

## Characterization of Cu-30Mo Alloys Synthesized by Mechanical Alloying

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There has been a growing interest in the synthesis of materials with nanoscale microstructures. Copper and its alloys have good electrical and thermal conductivity and good resistance to corrosion [1]. The final properties of copper and its alloys greatly depend on the amount of the alloying elements and even on small additions of impurities; hence the method of producing the copper alloys is very important and decisive for the final application.

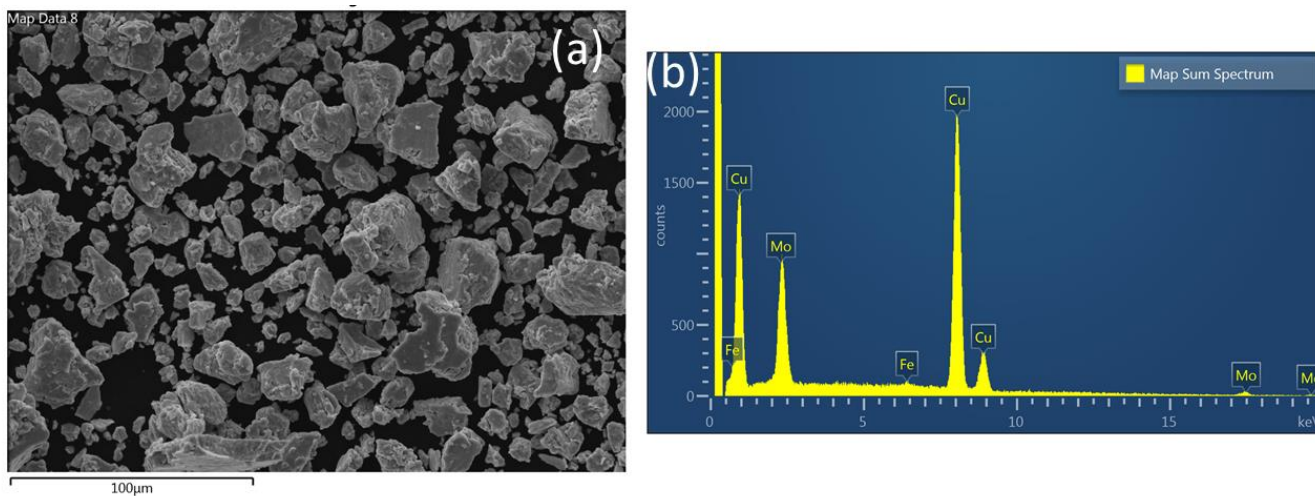
The common synthesis techniques are: chemical vapor deposition, chemical precipitation, sol-gel processing, gas-phase condensation, physical evaporation, sputtering and mechanical alloying [2,3]. In conventional casting, alloy grain size is difficult to control. Coarse grains deteriorate the mechanical properties of the alloys. It has been reported that mechanical alloying (MA) and powder metallurgical techniques can be used to prepare alloys Cu-based. MA can reduce oxidation because a new phase is formed, and produces a pre-alloyed powder that shortens the sintering time.

In this work we report on the structural and chemical characterization of Cu-30Mo alloy synthesized by mechanical alloying, with milling time of 10 h. Scanning electron microscopy (SEM) and energy dispersive spectroscopy (EDS) were used to identify the morphology formed after the mechanical alloying process and chemical composition of the particles formed respectively.

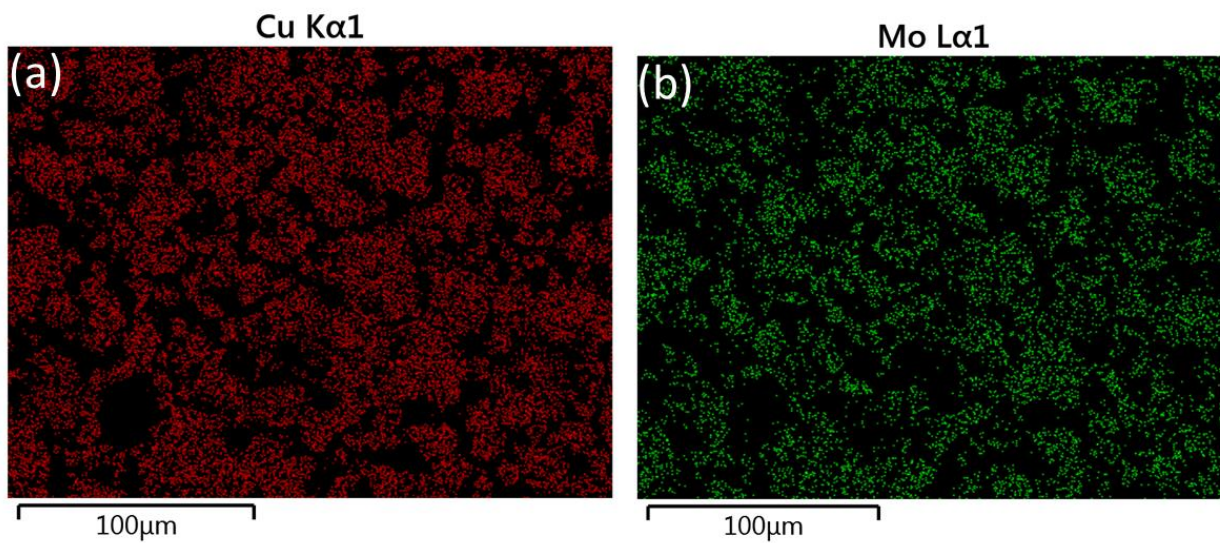
Figure 1a shows particles milled for 10 hours. The particles are rounded and the average size is between 5-25  $\mu\text{m}$ . Agglomeration of the particles can also be seen in the same figure, indicating welding during the milling stage. The average size of the particles depends on the extent of welding and fragmentation. The chemical composition of the particles is observed in the Figure 1b where we can observe the Copper and Molybdenum peaks, we can also appreciate contamination of the Iron which comes from the steel balls used during the milling. Results of the quantification of the elements showed that the particles are composed of 76.04 % wt, 23.57 % wt and 0.39 % wt of Cu, Mo and Fe respectively. Figure 2a and 2b shows the mapping of the particles shown in figure 1a, indicating that the milling time used formed particles of copper with embedded molybdenum.

### References

- [1] O. Alvarez-Fregoso, S. López, J. A. Juárez-Islas, M. García, E. Martínez, M. A. Alvarez-Pérez, J. Ch. Ramírez, S. Granados. *Phys. Stat. Sol. B*. **Volume 220**, (2000), p. 575
- [2] R. Mendoza, J. Huante, V. Camacho, O. Alvarez-Fregoso, and J. A. Juárez-Islas, *J. Mater. Engng. and Perf.* **Volume 8**, (1999), p. 1
- [3] Maria do Carmo, Amorim da Silva, Severino Jackson Guedes de Lima, *Mat. Res.* **Volume 8**. (2005), p. 169



**Figure 1.** (a) SEM image of Cu-30Mo milled for 10 h, (b) EDS analysis Co-30Mo milled for 10 h



**Figure 2.** (a) and (b) Mapping analysis of the Cu and Mo respectively of Cu-30Mo milled for 10 h