A PRETREATMENT PROCEDURE FOR THE AMS RADIOCARBON DATING OF SUB-FOSSIL INSECT REMAINS

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ABSTRACT. Two pretreatment methods for accelerator mass spectrometry (AMS) dating of insect remains were explored. One method involves a simple acid wash that removes carbonate, while the other is based on the industrial purification of chitin and results in isolation of polymeric chitosan. No contamination is observed from Maillard reactions during the deacetylation reaction used to isolate the chitosan. The methods were tested on Coleoptera samples from two Roman Britain sites. Our results demonstrate that both methods produce acceptable AMS dates that correspond well to the expected age of the deposits from which they came.

INTRODUCTION

Remains of insects are often found in an archaeological context and because many of them are temperature-sensitive organisms, they have been used as temperature and climate proxies in environmental studies (Robinson 2001). Accelerator mass spectrometry (AMS) radiocarbon dating is the most common method utilized to provide the chronological framework for these studies; however, it is typically associated with organic remains from peat or swamp material that are used for dating, not the insects themselves. This is often due to difficulty in obtaining pure material for dating.

For single-compound ¹⁴C dating, the best candidate from insects is chitin or a derivative. Chitin (Figure 1) is a polymer composed of repeating *N*-acetyl-D-glucosamine units that is a major structural component of insect exoskeletons. In insects, chitin is bundled into microfibrils that are peripherally bound to proteins and arranged in a carbonate matrix to form the exoskeleton (Neville 1975). After death, the chitin-protein complex can undergo depolymerization or various condensation reactions with substances from the soil, resulting in numerous compounds, including humics, that may or may not be indigenous to or of the same age as the insects. While chitin is more resistant to diagenetic degradation when it is complexed to proteins, there is often significant chitin degradation even in cuticles that appear well-preserved (Stankiewicz et al. 1998). However, preservation of insect chitin under certain conditions in specimens as old as 25 million yr has been reported (Stankiewicz et al. 1997).



Figure 1 Structure of chitin

Certain limitations and requirements must be considered when developing a ¹⁴C pretreatment procedure. Consistent, high recovery of pure material, with some way to assess the purity, is necessary. The procedure should not be too time-intensive and should use current protocols and techniques

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where possible. In addition, isotope fractionation must be minimized. Previous work in our laboratory towards a pretreatment procedure for insects involved isolation of D-glucosamine from hydrolyzed chitin using ion-exchange chromatography, but this method identified serious problems of low recovery and incomplete purification (Hodgins et al. 2001). In addition, a significant and relatively consistent offset between AMS dates of insect remains and surrounding organic material was observed, and has also been reported by others (Elias and Toolin 1990; Elias et al. 1991; Walker et al. 2001). Possible explanations for this offset are different ecological and feeding adaptation of the insects, perhaps resulting in the uptake of differently aged ¹⁴C, or insufficient purification of the isolated chitin.

In this paper, we report a study comparing two purification methods. The first is a simple acid treatment to remove carbonate (Method A), while the other involves a harsher reaction to remove the protein component and isolate only the chitin polymer (Method B). This second method is based on the industrial purification of chitin (Roberts 1992) and involves the deacetylation, with concurrent deproteinization, of insect exoskeletons, followed by isolation of chitosan (deacetylated chitin) by a dissolution-precipitation procedure. The procedures were tested on Colepotera remains from two Roman-period sites in Britain.

EXPERIMENTAL SECTION

Materials and Methods

Commercial chitin, isolated from crab shells, was obtained from Aldrich. Water was purified using a Millipore Milli-Q system. All other solvents and reagents were purchased from Fisher Scientific. Coloeoptera remains were obtained by Dr Mark Robinson of the Natural History Museum at Oxford from the sites of Priors Gate and Godmanchester in Cambridgeshire, UK. Elemental and mass spectrometric analyses were undertaken using a Europa ANCA Roboprep CHN analyzer interfaced to a Europa 20/20 MS operating in continuous-flow mode. Graphite was prepared by reduction of CO₂ over an iron catalyst in an excess H₂ atmosphere at 560 °C prior to AMS ¹⁴C measurement (Bronk Ramsey and Hedges 1999; Bronk Ramsey et al. 2000). Samples of insect chitin <1.6 mg C in size were AMS dated as directly-injected CO₂ using the ORAU gas ion source. δ^{13} C values in this paper are reported with reference to VPDB and δ^{15} N results are reported with reference to AIR (Coplen 1994).

Deacetylation of Commercial Chitin

Chitin (1.6 g) was suspended in 50 mL of 50% NaOH and heated to 120 °C for 1 hr. The reaction was filtered and the solid resuspended in purified water. Next, the 6M HCl was added to make the solution weakly acidic (about pH 3) so that the solid dissolved. The solution was again filtered to remove any residual solids, and 6 M HCl added to the filtrate to make the solution strongly acidic (pH < 1). A white solid (chitosan•HCl) precipitated from the solution. This was filtered, rinsed, dried, and analyzed.

Maillard Reactions with Chitin and D-Glucosamine

One-hundred g of chitin or D-glucosamine was mixed with 100 mg of glycine (when used) and 10 mL of either water or 50% NaOH. The reactions were heated at 120 °C for 1 hr. The neutral reactions were washed several times with water. Those with NaOH were filtered and acidified with HCl as described above. All samples were freeze-dried prior to analysis.

Preparation of Coleoptera Remains for ¹⁴C Dating (Method A)

The insect remains (elytra, pronotum, head capsules, and legs) were picked out of suspension under a microscope, and rinsed with acetone, methylene chloride, and acetone again, and dried under vacuum for 5 hr. The insect pieces were submerged in 0.5 M HCl for 3 days. They were then filtered, rinsed, and lyophilized.

Preparation of Coleoptera Remains for ¹⁴C Dating (Method B)

The insect remains were treated as in Method A, but following removal of the acid, they were heated in 5 mL 50% NaOH for 30 min. The resulting product was filtered and the solids resuspended in water. The resulting solution was made weakly acidic by addition of 6 M HCl, filtered to remove residual solids, and then made strongly acidic by further addition of 6 M HCl. The resulting solids were captured in pre-combusted glass-fiber filters, lyophilized, and analyzed.

Preparation of Seeds for ¹⁴C Dating

Carbonized and waterlogged woody seeds were prepared using the acid-base-acid (A-B-A) method (Hedges et al. 1989) to remove carbonates, fulvics, and humics, then rinsed and dried.

RESULTS AND DISCUSSION

Deacetylation of Chitin

In order to test the parameters for the deacetylation reaction in Method B, experiments were first performed on commercial chitin. Deacetylation (Scheme 1) results in a loss of two carbons from each residue, and thus, the C/N ratio can be used to estimate the degree of deacetylation. A completely acetylated chitin chain will have a C/N of 8, while completely deacetylated chitin (chitosan) will have a C/N of 6. Using the C/N ratio instead of the absolute values for % N and % C eliminates the effects of residual moisture in this determination (Roberts 1992). A C/N less than 6 suggests the presence of proteins, which usually have C/N between 2 and 6. Proteins present in insect samples may be either indigenous or contaminating, and thus, their removal is important.



Scheme 1 Deacetylation of chitin

After deacetylation, the resulting chitosan polymer is soluble in weak acid, but will precipitate out of solution when strong acid is added due to the presence of multiple charges along the length of the polymer. It should be noted that the carbohydrate polymers are stable to even strong base, while similar conditions will result in cleavage of the amide backbone in proteins.

One necessary consideration for the deprotection reaction is the Maillard reaction, a condensation reaction that occurs between carbohydrates and amino acids that is responsible for a number of effects including the browning of cooked meat (Fayle and Gerrard 2002). The occurrence of Maillard reactions during the deprotection reaction may lead to covalent attachment of non-native and differently-aged proteins or amino acids onto the chitin. The mechanism of the initial steps of the reaction

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is shown in Scheme 2 using D-glucosamine as the substrate. The first step is a nucleophilic attack of the amine group of an amino acid on the anomeric carbon of the glycoside. This step can only occur on the linear form of the glycoside (2), and since polymerization requires the hemiacetal form (1), it suggests that if the chain integrity is preserved, no Maillard reaction will be observed. The initial product condenses to a Schiff base (4) and rearranges to form the Amadori product (6), which can then undergo numerous other reactive steps to produce a variety of highly colored products.



Scheme 2 First steps of Maillard reaction mechanism

We ran a series of experiments in which D-glucosamine and chitin were reacted under conditions I–IV listed below.

I. water II. 50% NaOH III. water + glycine IV. 50% NaOH + glycine.

The products of the Maillard reaction are highly colored, so this color change, as well as C/N ratios of the resulting polymeric products can be used to determine if the reactions occurred or not. Elemental analysis was not undertaken on the D-glucosamine solutions because a range of compounds was formed, thwarting our attempts to isolate specific species for analysis.

As expected, all solutions containing D-glucosamine turned dark brown upon heating. The presence of base appeared to accelerate the reaction, as solutions II and IV changed color faster and became darker brown than reactions I and III. Even solutions that did not contain glycine (I and II) darkened significantly, probably due to a related reaction involving nucleophilic attack of the amino group of one D-glucosamine on another molecule using the same mechanism as shown in Scheme 2.

In contrast, the solutions containing chitin did not change color and showed little evidence of Maillard products. Results are shown in Table 1. C/N ratios and stable isotope measurements show very little contamination of polymeric chitin from glycine which has significantly different values. This suggests that Maillard reactions of the chitin are unlikely to occur during the deacetylation reaction even with the presence of proteins and amino acids. In addition, the chitin chain does not appear to depolymerize under the reaction conditions.

Conditions ^a	C/N (σ) ^b	$\delta^{13}C$, ‰ (σ)	δ^{15} N, ‰ (σ)
Ι	8.41 (0.11)	-23.01 (0.19)	-1.96 (0.17)
II	7.80 (0.30)	-22.52 (0.28)	-2.34(0.14)
III	8.28 (0.05)	-23.04(0.14)	-1.91 (0.31)
IV	7.23 (0.32)	-22.52(0.42)	-2.17 (0.39)
glycinec	2.04 (0.01)	-37.11 (0.01)	13.33 (0.03)

Table 1 Elemental and isotopic results from Maillard reactions with chitin.

^aAll reactions were heated at 120 °C for 1 hr. Conditions: I-water; II-50% NaOH; III-water + gly; IV-50% NaOH + gly.

^bn=6 for reactions I – IV, n=3 for glycine.

^cData are given for the glycine used in reactions III and IV.

¹⁴C Dating of Coleoptera

Coleoptera remains were obtained from wells at two Roman-era settlement sites in Cambridgeshire, England. The Priors Gate settlement is at the site of present-day Eaton Socon. A full analysis of the insect remains suggested that the area around the settlement was an open grassland or meadow, with trees growing in hedgerows and timber buildings (M Robinson, personal communication 2002). Pottery found in context with the Coleoptera was stylistically dated to the late 1st to 3rd century AD (Mepham and Loader 2001). The specimens were divided into 2 aliquots and treated with Methods A and B as described above.

Recovery, stable isotope, and AMS results are shown in Table 2. Clearly, Method B results in the loss of about 85% of the mass, but the increase in C/N indicates that much of the lost material may be proteinaceous. The identical δ^{13} C values indicate no carbon fractionation using the procedure. The uncalibrated AMS determinations for both procedures are indistinguishable within error and correlate well with the date of the pottery found in context with the Coleoptera. Calibrated age ranges (AD/BC) are shown in Figure 2 and these also demonstrate significant overlap.

Table 2 Elemental, isotopic, and AMS data for Coleoptera remains from Priors Gate. OxA-X-*nnnnnn* numbers are given to the insect determinations because of their non-routine/experimental pretreatment chemistry.

OxA	Method ^a	Yield, %	C/N	δ ¹³ C, ‰	δ ¹⁵ N, ‰	¹⁴ C age (yr BP)
OxA-X-2020-21	А	100	5.6	-25.9	9.7	1851 ± 23
OxA-X-2039-11	В	15	7.2	-25.6	4.5	1830 ± 50

^aA-acid wash, B-full isolation of polymeric chitin (see text for details).

Further AMS determinations were obtained on Coleoptera samples from the London Road excavation of the Roman settlement at Godmanchester. Godmanchester was an active Roman settlement from the late 1st century to the 4th century AD (Jones 2003), although later activity seems to have been confined to inhumation cemeteries. It was a small settlement but had certain community buildings, including a forum and temples, suggesting that it may have functioned as a seat of local government. Evidence for farming wheat and for specialization of labor within the town (including animal husbandry and pottery production) was also found.



Figure 2 Calibrated AMS dates obtained from Coleoptera from Priors Gate, Eaton Socon, Cambridgeshire, treated by Methods A and B.

AMS determinations of Coleoptera (*Geotrupes* and *Aphodius*) treated with Method A were compared with carbonized (*Triticum spelta*) and waterlogged (*Malva sylvestris*, *Onopordum acanthium*, *Fallopia convolvulus*) seeds found with them. Results are shown in Table 3. The three AMS determinations are statistically indistinguishable as a group [error weighted mean = 1716 ± 19 BP (*T*'=5.55; $\chi^2_{2:0.05}$ =5.99)]. Calibrated date ranges are shown in Figure 3, and correspond to the settlement era.

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OxA number	Sample	C/N	δ ¹³ C, ‰	δ^{15} N, ‰	¹⁴ C age (yr BP)
OxA-X-2040-22	Coleoptera	5.6	-25.9	8.4	1764 ± 29
12418	Waterlogged seeds	29.9	-26.5	7.7	1690 ± 32
12417	Carbonized seeds	18.0	-23.2	6.5	1650 ± 45

Table 3 Elemental, isotope, and AMS data for Coleoptera remains and seeds from the Godmanchester site.

Atmospheric data from Stuiver et al. (1998); OxCal	l v3.6 Bronk Ramsey	(2000); cub r:4 sd:12 p	prob usp[chron]			
OxA-X-2040-22 1764+29BF						
OxA-12417 1650±45BP	_ _					
OxA-12418 1696±32BP						
200CalBC CalBC/CalAD 2	200CalAD	400CalAD	600CalAD	800CalAI		

Calibrated date

Figure 3 Calibration plots of AMS dates on beetles (OxA-X-2040-22) and carbonized (OxA-12417) and waterlogged (OxA-12418) seeds from Godmanchester, Cambridgeshire.

CONCLUSION

Our results suggest that a simple pretreatment method involving washing with organic solvents and treating the samples with acid is sufficient to yield accurate AMS dates in line with archaeological expectation. This method is simple and utilizes existing laboratory protocols. Future work will involve verification of this procedure using samples from other locations and of other ages.

Isolation of polymeric chitin is also a suitable technique, and appears to lead to ¹⁴C determinations of the same age as insects purified using the simple acid wash. While this method does not result in pure isolated chitin, as seen by the C/N ratios, it is a quick and simple way of purifying insect remains for AMS dating. The substance obtained is partially deacetylated chitin with little associated protein. Experiments suggest that the chain integrity is preserved during the reactions, and no external protein contamination via a Maillard mechanism was observed. The yield of this procedure will need to be improved before it can be used routinely, but it holds promise as a purification method for the most contaminated samples.

ACKNOWLEDGEMENTS

Funding from the National Science Foundation, USA (INT-0202648) is gratefully acknowledged. We also thank Dr Mark Robinson for providing samples for dating and for helpful discussion.

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