TEM Characterization on Oxygen-Deficient Titania Supported Pt Electrocatalysts for Energy Conversion

Vahid Rastegar, Justin Roller, M. Josefina Arellano-Jimenez, Matthew T. Janish, Rishabh Jain, Radenka Maric and C. Barry Carter

Department of Chemical, Materials and Biomolecular Engineering, University of Connecticut, 191 Auditorium Road Unit 3222, Storrs, CT 06269-3222

Platinum-based catalysts supported on carbonaceous materials (Pt/C) have become the standard catalyst-support choice in Proton exchange membrane fuel cells (PEMFC) [1]. Corrosion of the carbon support is one degradation mechanism leading to poor durability and unacceptable lifetimes hindering further market penetration [2-4]. Discovery of a support that offers comparable electrical conductivity, surface area, structure, and Pt activity enhancement that can be manufactured at a competitive price to carbon and possesses a greatly increased corrosion resistance would represent a breakthrough in attempts to commercialize PEMFCs. Several reviews of potential support materials have recently been published [5,6] on alternative materials to carbon. Of particular interest are the sub-stoichiometric titanium oxides of the general formula Ti_nO_{2n-1} (Magnéli phases), where n is a number between 4 and 10 [7]. These Magnéli phases are characterized by extended planar defects and crystallographic shear planes [8] which vary according to the oxygen deficiency.

The aim of this project is basis research on Pt nano-particles on ball-milled Ti₄O₇ particles. High-quality characterization of these particles is required to investigate their composition and their structure. In order to evaluate the chemical composition, scanning transmission electron microscopy (STEM) and high-angle annular dark field were performed on the specimens to get information on microstructure, atomic structure and atomic number. X-ray Energy dispersive spectroscopy (XEDS) permits subnanometer elemental identification and compositional analysis. The addition of electron energy loss spectroscopy (EELS) and energy-filtered imaging allows for the detection of elements at higher spatial resolution, phase identification and bonding information. For structure analysis, the Z-contrast image was particularly suited to imaging catalyst particles with high Z that are often used to grow nanotubes. In the Z-contrast image, the support was seen more clearly in the phase contrast image. Due to the large size of the Ti_nO_{2n-1} particles, even after ball-milling, imaging in electron transmission modes is challenging. Fig. 1a is illustrative of the as-received Ti_nO_{2n-1} particles (i.e., prior to Pt deposition); Fig. 1b shows the resulting size after 100 hours of ball-milling at 400 rpm. It is also worth noting that the zcontrast between Pt (z = 78) and Ti (z = 22) is slightly smaller than Pt and C (z = 6). Fig. 2 shows a bright field view of the top surface of a Ti_nO_{2n-1} particle that may have been cleaved from a larger particle. Fig. 2a shows that the support is again evenly covered in Pt particles with an average size of ~ 2.3 nm. Deposition appears to be concentrated along the left side edge of the Ti_nO_{2n-1} particle which measures 150 nm in width and over 165 nm in height. Fig. 2b shows another Ti_nO_{2n-1} particle that is approximately 30 nm in diameter and tilted to display the Pt attached to the surface. The attached Pt particles range from 2-3.8 nm in size and appear to be coated in a film that is approximately 5Å in thickness [8]. Such a coating would have potential implications for the reactivity of these particles [9].

References:

[1] F. Coloma, et al, J. Catal. **154** (1995) 299-305.

- [2] L.M. Roen, et al, Solid-State Lett. 7 (2004) A19-A22.
- [3] T.R. Ralph, et al, ECS Transactions 1 (2006) 67-84.
- [4] B. Merzougui, S. Swathirajan, J. Electrochem. Soc. 153 (2006) A2220-A2226.
- [5] K. Malek, et al, J. Phys. Chem. C 111 (2007) 13627-13634.
- [6] R. Maric, et al, ECS Trans. 12 (2008) 59-63.
- [7] B.S. Mun, et al, Phys. Rev. B 71 (2005) 115420-115426.
- [8] C.B. Carter and M.G. Norton, Ceramic Materials: Science and Engineering, Springer 2007.
- [9] VR would like to acknowledge GAANN Fellowship from the NSF Department of Education. This work was supported by the University of Connecticut.

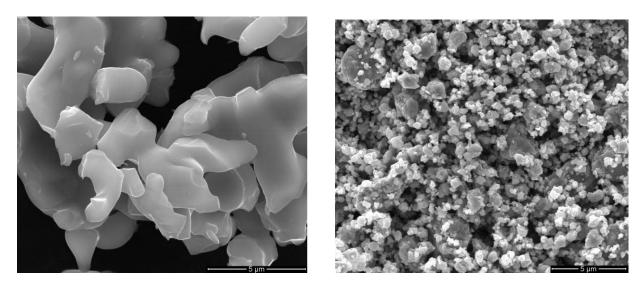


Figure 1. SEM images of a) pre and b) post ball milled Ti_nO_{2n-1} .

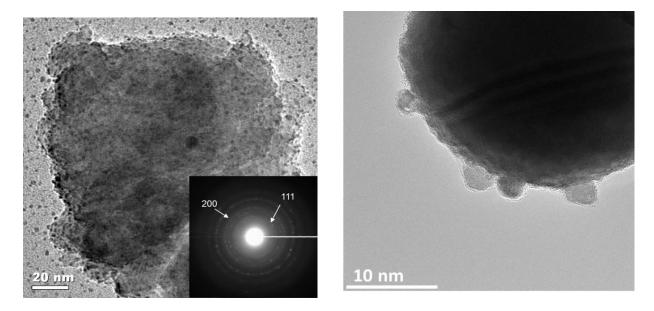


Figure 2. TEM images of Pt deposited Ti_nO_{2n-1} a) cleaved Ti_nO_{2n-1} particle and b) an area of another support particle revealing the attached Pt.