OEP_{23:1-30:1} of *n*-alkenes were 0.39 and 0.37 in the YL Snow Mountains, respectively. This demonstrated an obvious odd-over-even carbon number predominance from C_{17:1} to C_{22:1}, and an even-over-odd carbon number preference from C_{23:1} to C_{30:1} in the YL Snow Mountains. The CPI_{17:1-23:1} and OEP_{17:1-22:1} of *n*-alkenes in the DKMD, YZF, LHG and TS glaciers ranged from 0.89 to 1.35 and from 0.76 to 0.98, respectively. The CPI_{23:1-30:1} and OEP_{23:1-30:1} of *n*-alkenes in the DKMD, YZF, LHG and TS glaciers ranged from 0.56 to 0.96 and from 0.65 to 0.91, respectively. This showed a weak even-over-odd carbon number preference from C_{17:1} to C_{30:1} in these four glaciers. In contrast, the CPI_{17:1-23:1} and OEP_{17:1-22:1} of *n*-alkenes in the QY glacier were 1.87 and 1.03, respectively. The CPI_{23:1-30:1} and OEP_{23:1-30:1} of *n*-alkenes in the QY glacier were 1.15 and 1.12, respectively. This showed a weak odd-over-even carbon number predominance from C_{17:1} to C_{30:1} in the QY glacier. The $\Sigma C_{21}^- / \Sigma C_{22}^+$ of *n*-alkenes were 0.69 in the YL Snow Mountains, and ranged from 5.56 to 23.40 in the other five glaciers, demonstrating that *n*-alkenes in the YL Snow Mountains has different source from the other five glaciers.

The *n*-alkenes coexisted with *n*-alkanes in the cryoconite, but it was inferred that *n*-alkene has different source with *n*-alkane due to the unusual distribution characteristic, different carbon number ranges and C_{max} between them. Previous studies indicated that *n*-alkenes could originate from rare cases the epicuticular waxes of higher plants (Grimalt and Albaigés, 1990; Pu et al., 2018), algae,

fungi and cyanobacteria organisms living in aquatic environment (Patterson et al., 1967; Gelpi et al., 1968, 1970; Matsumoto et al., 1990), the reduction of diagenesis of monounsaturated fatty acids for the main source of *n*-alkene (Ekpo et al., 2005), and the microbial transformation of the corresponding *n*-alkanes or direct inputs from organisms (Jaffé et al., 2001). The relatively *n*-alkene abundance of C_{17:1}-C_{20:1} in DKMD, YZF, QY, LHG, and TS glaciers were higher than that of C_{21:1}-C_{30:1}, and the *n*-alkanes of C₁₈, C₁₉ and C₂₀ were below detection limit. This indicated that C_{17:1}, C_{18:1}, C_{19:1} and C_{20:1} *n*-alkenes in DKMD, YZF, QY, LHG and TS glaciers could be derived from the microbial transformation of the corresponding *n*-alkanes. However, *n*-alkenes were not detected in aerosols and the surface soil around the TP glacier (Matsumoto et al., 1990; Bai et al., 2014), which combined with the special distribution in the cryoconites determines that the *n*-alkenes in the cryoconites is from microorganism.

Cryoconite holes are covered by ice lids and snow accumulation on the glacier, which can keep them isolated from the atmosphere. Therefore, wind-borne material cannot be deposited in cryoconite holes in winter (Foreman et al., 2007). During the summer melting season, the snow and ice on the glacier slowly melt, more and more cryoconite holes are opened to the atmosphere, and more microbes enter or *in situ* production in the cryoconite holes (Musilova et al., 2017). According to the previous reports (Takeuchi et al., 2001), a large amount of snow algae is contained in the cryoconite. Singer et al. (2012) identified a prominent population of unsaturated aliphatic lipids and peptides in glaciers, which supports the *in situ* production of these compounds by microorganisms. Antony et al. (2014) reported that the aliphatic molecules (double bond equivalents per carbon atom) were most likely produced from microbial and algal biomass, constituting approximately 37-52% of the total formulas assigned to each glacial ice sample. Moreover, a large number of cyanobacteria have been identified on glacial surfaces on the TP (Takeuchi and Li, 2008; Liu et al., 2009; Feng et al., 2016). Therefore, we conclude that *n*-alkene identified in the cryoconites from the TP glacier may have been mainly produced *in situ* by various microbes. More knowledge is needed about the biosynthetic origins of long-chain *n*-alkenes and factors deciding their accumulation in cryoconites.

4. Carbon Stable Isotope Compositions of *n*-Alkanes

Fig. 4 illustrates the $\delta^{13}\text{C}$ values of the *n*-alkanes (*n*-C₂₇, *n*-C₂₉, *n*-C₃₁) in the cryoconite samples of six glaciers. Generally, the $\delta^{13}\text{C}$ values of the *n*-C₂₉ and *n*-C₃₁ alkanes in most glaciers were more

negative than those of the *n*-C₂₇ alkanes. The difference was likely due to the involvement of different biosynthetic pathways of these long-chain *n*-alkanes (Collister et al., 1994). In this study, the $\delta^{13}\text{C}$ values of the *n*-C₂₉ and *n*-C₃₁ *n*-alkanes were more negative in the YL Snow Mountains than those of the other five glaciers. The reason may be that YL Snow Mountains is located on the southeastern TP and is a part of the southern Hengduan Mountains, which have a warm, humid climate and dense vegetation surrounded. In contrast, cold and dry climate highly influence the other five glaciers, because the compound-specific carbon isotopic values of the long chain *n*-alkanes are sensitive to regional moisture (Wiltshire et al., 2023). Previous study indicated that ^{13}C enrichment in plants occurs under drought stress because leaf stomata are closed to prevent transpiration. Therefore, the ratio of intercellular to the ambient CO₂ concentration decreases, resulting in increased carbon isotope fractionation (Diefendorf and Freimuth, 2017). DKMD and YZF glaciers in the central TP is surrounded by Taklamakan Desert and the Qaidam Basin with sparse vegetation coverage and arid climate. In summer, the evaporation of the soil water and leaf surface water is stronger in the central TP than the YL Snow Mountains, QY, LHG and TS glaciers. Therefore, the largest $\delta^{13}\text{C}$ values of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ *n*-alkanes ranged from -32.0% to -31.4% in the YZF glacier and from -33.5% to -31.7% in the DKMD glacier. The TS, QY, and LHG glaciers are all located on the northern edge of the TP, which has a lower relative humidity than the southeastern TP. The $\delta^{13}\text{C}$ values of the *n*-C₂₇, *n*-C₂₉ and *n*-C₃₁ *n*-alkanes detected in the northern edge of TP were at intermediate levels, ranging from -33.7% to -32.9% in the TS glacier, from -33.3% to -30.8% in the QY glacier, and from -34.5% to -31.9% in the LHG glacier. Similar *n*-alkane distributions have been documented in the snowpits from the TP glacier (Xie et al., 2000; Li et al., 2009), fresh snow in Sapporo, northern Japan (Yamamoto et al., 2001), and fresh snow in Hokkaido, Japan (Sankelo et al., 2013).

[Fig. 4. near here](#)

The $\delta^{13}\text{C}$ of *n*-alkanes can provide information on the source of organic materials (Aichner et al., 2010a). For example, plants with different photosynthetic pathways have distinctive $\delta^{13}\text{C}$ values: C₃ plants produce long chain *n*-alkanes with $\delta^{13}\text{C}$ values between -32% and -39% , while C₄ plants have $\delta^{13}\text{C}$ values between -18% and -25% (Collister et al., 1994). A quantitative estimate of the relative contribution of C₃ plants, compared to that of C₄ plants, was conducted using the $\delta^{13}\text{C}$

of the *n*-alkanes and a binary mixing model. The isotopic composition and relative abundance of *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ alkanes in each cryoconite sample were used to calculate their weighted mean $\delta^{13}\text{C}$ values ($\delta^{13}\text{C}_{\text{mean}}$) (Eq.8). According to a previous study, the endmember values for wax *n*-alkanes should be -36‰ for C₃ plants and -21‰ for C₄ plants (Collister et al., 1994). In Equation (8), C_n is the concentration (units). The relative contribution of C₃ plants (*f*) was calculated using the following Equation (9). The hydrogen isotope compositions and relative abundances of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ alkanes in each cryoconite sample were used to calculate their weighted mean δD values ($\delta\text{D}_{\text{mean}}$) Equation (10).

$$\delta^{13}\text{C}_{\text{mean}} = \frac{\delta^{13}\text{C}_{27} \times C_{27} + \delta^{13}\text{C}_{29} \times C_{29} + \delta^{13}\text{C}_{31} \times C_{31}}{C_{27} + C_{29} + C_{31}} \quad (8)$$

$$\delta^{13}\text{C}_{\text{mean}} = f \times (-36\text{‰}) + (1 - f) \times (-21\text{‰}) \quad (9)$$

$$D_{\text{mean}} = \frac{\delta\text{D}_{27} \times C_{27} + \delta\text{D}_{29} \times C_{29} + \delta\text{D}_{31} \times C_{31}}{C_{27} + C_{29} + C_{31}} \quad (10)$$

As shown in Table 3, the weighted mean $\delta^{13}\text{C}$ values ($\delta^{13}\text{C}_{\text{mean}}$) of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ *n*-alkanes in all cryoconite ranged from -31.7‰ to -34.8‰ . The $\delta^{13}\text{C}_{\text{mean}}$ values were $-34.8\text{‰} \pm 1.4$ in the YL Snow Mountains, $-32.9\text{‰} \pm 1.3$ in the DKMD glacier, $-31.7\text{‰} \pm 0.3$ in the YZF glacier, $-32.2\text{‰} \pm 1.3$ in the QY glacier, $-33.7\text{‰} \pm 1.4$ in the LHG glacier and $-33.2 \pm 0.5\text{‰}$ in the TS glacier. The application of the two endmember mixing model revealed that C₃ plants contributed 71.3 – 92.2% of the total *n*-alkanes in the cryoconites from all glaciers (Table 3), indicating that C₃ plants were the dominant source of the *n*-alkanes in the cryoconites on the TP glaciers.

[Table 3. near here](#)

5. Hydrogen Stable Isotope Compositions of *n*-Alkanes

In all cryoconites, the δD values of the predominant odd carbon number *n*-alkanes could be measured, but that of the even carbon number *n*-alkanes could not be measured due to their low abundances. The measured δD values of the odd carbon number *n*-alkanes (*n*-C₂₇, *n*-C₂₉, and *n*-C₃₁) and weighted mean δD values ($\delta\text{D}_{\text{mean}\text{‰}}$) of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ were listed in Table 3. Previous studies indicated that the $\delta\text{D}_{\text{wax}}$ value of woody, shrub, and herbaceous plants decreased sequentially (Hou et al., 2007). In this study, the C_{max} value was *n*-C₃₁ in the cryoconite of the DKMD and YZF glacier, which was consistent with the vegetation coverage around the DKMD and YZF glacier, indicating that the terrestrial *n*-alkanes in the cryoconite of these two glaciers mainly came from

herbaceous plants around the glaciers. Therefore, the δD_{mean} of the DKMD glacier was the most negative (-203.5 ± 9.1) among the six glaciers. However, the YZF glacier was located near the Qaidam Basin with cold and dry climate, so the δD_{mean} value of the YZF glacier was relatively more positive (-174 ± 6.1) than DKMD glacier. In the cryoconite of the other four glaciers, including YL Snow Mountains, QY, LHG and TS glacier, the C_{max} was *n*-C₂₉ in these four glaciers, which mainly came from woody plants and shrubs. The vegetation and precipitation around YL Snow Mountains was more abundant than that of the other five glaciers. As a result, the δD_{mean} value of YL Snow Mountains was the only second negative (-183.5 ± 7.2) to the DKMD glacier. The δD_{mean} values from DKMD and YL Snow Mountains were more negative than that from the other four glaciers, consistent with the local river and water vapor isotopic characteristics (Li et al., 2017). The TS, QY and LHG glaciers all belong to cold and dry climate, but the air temperature and precipitation in the TS glacier was higher than the QY and LHG glacier. However, the woody plants, shrubs and vegetation in the TS glacier were more abundant than QY and LHG glaciers. Therefore, the δD_{mean} value of TS glacier (-178.1 ± 13.6) was more negative than that of QY glacier (-170.7 ± 6.3) and LHG glacier (-162.7 ± 6.8).

The plot of $\delta^{13}\text{C}$ versus δD of six glaciers was presented in Fig. 5. The weighted mean $\delta^{13}\text{C}$ values of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ alkanes ranged from -33.3‰ to -35.0‰ in the all cryoconite samples were within the range of C₃ leaf wax (-30‰ to -37.9‰) (Bi et al., 2005). Moreover, the $\delta^{13}\text{C}_{\text{mean}}$ values of the *n*-alkanes in the cryoconites were consistent with the snow in Hokkaido (-28.2‰ to -34.4‰) (Sankelo et al., 2013) and lake sediment on the TP (-20.3‰ to -35.8‰) (Aichner et al., 2010a). The weighted mean δD values of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ alkanes ranged from -153.0‰ to -212.3‰ in the cryoconites from the TP glaciers, which were within the range of the δD values of the *n*-alkane in snow from Hokkaido (-169.9‰ to -223.1‰) (Sankelo et al., 2013), lake sediments on the TP (-158.0‰ to -237‰) (Aichner et al., 2010b), and soil on the TP (-142.3‰ to -277.1‰) (Luo et al., 2011; Bai et al., 2014). However, these δD_{mean} values were more negative than those of C₃ plant leaf wax (-95.9‰ to -209.1‰) (Bi et al., 2005), which could be potentially caused by glacial environment and the local climate due to low temperature.

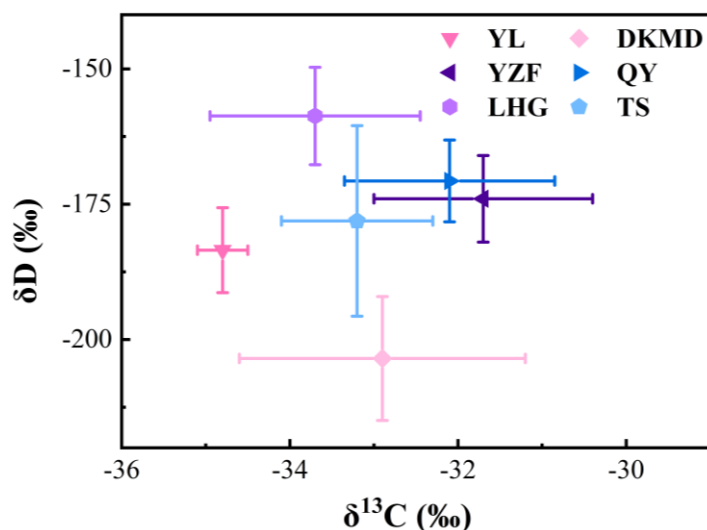


Fig. 5. near here

5. Conclusions

This study provides a valuable information about the distribution and sources of *n*-alkanes and *n*-alkenes in the cryoconite samples of six glaciers on the TP. The *n*-alkanes ranged from *n*-C₁₄ to *n*-C₃₅, with a C_{max} value at *n*-C₂₉ or *n*-C₃₁ in the DKMD, YZF, QY, LHG, and TS glaciers, but showed the wider distribution ranged from *n*-C₁₃ to *n*-C₃₇ in the YL Snow Mountains. The CPI is higher in the DKMD, YZF, QY, LHG, and TS glaciers under cold and dry environments and the lowest in the YL Snow Mountains due to the higher environmental air temperature and precipitation. The ratio of *n*-C₂₇/*n*-C₃₁ ranged from 0.50 and 0.89, which indicates *n*-alkanes in TP glacier mainly derived from herbaceous plants. The $\delta^{13}\text{C}$ and δD values of the *n*-alkanes demonstrated that the isotopic signature of *n*-alkanes in cryoconites correlated with that derived only from vascular plants. The *n*-alkene may have been mainly produced *in situ* by various microbes in the cryoconite due to their special distribution characteristic. The molecular distribution of the *n*-alkanes and *n*-alkenes in the cryoconites revealed that both allochthonous and autochthonous materials were essential contributors to the organic matters in the cryoconites on the TP glaciers.

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Figure captions:

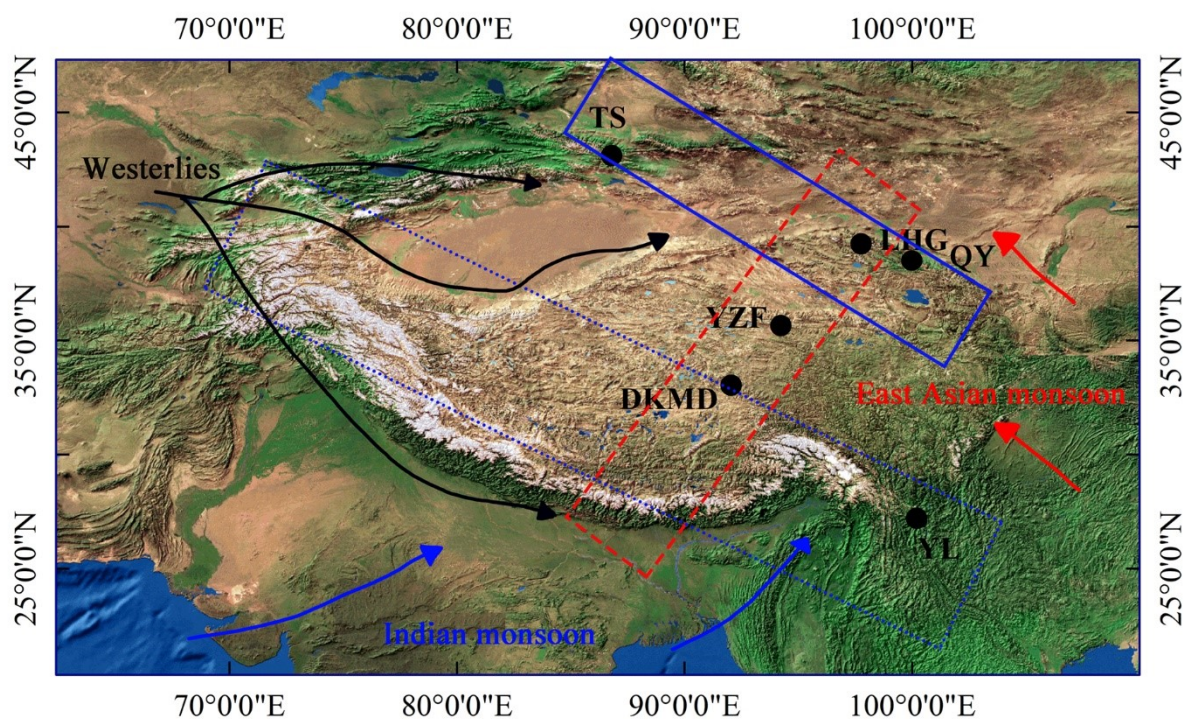


Fig.1. Sampling sites of cryoconites on the TP glaciers

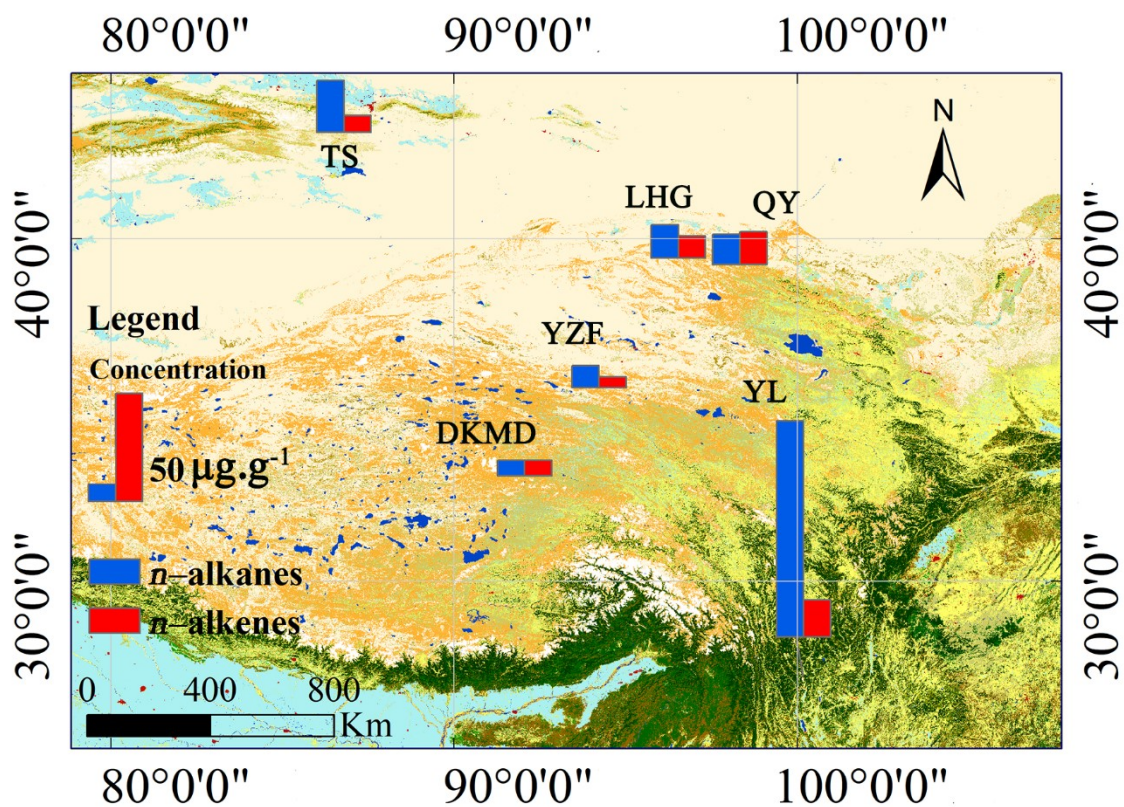


Fig. 2. The concentration of *n*-alkanes and *n*-alkenes in cryoconites

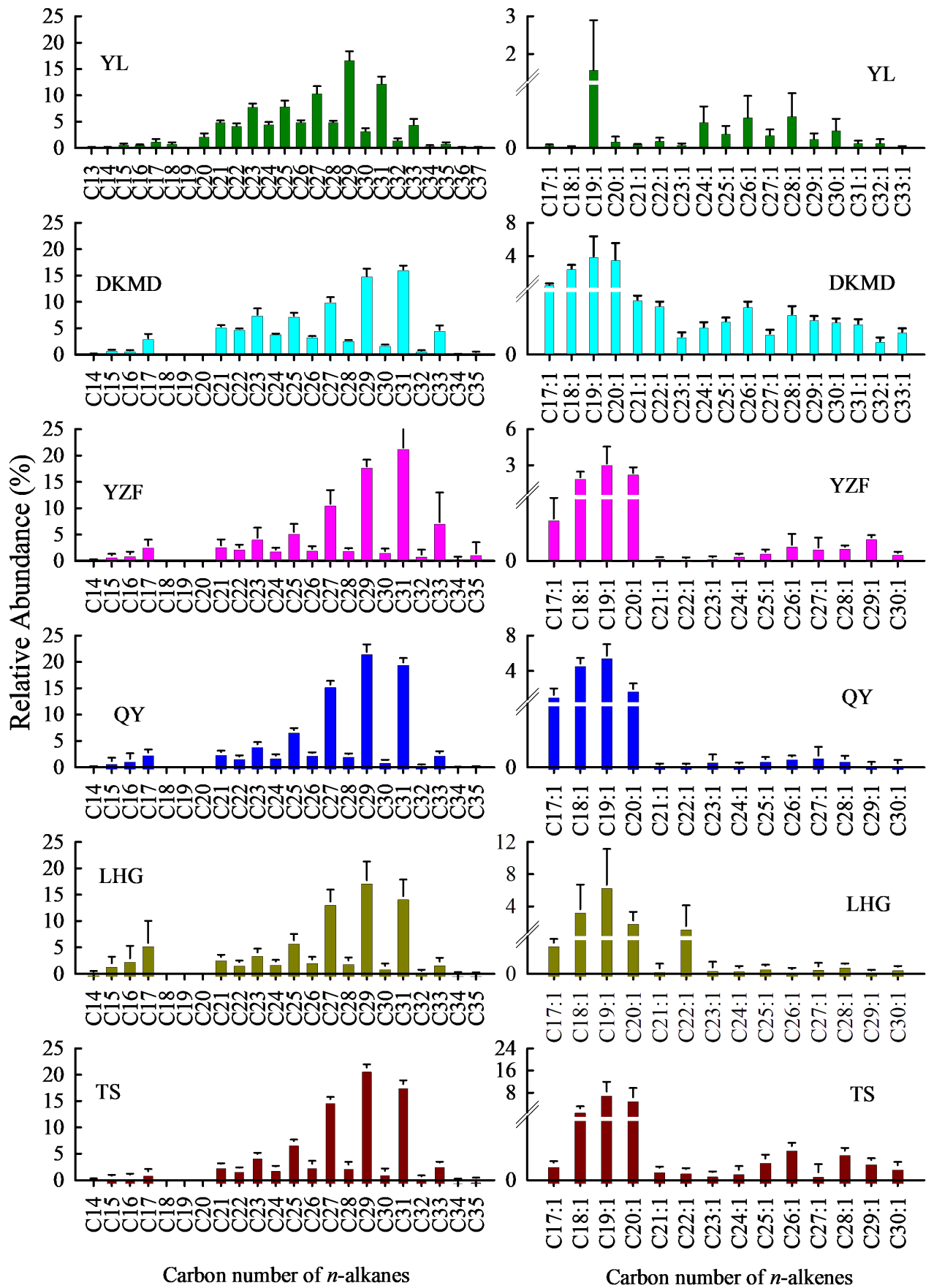


Fig.3. The average relative abundances of the *n*-alkanes and *n*-alkenes in the cryoconites

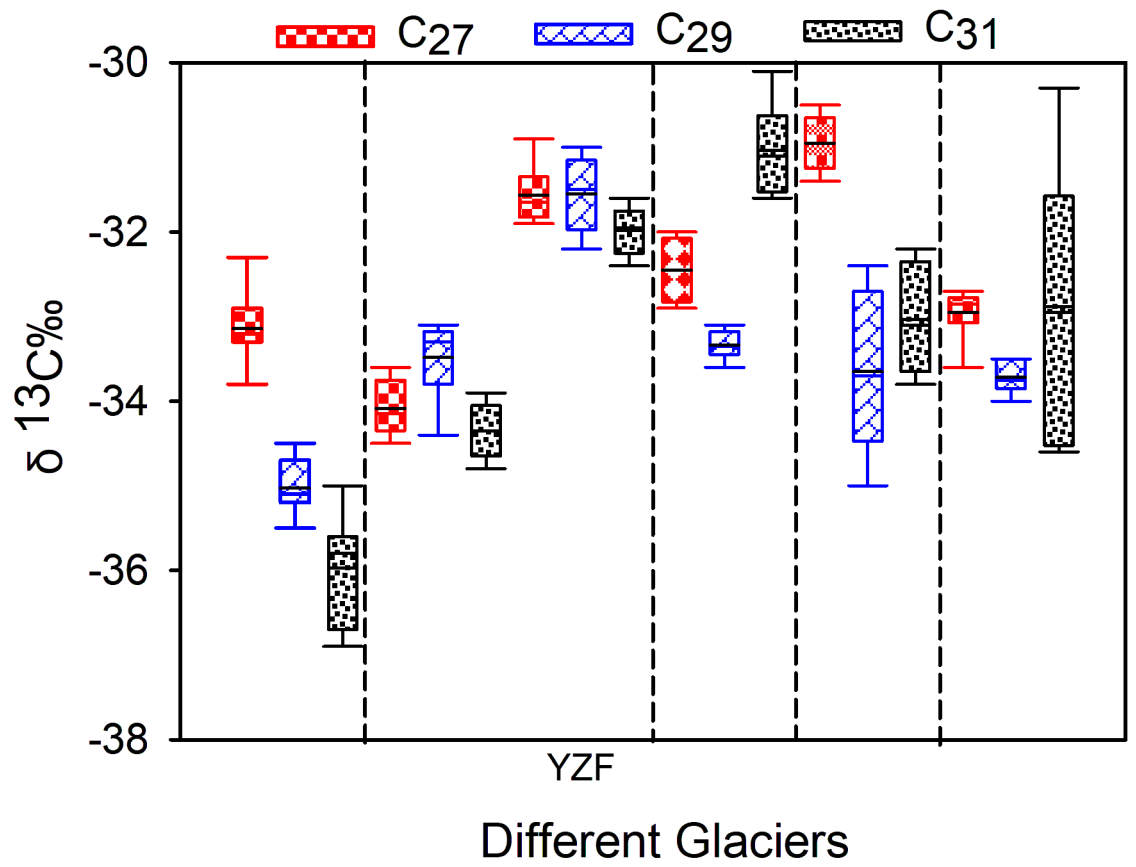


Fig.4. Box plots of the mean $\delta^{13}\text{C}$ values of the *n*-alkanes in all six glaciers

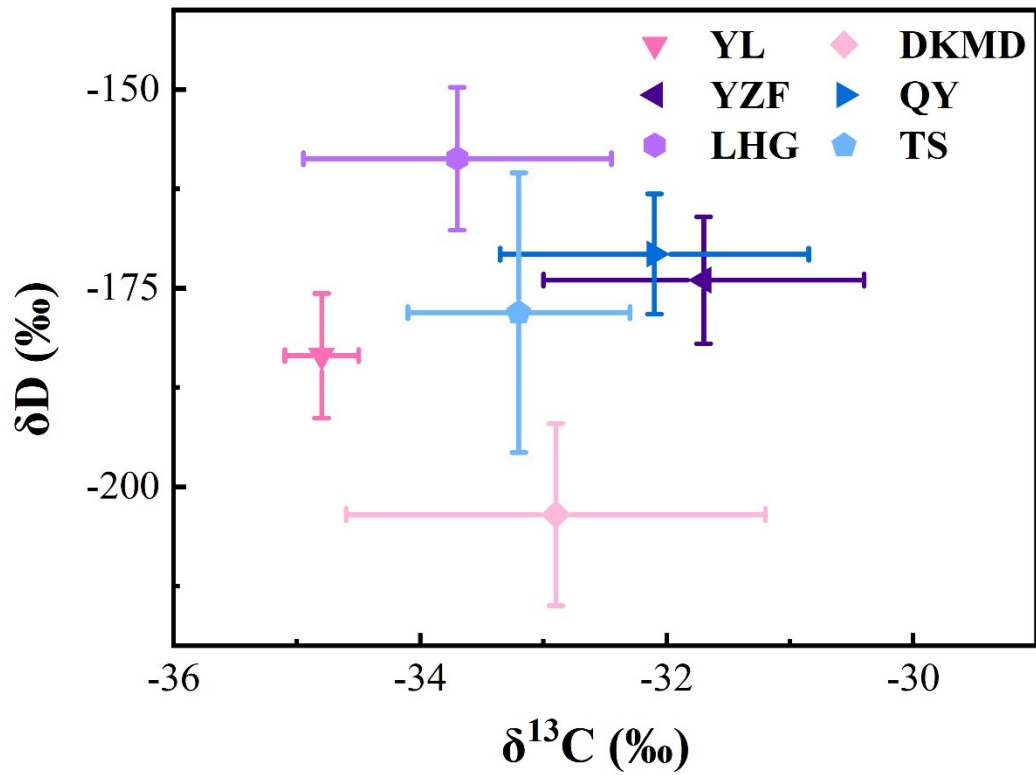


Fig. 5. Comparison of the average $\delta^{13}\text{C}$ and δD values of the *n*-alkanes in all six glaciers

Table 1. Information about the cryoconite samples collected from different glaciers of the TP

	Glacier	Latitude	Longitude	Altitude Range (m a.s.l.)	Sample Number
Southeastern Tibetan Plateau	YL Snow Mountains	27°40' N	100°11'E	4637 - 4825	9
Central Tibetan Plateau	DKMD	33°35' N	92°43' E	5464 - 5657	6
	YZF	35°39' N	94°14' E	5226-5454	3
Northern Tibetan Plateau	QY	39°14' N	97°46' E	4356-4766	11
	LHG	39°28' N	96°33' E	4386-4850	13
	TS	43°06' N	86°48' E	3772-4000	14

Notes: m.a.s.l. (meter above sea level)

Table 2. Molecular distribution of the *n*-alkanes and *n*-alkenes in the cryoconites

	Sample	YL	DKMD	YZF	QY	LHG	TS
	CPI ₂₅₋₃₃	3.3 ± 0.4	5.2 ± 0.1	7.5 ± 0.6	8.4 ± 0.7	6.5 ± 1.3	6.9 ± 1.3
<i>n</i> -alkanes	OEP	2.7	5.2	7.7	8.6	7.2	5.2
	ACL	28.8 ± 0.2	29.0 ± 0.2	29.1 ± 0.1	28.9 ± 0.1	28.7 ± 0.1	28.8 ± 0.1
	C ₂₇ /C ₃₁	0.82	0.50	0.66	0.71	0.82	0.89
	L/H ^a	0.13	0.11	0.11	0.09	0.16	0.09
<i>n</i> -alkenes	CPI _{17:1-22:1}	4.40	0.89	1.02	1.87	1.35	1.15
	OEP _{17:1-22:1}	6.32	0.76	0.82	1.03	0.98	0.98

CPI _{23:1-30:1}	0.39	0.69	0.96	1.15	0.56	0.66
OEP _{23:1-30:1}	0.37	0.72	0.89	1.12	0.91	0.65
L/H ^b	0.69	11.19	9.03	23.40	5.56	17.56

Table 3. δD and $\delta^{13}C$ values of individual *n*-alkanes in cryoconites

	YL	DKMD	YZF	QY	LHG	TS
<i>n</i> -Alkane ($\delta^{13}C$)						
$\delta^{13}C_{27}\text{‰}$	-34.2	-33.1	-31.7	-32.5	-31.9	-33.0
$\delta^{13}C_{29}\text{‰}$	-35.0	-33.5	-31.4	-33.3	-34.5	-33.7
$\delta^{13}C_{31}\text{‰}$	-36.0	-31.7	-32.0	-30.8	-34.3	-32.9
$\delta^{13}C_{\text{mean}}\text{‰}^a$	-34.8 ± 1.4	-32.9 ± 1.3	-31.7 ± 0.3	-32.2 ± 1.3	-33.7 ± 1.4	-33.2 ± 0.5
C ₃ %	92.0	79.0	71.3	74.3	84.5	81.5
C ₄ %	7.8	21.0	28.7	25.7	15.5	18.5
<i>n</i> -Alkane (δD)						
C ₂₇ %	-191.6	-202.2	-163.7	-162.9	-153.0	-171.8
C ₂₉ %	-181.0	-198.4	-172.5	-168.5	-159.2	-174.8
C ₃₁ %	-181.9	-209.3	-179.7	-178.0	-171.0	-207.0
$\delta D_{\text{mean}}\text{‰}^b$	-183.5 ± 7.2	-203.5 ± 9.1	-174 ± 6.1	-170.7 ± 6.3	-158.7 ± 6.8	-178.1 ± 13.6

a, abundance weighted mean $\delta^{13}C$ value ($\delta^{13}C_{\text{mean}}\text{‰}$) of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ *n*-alkane

b, abundance weighted mean δD values ($\delta D_{\text{mean}}\text{‰}$) of the *n*-C₂₇, *n*-C₂₉, and *n*-C₃₁ *n*-alkane