Holographic Materials
Outperforms Traditional Photopolymerizable Glass

Studies of organically modified silica glass have produced a material with high refractive index modulation, high optical quality, and photosensitivity. Plus, the material can be cast as a monolith or a film. In the March 12 issue of *Applied Physics Letters*, Pavel Cheben of the National Research Council of Canada and Maria L. Calvo of the Complutense University of Madrid have reported that by combining a titanocene photoinitiator and an acrylic monomer with a sol-gel precursor, a material is produced which is ideal for volume holographic recording.

Photoinitiation of ethylene glycol phenyl ether acrylate was accomplished with bis(μ3-2,4-cyclopentadien-1-yl)-bis[2,6-difluoro-3-(1H-pyrrol-1-yl)pheny]titanium. Both photoinitiator and monomer were added to the silica sol prior to the gelation point in concentrations of 0.4–0.8 wt% and 40 wt%, respectively. The solution was stirred for 20 min at 40°C until homogeneous, at which point it was cast into Teflon vials and allowed to gel and dry for 10 days at 60°C. Thin films of ~260 µm thickness were prepared by casting 0.1 mL of the solution onto a borosilicate glass plate and dried for 5 days at 60°C.

Characterization of the holographic performance of the glass was achieved by writing a grating pattern into the monolith or thin film, followed by reading the pattern with another beam. The glass refractive index was measured as \( n = 1.47 \) by a prism-coupling technique. The defraction efficiency dependence on the angle of incidence was measured by detuning the readout beam from the Bragg condition, yielding the grating angular selectivity. The refractive index modulation (Δn) amplitude and effective thickness of the grating were calculated using Kogelnik’s coupled-wave theory. The gratings in the monoliths produced \( \Delta n = 1.35 \times 10^{-3} \) for \( E \sim 1.5 \text{ J/cm}^2 \), and thus a sensitivity of \( S = 8.6 \times 10^{-4} \text{ cm}^2/\text{J} \) (\( E \) is exposure). A dynamic sensitivity \( S' = d(\Delta n) / dE \)

Sensitivity of \( \Delta n \) for exposure \( 0.1-0.35 \text{ J/cm}^2 \), the gratings in the 260-µm-thick films produced \( \Delta n = 4.5 \times 10^{-3} \) for \( E = 0.72 \text{ J/cm}^2 \) and thus a sensitivity of \( S = 6.2 \times 10^{-3} \text{ cm}^2/\text{J} \).

The researchers discussed the highly efficient formation of gratings in this material in terms of photopolymerization in the light regions. As the monomer is depleted in the light regions, mechanisms involving diffusion and capillary forces move monomers from the dark regions into the light regions where they participate in the photopolymerization process.

The hybrid organic-inorganic glass studied here exceeds the performance of traditional holographic materials. Furthermore, the ability to be cast as a monolith or as a thin film, the porosity of the final glass, the rigid silicon oxide backbone, and no requirement for solvent processing overcomes some problems associated with traditional holographic materials. These combined advantages demonstrate excellent prospects for practical holographic storage devices using this material.

Maria L. Calvo of the Complutense University of Madrid and Pavel Cheben of the National Research Council of Canada are developing a freeze-dry process starting from a water-based slurry. They maintain that their method is more environmentally friendly since it eliminates the necessity for the use of a binder. Furthermore, the resultant material has good mechanical strength and the pore distribution is easily controlled. In the January issue of the *Journal of the American Ceramics Society*, these researchers explain how they obtained such a porous structure. They started by mixing alumina powders and a small amount of a dispersant into slurry with distilled water; ball milling occurred for volume holographic recording.

Alumina With 50% Porosity Obtained After Freeze-Drying a Water-Based Slurry

Ceramic materials with a large surface area are commonly used as catalysts, filters, chemical sensors, and for other similar applications. Freeze-drying is a versatile fabrication method for this type of material and several variants of its use are found in the literature. A group of researchers from the Synergy Ceramics Laboratory and the National Industrial Research Institute of Nagoya in Japan are developing a freeze-dry process starting from a water-based slurry. They maintain that their material is more environmentally friendly since it eliminates the necessity for the use of a binder. Furthermore, the resultant material has good mechanical strength and the pore distribution is easily controlled. In the January issue of the *Journal of the American Ceramics Society*, these researchers explain how they obtained such a porous structure. They started by mixing alumina powders and a small amount of a dispersant into slurry with distilled water; ball milling occurred for volume holographic recording.

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for about 20 h. Afterwards, stirring the slurry in a vacuum desiccator helped to eliminate any trapped air. Then the slurry passed to a cylindrical container specially designed for this experiment. This container had one end immersed in ethanol cooled at −50°C. They fabricated this section of the container with a metallic material of high conductivity. For the sides of the container, they used a resin. The other end of the container remained open to the air allowing growth of frozen ice in a vertical direction. Drying of the ice thus formed took one day in conditions of reduced pressure. The sintering conditions of each sample were 1400°C and 1550°C, both for 2 h.

Two slurries were prepared at different concentrations. Measurements of their rheological properties revealed a uniform distribution of the alumina powders. The freeze-dried samples showed large pores aligned with the growth direction. After freeze-drying, the volume change measured was almost negligible. Pore-size distributions measured after sintering have two peaks at 0.1 µm and 10 µm for the samples sintered at 1400°C. On the other hand, the pore-size distributions of the samples sintered at 1550°C only have one peak at around 10 µm. Inspection of these samples using a scanning electron microscope allowed the detection of a dendritic structure inside the large pores for all samples. In the case of the sample sintered at 1400°C, inside the dendritic structures there were small pores. These pores correspond to the peak at 0.1 µm observed in the pore-size distribution. It was also possible to detect a decrease in the pore size with increasing concentration of the slurry. Furthermore, the researchers’ current results indicate that by changing the freezing temperature, control of the macroscopic pore size is possible. Since the pores appear before sintering, this method is applicable to any ceramic. Recently, the researchers obtained porous silicon nitride after application of this technique.

Electrostatic Attraction Enables Precise Control in Thin-Film Assembly

A research group at the University of Massachusetts—Lowell has prepared a thin film consisting of oppositely charged alternating layers of organic laser dye molecules intercalated into an aluminosilicate lattice and polycation spacer layers. The components of the film were sequentially deposited onto a glass substrate and allowed to assemble by electrostatic attraction. This method allowed for precise control of film thickness and dye molecule orientation in the film. The dye molecules in the assembly were more thermally stable than free dye and exhibited a high degree of order.

As reported in the March issue of *Chemistry of Materials*, the dye molecules (coumarin 1 or 7-diethylamino-4-methyl coumarin) were first protonated by hydrochloric acid treatment. A suspension of the aluminosilicate hectorite was then added to the protonated dye solution and ion exchange was allowed to proceed for 24 h. The resulting host/guest composite was separated by centrifugation, washed with water to remove excess coumarin, and suspended in de-ionized water. Composites with coumarin content ranging between 11% and 25% of the cation exchange capacity (CEC) of the hectorite were prepared by adjusting the dye/aluminosilicate ratio. The thin-film assembly of the composite was prepared by alternately depositing layers of positively charged poly(diallyldimethylammonium) chloride and negatively charged composite onto hydrophilized glass slides.

The absorbance and emission intensities of the coumarin were monitored during film assembly and were found to increase linearly with the number of deposition cycles. This is evidence of a controlled buildup of layers, a feature not present in existing aluminosilicate thin-film preparation methods.

Thermogravimetric analysis of the thin films showed that the loss of coumarin began at 340°C, 190° higher than that observed for the free dye. The researchers attributed this improved thermal stability to the restricted mobility of the dye molecules in the hectorite lattice. The orientation of the dye molecules was inferred from the changes in the interlamellar spacing of the hectorite as measured by powder x-ray diffraction. These measurements showed that the interlamellar spacings did not change for composites having coumarin content of up to 100% CEC. The spacing was found to increase with increasing coumarin content above this value. The researchers propose that at low CEC the guest molecules are oriented flat with respect to the host lattice, while at a higher percentage of CEC they tilt for more efficient packing, the angle of the tilt increasing with increased doping level. This allows for precise control of chromophore orientation in the assembly.

The researchers believe that the ability to control the film thickness and chromophore orientation coupled with the stability and high order of the resultant structure make the preparative method described here “a simple but powerful strategy for preparing a nanocomposite film for photonic applications.” They add that the method “may be adaptable to other photofunctional chromophores and fluorophores.”

Gregory Khitrov

Raman Spectroscopy Detects Carotenoid Levels in Human Retina

A team of researchers at the University of Utah has recently developed a resonant Raman scattering method for the detection of macular carotenoid pigments. In their article in the February 15 issue of *Optics Letters*, they demonstrate that this approach can be used as a noninvasive technique for the determination of carotenoid levels in the living human retina. Quantitative information could be obtained with laser excitations that were well below the safety limits for macular exposure.

“Age-related macular degeneration is the leading cause of blindness in the elderly in the United States,” said Paul S. Bernstein, a professor of ophthalmology at Utah. “The vision loss can be prevented if low levels of carotenoid pigments in the retina are detected at an early stage.”

The most commonly used noninvasive technique for quantification of carotenoid pigment levels is a subjective psychophysical flicker photometric test that involves color matching of two light beams aimed at the fovea and the perifoveal area. This method is not only time intensive, but also requires the attention of the subject and good visual acuity. Bernstein said that “this limits the usefulness for assessing macular pigment levels in the elderly population that is most at risk.” In collaboration with colleagues in the Department of Physics, he developed a fast and objective method for pigment level measurements based on resonant Raman scattering.

The two carotenoid pigments that are concentrated in the human retina, lutein and zeaxanthin, are polyene-like molecules with a conjugated \(\pi\)-bonding system along the carbon backbone.

“The molecules exhibit extremely low luminescence efficiency and no fluorescence due to nonradiative decay from the excited state to a second excited state, from which electronic emission to the ground state is symmetry forbidden,” said Werner Gellermann, a professor in the Physics Department. “This allows us to explore the resonant Raman scattering response without having to worry about potentially masking fluorescence signals from the carotenoids.”

Preliminary experiments were conducted in a 180° backscattering geometry using an optically correct model eye to which