FRESHWATER RESERVOIR EFFECT VARIABILITY IN NORTHERN GERMANY

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ABSTRACT. The freshwater reservoir effect is a potential problem when radiocarbon dating fish bones, shells, human bones, or food crusts on pottery from sites near rivers or lakes. The reservoir age in hardwater rivers can be up to several thousand years and may be highly variable. Accurate 14C dating of freshwater-based samples requires knowing the order of magnitude of the reservoir effect and its degree of variability. Measurements on modern riverine materials may not give a single reservoir age correction that can be applied to archaeological samples, but they show the order of magnitude and variability that can also be expected for the past. This knowledge will be applied to the dating of food crusts on pottery from the Mesolithic sites Kayhude at the Alster River and Schlamersdorf at the Trave River, both in Schleswig-Holstein, northern Germany.

INTRODUCTION

Reservoir effects can lead to erroneously large radiocarbon ages if the sample’s carbon originates from marine or freshwater systems. These sample types include aquatic plants, shells, and fish bones, but also bones of omnivores such as humans, pigs, and dogs. Charred food remains on pottery, the so-called food crusts, also fall into this category.

The freshwater reservoir effect has been known for over 60 yr (Godwin 1951; Deevey et al. 1954). Since then, it has mainly been examined in modern aquatic plants and animals (Broecker and Walton 1959; MacDonald et al. 1987). The past few decades have seen growing recognition of its effect on archaeology (Lanting and van der Plicht 1995/1996; Cook et al. 2001; Shishlina et al. 2007; Olsen and Heinemeier 2009; Smits and van der Plicht 2009; Olsen et al. 2010), and also of its importance in pottery dating (Fischer and Heinemeier 2003; Boudin et al. 2009; Philippsen 2010; Philippsen et al. 2010).

For a characterization of the reservoir effect in a certain freshwater system, modern samples of water, aquatic plants, and fish are often dated. In early studies, limited sample numbers led to simplified conclusions (Deevey et al. 1954; Oana and Deevey 1960). However, research has indicated a large degree of variability of the freshwater reservoir effect (Olsson et al. 1969; Stuiver 1975; Philippsen 2012). This study examines the short-term variability of the freshwater reservoir age Rf and variations between different species of aquatic plants and animals. The sample types we analyze follow the food chain, from water dissolved inorganic carbon (DIC) over aquatic plants to fish and shellfish, in the Alster and Trave rivers in northern Germany.

The archaeological purpose of this study is to date the region’s earliest pottery, which belongs to the Late Mesolithic Ertebølle culture. The oldest dates for pottery of this hunter-fisher-gatherer culture have been obtained from the sites Kayhude at the Alster River and Schlamersdorf at the Trave River. Frequent inundations of the rivers caused an unclear stratigraphy. Therefore, it is very difficult to find terrestrial samples clearly associated with the potsherds and the pottery has to be dated directly. Because of the high age of the pottery, up to 1000 yr older than the same type of pottery from coastal sites in Schleswig-Holstein, a freshwater reservoir effect had been suspected.

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LOCATION

Figure 1 shows a map with the rivers and archaeological sites that are examined in this study: Kayhude at the Alster River and Schlammersdorf at the Trave River. The Alster discharges into the Elbe in Hamburg, the Trave into the Baltic Sea at Lübeck/Travemünde. The rivers run through a morainal landscape from the last 2 glaciations (Stewig 1982).

The Alster passes upper and lower moraines (the former from the Weichselian glaciation, the latter from the Saale glaciation). The pedogenic bedrock in the upper moraines is glacial till, Geschiebemergel, with about 20% calcium carbonate. The glacial sand in the lower moraines, Geschiebesand, only contains 0–5% calcium carbonate (Stewig 1982 and references therein).

The site Kayhude is situated in a narrow floodplain (~50 m wide) with geest ridges on both sides of the Alster. Two fens, the Wakendorfer Moor to the north of the site, to the south the Niendorfer Moor, are likely to be former lakes. Pollen analysis indicated shallow-water slow sedimentation. The Alster at that time was ~50 m wide and often changed its riverbed (Clausen 2007; I Clausen, personal communication, 2007).

The Trave runs through a morainal landscape originating from the Weichselian glaciation, the upper moraines of which contain about 20% calcium carbonate (Stewig 1982 and references therein). The Trave, a lake/river system in the Mesolithic, has been straightened in historical times. On one of these occasions in the 1930s, stone artifacts were found 3 km northwest of Schlammersdorf and provided a first indication for the presence of an archaeological site (Hartz 1997).
During the Atlantic period, there was a large body of water in the Trave Valley with a slow current. Probably, the Trave ran through this lake system already during the Boreal (Cimiotti 1983). In the Atlantic, a strong aggradation of the lake basin started the formation of a lowland with fens through which the Trave River was flowing, surrounded by wetlands. Through the dry phase of the Sub-Boreal, the water level sank and forests spread along the river, resulting in the formation of peat (Bruchwaldtorf) but were drowned again during the cool and humid Sub-Atlantic, leading to fens and the formation of fen peat. Today, the Trave Valley would still be dominated by fens if there had not been changes of the landscape through straightening of the river and drainage (Cimiotti 1983). The present-day Trave flows through a lake, Wardersee, which has an average depth of 3.7 m and consists of 2 parts that are only connected via a 5-m-wide, shallow ditch (Nixdorf et al. 2003). In spite of the straightenings, the Trave Valley is called “one of the best-preserved examples of subglacial meltwater tunnel valleys” with an almost undisturbed landscape (Ministerium für Umwelt, Naturschutz und Landwirtschaft des Landes Schleswig-Holstein 2003).

**MATERIALS AND METHODS**

In this section, we describe the sampling, cleaning, CO₂ preparation, and measurement for the modern and archaeological samples analyzed in this study.

**Water**

On 21 August 2007, 25 September 2008, 18 February 2009, and 6 July 2010, water samples were collected in 0.5L bottles and preserved with HgCl₂. The samples were kept dark and cool until analysis. The water was acidified with 100% H₃PO₄, which converted all DIC into CO₂. N₂ was bubbled through the water and the CO₂ was trapped cryogenically.

**Modern Plants and Animals**

Aquatic macrophytes and animals were collected at the same sites as the water samples. They were freeze-dried prior to analysis. No visible carbonate encrustations were found on the aquatic plants. HCl pretreatment was therefore not considered necessary. Local fishermen provided fish from the rivers. Collagen was extracted from some recent fish bones, as this is the material used for analyses of archaeological bones. A modified Longin procedure with ultrafiltration was used (Longin 1971; Brown et al. 1988; Jørkov et al. 2007).

**Archaeological Samples**

Archaeological charcoal samples were pretreated with 1M HCl at 80 °C for 1 hr, 1M NaOH at 80 °C for at least 3 hr, and 1M HCl at 20 °C overnight. Archaeological food crusts were pretreated like charcoal, but at 20 °C, and with 0.5 or 0.2M NaOH. The samples were converted to CO₂ by combustion in sealed evacuated quartz tubes containing CuO.

**Shells**

Shells were cleaned with ultrasound in demineralized water. Depending on size, the outer 10–25% of the shell was dissolved with 1M HCl. Possible organic remains were removed with KMnO₄ at 80 °C. Some 13–14 mg of pretreated shell was dissolved in 100% H₃PO₄ at 25 °C, to produce CO₂ for ¹⁴C dating and measurement of δ¹³C_shell and δ¹⁸O_shell.
Graphitization and Radiocarbon Dating

For $^{14}$C dating, the CO$_2$ was converted to graphite with the H$_2$ reduction method (Vogel et al. 1984). The samples were measured at the AMS $^{14}$C Dating Centre at Aarhus University (lab code AAR-). The dating results are reported according to international convention (Stuiver and Polach 1977) as conventional $^{14}$C age BP.

Marine reservoir ages are commonly denoted $R$ (Stuiver et al. 1986). Accordingly, we call freshwater reservoir ages $R_f$, measured in $^{14}$C yr. $R_f$ have been estimated from the measured age BP and the $^{14}$C-concentration of the atmosphere during the preceding growing season. $R_f$ of water samples has been estimated by comparing the measured age BP with the $^{14}$C level of the atmosphere at the time of sampling (rather than averaging over the entire growing season). The atmospheric data used for these calculations is shown in Table 1.

Table 1 Atmospheric $^{14}$C levels, measured on atmospheric CO$_2$, expressed as pMC and $^{14}$C age, which were used for estimating reservoir ages. Measurements from the Black Forest station Schauinsland are used (Levin et al. 2010; I Levin, personal communication, 2012). In spite of the high altitude, they are assumed to be a better estimate than the available data from a low-altitude station, Heidelberg, in the heavily polluted Rhine-Neckar area, which is affected both by additional $^{14}$C from a nearby nuclear power plant and $^{14}$C-free CO$_2$ from industry, heating, and transport (Levin et al. 2008).

<table>
<thead>
<tr>
<th>Month</th>
<th>Mean pMC</th>
<th>Mean $^{14}$C age (BP)</th>
</tr>
</thead>
<tbody>
<tr>
<td>August 2007</td>
<td>105.80</td>
<td>–453</td>
</tr>
<tr>
<td>September 2008</td>
<td>105.44</td>
<td>–425</td>
</tr>
<tr>
<td>February 2009</td>
<td>104.47</td>
<td>–352</td>
</tr>
<tr>
<td>July 2010</td>
<td>104.83</td>
<td>–379</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Growing season up to</th>
<th>Mean pMC</th>
<th>Mean $^{14}$C age (BP)</th>
</tr>
</thead>
<tbody>
<tr>
<td>August 2007</td>
<td>105.64</td>
<td>–440</td>
</tr>
<tr>
<td>September 2008</td>
<td>105.39</td>
<td>–421</td>
</tr>
<tr>
<td>February 2009</td>
<td>105.39</td>
<td>–421</td>
</tr>
<tr>
<td>July 2010</td>
<td>104.52</td>
<td>–355</td>
</tr>
</tbody>
</table>

Monthly means are given for the months in which water samples had been collected, and averages of the growing seasons before the sampling of aquatic plants and animals (data from Levin et al. 2010; I Levin, personal communication, 2012). Errors are insignificant compared to the measurements on water DIC, plants, and animals, and thus not given in this table.

Stable Isotope Measurements

$\delta^{13}$C and $\delta^{15}$N analyses on organic samples were performed by combustion in a EuroVector elemental analyzer coupled to an IsoPrime stable isotope ratio mass spectrometer at the Aarhus AMS Centre. $\delta^{13}$C values are reported as ‰ VPDB (Coplen 1994), $\delta^{15}$N values as ‰ AIR (Mariotti 1983). The C/N ratios were derived from total organic carbon (TOC) and total nitrogen (TN) measurements and are presented as atomic ratios.

$\delta^{13}$C of water DIC, shell carbonate, and many organics was measured on a CO$_2$ aliquot from the $^{14}$C preparation using a dual-inlet IsoPrime stable isotope mass spectrometer at the Aarhus AMS Centre. Measurements were performed relative to the internal standard material Carrara CaCO$_3$ and reported as ‰ VPDB. $\delta^{13}$C values obtained using the elemental analyzer are marked (EA), those using the dual inlet are marked (DI).
Additional Methods

In 2007, 2008, and 2010, the bicarbonate concentration was measured immediately after sample collection with test strips (Merckoquant Carbonate Hardness Test 1.10648.0001). Measurements of the concentrations of Ca\(^{2+}\), Mg\(^{2+}\), Na\(^+\), and K\(^+\) were obtained from “Landesamt für Landwirtschaft, Umwelt und ländliche Räume des Landes Schleswig–Holstein,” formerly “Landesamt für Natur und Umwelt.” Both survey stations, Wulksfelde/Alster and Warderbrück/Trave, downriver of the lake Wardersee, are marked with a black X on Figure 1.

RESULTS AND DISCUSSION

Water Samples

The \(^{14}\)C and \(\delta^{13}\)C results on water DIC are presented in Table 2 and Figure 2. All water samples have \(^{14}\)C ages >1000 BP (1170–2620 BP), instead of the –400 BP expected in case of equilibrium with the atmosphere. The \(^{14}\)C age of Alster DIC is greater than that of Trave DIC for every sampling date (Table 2). On average, the estimated reservoir age of Alster DIC is \(R_f = 2490 \pm 490\), and for the Trave \(R_f = 2050 \pm 560\). \(\delta^{13}\)C values range between –15 and –9‰. \(\delta^{13}\)C values in the Alster are, with the exception of September 2008 between 1.37 and 5.91‰ lower than in the Trave.

\[\delta^{13}\C\] and the \(^{14}\)C age (Figure 2) are strongly correlated in the Alster (\(r = 0.82, R^2 = 0.67\)), but not in the Trave (\(r = 0.11, R^2 = 0.01\)). Without the samples from February 2009, though, the correlation coefficients are \(r = 0.95, R^2 = 0.91\) for the Alster and \(r = 0.89, R^2 = 0.79\) for the Trave, as the removal of the February values reduces most of the seasonal spread. This agrees with our expectation that higher \(\delta^{13}\)C values indicate a higher proportion of fossil, \(^{14}\)C-free carbonates (which have \(\delta^{13}\C \approx 0\)). The \(^{14}\)C age of Alster DIC is greater than that of Trave DIC for every sampling date (Table 2), although the \(\delta^{13}\)C values are lower. If the only source for large \(^{14}\)C ages was dissolved carbonate minerals, and the only source for small \(^{14}\)C ages soil CO\(_2\), then the low \(\delta^{13}\)C values of the Alster would be inconsistent with the large \(^{14}\)C ages. A possible explanation is mineralization of old organic matter, such as peat, in the catchment of the Alster. Another possible explanation is the fact that the Trave flows through the shallow lake Wardersee (Nixdorf et al. 2003). This leads to a comparatively long residence time of the water and facilitates exchange with atmospheric CO\(_2\).
The water hardness measured with test strips and the water hardness derived from Ca and Mg measurements agree within the large error margins and vary between 9 and 13 °dH. The exception is the Ca and Mg measurement of the Trave in February 2009, which corresponds to 18 °dH.

Daily sums (in mm) of precipitation were obtained from the German Weather Service for the 3 stations Schleswig, Fehmarn, and Hamburg-Fuhlsbüttel (see Figure 1; Deutscher Wetterdienst 2007–2010). The amount of precipitation in the week and in the month before sample collection was calculated and is presented in Table 3.

Table 3 Sum precipitation of the 7-day period before sampling and of the 30-day period before sampling is given for 3 measurement stations: Hamburg-Fuhlsbüttel, Schleswig, and Fehmarn (see Figure 1). Data from Deutscher Wetterdienst (2007–2010).

<table>
<thead>
<tr>
<th>Date</th>
<th>Precipitation (mm/week) Fuhlsbüttel</th>
<th>Precipitation (mm/month) Fuhlsbüttel</th>
<th>Precipitation (mm/week) Schleswig</th>
<th>Precipitation (mm/month) Schleswig</th>
<th>Precipitation (mm/week) Fehmarn</th>
<th>Precipitation (mm/month) Fehmarn</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aug 2007</td>
<td>10</td>
<td>134.9</td>
<td>19.3</td>
<td>116.8</td>
<td>10.1</td>
<td>103.8</td>
</tr>
<tr>
<td>Sep 2008</td>
<td>1.3</td>
<td>14</td>
<td>2.4</td>
<td>33.9</td>
<td>0.6</td>
<td>19.2</td>
</tr>
<tr>
<td>Feb 2009</td>
<td>6.7</td>
<td>27.1</td>
<td>11.1</td>
<td>34.4</td>
<td>5.8</td>
<td>24.1</td>
</tr>
<tr>
<td>July 2010</td>
<td>7.1</td>
<td>26.8</td>
<td>0.2</td>
<td>36.1</td>
<td>0.6</td>
<td>28.7</td>
</tr>
</tbody>
</table>

pH measurements for the rivers were not available, but in the Wardersee the pH value is relatively stable around 8.4 (Nixdorf et al. 2003). At this pH, most of the DIC occurs in the form of bicarbonate (Olsson and Kaup 2001). The carbonate hardness (Table 2) should thus be a good indicator of DIC concentration. Carbonate hardness is also expected to indicate the DIC age, as larger amounts of dissolved 14C-free carbonates lead to greater 14C ages. However, the Alster water has greater 14C ages but lower Ca concentrations. None of the rivers show a correlation between DIC 14C ages and water hardness.

For both rivers, it was checked whether the amount of precipitation correlated with the 14C age or the δ13C values. In all cases, the correlations of 14C or δ13C with precipitation data are negative; maximum values are given in the list below. Thus, the more rain in the period before sampling,
Freshwater Reservoir Effect Variability in N. Germany

1. The smaller the $^{14}$C age (more “modern” carbon, relatively less CaCO$_3$; e.g. $r = -0.65$ for the $^{14}$C age of Alster DIC and weekly precipitation sum from Fehmarn and $r = -0.92$ for the $^{14}$C age of Trave DIC and weekly precipitation sum from Schleswig);
2. The lower the $\delta^{13}$C values (more terrestrial material; e.g. $r = -0.94$ for $\delta^{13}$C values of Alster DIC and weekly precipitation sum from HH-Fuhlsbüttel and $r = -0.76$ for $\delta^{13}$C values of Trave DIC and monthly precipitation sums from Schleswig and Fehmarn).

In almost all cases, the correlation of the precipitation of the week before sample collection and the $^{14}$C age is stronger than the correlation between monthly precipitation and $^{14}$C age. Short-term precipitation fluctuations thus appear to have a considerable influence on $^{14}$C age and $\delta^{13}$C values. This could explain the large short-term variability of the values. The fact that the greatest correlation of Trave DIC $\delta^{13}$C values occurs with monthly precipitation sums indicates a larger water residence time in this river system. However, more data would be needed to draw any firm conclusions.

Aquatic Plants

The aquatic plants have $^{14}$C ages between −74 and 2273 BP (Table 4). Comparison with the average $^{14}$C level of the atmosphere during the preceding growing season (Table 1) yielded estimated reservoir ages (Rf) from 347 to 2700 $^{14}$C yr. $^{14}$C ages do not differ systematically between submerged and floating leaves. The $^{14}$C age range of the plants overlaps with that of the water (which has $^{14}$C ages between 1170 and 2620 BP), but is shifted towards lower values. The most striking result of the analysis of aquatic plants is the fact that floating leaves do not have younger $^{14}$C ages than submerged plants. A submerged plant with an estimated Rf of only 350 $^{14}$C yr contrasts with a floating plant, collected on the same day at the same part of the river, with a much higher Rf of 1300 $^{14}$C yr (Table 4).

Two leaves of the yellow water lily Nuphar lutea, one from Alster and one from Trave, have been subsampled twice, at the bottom of the petiole (stem) and at the tip of the leaf. In both cases, the tip of the leaf is slightly older than the end of the petiole, and both subsamples show a substantial reservoir effect. The estimated reservoir age of the N. lutea from the Trave is Rf = 600–640 $^{14}$C yr, from the Alster, Rf = 2300–2500 $^{14}$C yr. N. lutea is proven to assimilate CO$_2$ (Birks 2001). This CO$_2$ can derive from different sources, including the atmosphere, water, and sediment. Furthermore, the petiole of N. lutea uses nutrients from the rhizome (root stock) that were stored during the previous growing season (Osmond et al. 1981). These different carbon sources have potentially very different $^{14}$C ages. Sediment organic matter, for example, can be several decades old and thus heavily affected by bomb carbon (Olsson and Kaup 2001).

“Internal winds” in N. lutea transport atmospheric air from the younger leaves through petioles and rhizome to the older leaves, where most of the CO$_2$ from this transport is photosynthesized (Dacey 1980). This continuous air transport is most likely the reason for petioles and leaves having the same $^{14}$C ages. The high ages of the water lilies (Table 4) indicate a surprisingly large contribution from water CO$_2$ for photosynthesis.

Our results disagree with previous studies where emergent plants and floating leaves of N. lutea were found to have $^{14}$C contents in equilibrium with the atmosphere (Divey et al. 1954; Olsson and Kaup 2001). In a study by Törnvqvist et al. (1992), however, N. lutea showed a full hardwater effect of about Rf = 500 yr, while the white water lily Nymphaea alba had a terrestrial $^{14}$C age. Differences in $^{14}$C levels for aquatic plants from the same region, as well as high and variable reservoir ages without differences in floating and submerged plants have previously been found (Olsson et al. 1969; Srdoč et al. 1980).
Table 4 Modern aquatic plants and animals.

<table>
<thead>
<tr>
<th>River</th>
<th>Collection date</th>
<th>AAR-</th>
<th>Species</th>
<th>C fraction</th>
<th>N fraction</th>
<th>C/N ratio</th>
<th>Δ(^{13})C (% VPDB)</th>
<th>Δ(^{15})N(_{AIR}) (%)</th>
<th>Δ(^{18})O(_{VPDB}) (%)</th>
<th>pMC</th>
<th>(^{14})C age (BP)</th>
<th>Res. age (14C BP)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Alster</td>
<td>Sep 2008</td>
<td>12873</td>
<td>Submerged plant</td>
<td>0.221 ± 0.132</td>
<td>0.025 ± 0.018</td>
<td>7.54 ± 0.56</td>
<td>−31.62 ± 0.23(^{b})</td>
<td>12.63 ± 0.58</td>
<td>−2237 ± 41</td>
<td>75.36 ± 0.38</td>
<td>2273 ± 24</td>
<td>2694 ± 41</td>
</tr>
<tr>
<td>Alster</td>
<td>July 2010</td>
<td>14334</td>
<td>Nuphar lutea (tip of leaf)</td>
<td>0.419 ± 0.209</td>
<td>0.049 ± 0.001</td>
<td>10.02 ± 1.74</td>
<td>−31.50 ± 0.05(^{c})</td>
<td>10.40 ± 0.27</td>
<td>2117 ± 25</td>
<td>76.83 ± 0.24</td>
<td>2117 ± 25</td>
<td>2472 ± 25</td>
</tr>
<tr>
<td>Alster</td>
<td>July 2010</td>
<td>14335</td>
<td>Nuphar lutea (end of petiole)</td>
<td>0.364 ± 0.038</td>
<td>0.023 ± 0.008</td>
<td>17.58 ± 3.20</td>
<td>−31.24 ± 0.05(^{c})</td>
<td>10.39 ± 0.49</td>
<td>1944 ± 24</td>
<td>78.51 ± 0.23</td>
<td>1944 ± 24</td>
<td>2299 ± 24</td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12870</td>
<td>Submerged plant</td>
<td>0.263 ± 0.020</td>
<td>0.022 ± 0.007</td>
<td>10.36 ± 0.68</td>
<td>−25.42 ± 0.46(^{b})</td>
<td>9.89 ± 0.52</td>
<td>100.93 ± 0.44</td>
<td>74 ± 35</td>
<td>100.93 ± 0.44</td>
<td>74 ± 35</td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12871</td>
<td>Floating plant</td>
<td>0.377 ± 0.021</td>
<td>0.043 ± 0.003</td>
<td>10.94 ± 0.83</td>
<td>−28.09 ± 0.73(^{c})</td>
<td>12.05 ± 0.55</td>
<td>879 ± 37</td>
<td>1300 ± 37</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12872</td>
<td>Submerged plant</td>
<td>0.255 ± 0.001</td>
<td>0.024 ± 0.001</td>
<td>11.54 ± 0.46</td>
<td>−17.45 ± 1.88(^{c})</td>
<td>6.81 ± 2.51</td>
<td>2100 ± 55</td>
<td>2100 ± 55</td>
<td>2100 ± 55</td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>July 2010</td>
<td>14336</td>
<td>Submerged plant</td>
<td>0.319 ± 0.004</td>
<td>0.032 ± 0.001</td>
<td>11.58 ± 0.86</td>
<td>−34.22 ± 0.05(^{c})</td>
<td>12.82 ± 0.15</td>
<td>2269 ± 23</td>
<td>2269 ± 23</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>July 2010</td>
<td>14337</td>
<td>Submerged/floating plant</td>
<td>0.427 ± 0.006</td>
<td>0.063 ± 0.002</td>
<td>7.51 ± 0.79</td>
<td>−26.95 ± 0.05(^{c})</td>
<td>−3.35 ± 0.40</td>
<td>1618 ± 23</td>
<td>1618 ± 23</td>
<td>1618 ± 23</td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>July 2010</td>
<td>14338</td>
<td>Nuphar lutea (tip of leaf)</td>
<td>0.415 ± 0.017</td>
<td>0.061 ± 0.010</td>
<td>10.18 ± 1.40</td>
<td>−27.52 ± 0.05(^{c})</td>
<td>7.86 ± 0.23</td>
<td>643 ± 20</td>
<td>288 ± 20</td>
<td>643 ± 20</td>
<td>643 ± 20</td>
</tr>
<tr>
<td>Trave</td>
<td>July 2010</td>
<td>14339</td>
<td>Nuphar lutea (end of petiole)</td>
<td>0.350 ± 0.007</td>
<td>0.020 ± 0.001</td>
<td>20.51 ± 0.49</td>
<td>−26.67 ± 0.10(^{c})</td>
<td>5.35 ± 0.12</td>
<td>97.04 ± 0.30</td>
<td>241 ± 25</td>
<td>97.04 ± 0.30</td>
<td>241 ± 25</td>
</tr>
<tr>
<td>Alster</td>
<td>Sep 2007</td>
<td>11460</td>
<td>Mussel shell (probably Unio sp.)</td>
<td>−13.22 ± 0.05</td>
<td>−7.21 ± 0.06</td>
<td>85.98 ± 0.37</td>
<td>−1214 ± 34</td>
<td>1654 ± 35</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Alster</td>
<td>Sep 2007</td>
<td>11461</td>
<td>Snail shell</td>
<td>−15.36 ± 0.05</td>
<td>−8.01 ± 0.05</td>
<td>94.75 ± 0.37</td>
<td>433 ± 32</td>
<td>869 ± 34</td>
<td></td>
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</tr>
<tr>
<td>Alster</td>
<td>Sep 2007</td>
<td>11462</td>
<td>Roach, bone collagen</td>
<td>0.362 ± 0.007</td>
<td>0.132 ± 0.001</td>
<td>3.19 ± 0.03</td>
<td>−25.46 ± 0.05(^{b})</td>
<td>12.24 ± 0.33</td>
<td>660 ± 33</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2007</td>
<td>11394</td>
<td>Roach, bone collagen</td>
<td>0.422 ± 0.010</td>
<td>0.156 ± 0.003</td>
<td>3.15 ± 0.10</td>
<td>−25.91 ± 0.05(^{b})</td>
<td>14.85 ± 0.44</td>
<td>727 ± 33</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2007</td>
<td>11396</td>
<td>Roach, bone collagen (from cooked bone)</td>
<td>0.428 ± 0.011</td>
<td>0.161 ± 0.004</td>
<td>3.11 ± 0.10</td>
<td>−24.24 ± 0.05(^{c})</td>
<td>15.27 ± 1.03</td>
<td>685 ± 25</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12874</td>
<td>Mallard feather</td>
<td>0.498 ± 0.024</td>
<td>0.161 ± 0.004</td>
<td>3.74 ± 0.28</td>
<td>−23.99 ± 0.11(^{b})</td>
<td>4.81 ± 0.24</td>
<td>47 ± 31</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12875</td>
<td>Spined loach</td>
<td>0.485 ± 0.008</td>
<td>0.132 ± 0.001</td>
<td>4.32 ± 0.12</td>
<td>−27.24 ± 0.09(^{b})</td>
<td>15.55 ± 0.34</td>
<td>2085 ± 39</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12876</td>
<td>Crayfish</td>
<td>0.390 ± 0.114</td>
<td>0.060 ± 0.036</td>
<td>4.20 ± 0.06</td>
<td>−27.89 ± 0.46(^{b})</td>
<td>11.89 ± 2.04</td>
<td>1787 ± 40</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Trave</td>
<td>Sep 2008</td>
<td>12878</td>
<td>Roach, flesh</td>
<td>0.486 ± 0.022</td>
<td>0.144 ± 0.008</td>
<td>3.92 ± 0.28</td>
<td>−22.30 ± 0.10(^{c})</td>
<td>14.86 ± 0.19</td>
<td>488 ± 32</td>
<td></td>
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<td></td>
</tr>
</tbody>
</table>

\(^{a}\)Δ\(^{18}\)O values are shaded gray; all other (unshaded) values are Δ\(^{15}\)N.

\(^{b}\)Samples combusted in an elemental analyzer prior to measurement.

\(^{c}\)Measurements on sample CO\(_2\) from offline combustion or acidification with the dual-inlet (DI) method.
In our case, the differences between the 2 rivers are larger than the differences between the 2 collection dates. The average Rf values of the, admittedly few, samples are 2490 ± 200 for the Alster and 1270 ± 770 for the Trave. The samples collected in September 2008 have an average Rf of 1620 ± 1020, while the samples collected in July 2010 on average have Rf = 1650 ± 850. The average values are given below:

<table>
<thead>
<tr>
<th></th>
<th>September 2008</th>
<th>July 2010</th>
</tr>
</thead>
<tbody>
<tr>
<td>Alster</td>
<td>2690 ± 40 (1 sample)</td>
<td>2390 ± 120 (2 samples)</td>
</tr>
<tr>
<td>Trave</td>
<td>1260 ± 890 (3 samples)</td>
<td>1280 ± 810 (4 samples)</td>
</tr>
</tbody>
</table>

Also, the range of δ¹³C values is large for the plants analyzed here and spans from −34.2 to −17.5‰ (Table 4). Eight out of the 10 plants have values between −31.6 and −25.4‰; the broad range is thus caused by 2 extreme values. No correlation of δ¹³C values with growing location (submerged or floating) was seen. δ¹³C values of aquatic plants are expected to reflect the isotopic composition of their source carbon, apart from an isotopic shift due to fractionation, usually 18–19‰ from bicarbonate to plant cell (Olsson and Kaup 2001). Regarding water DIC δ¹³C values (Table 2), the water plants in this study are expected to have δ¹³C values between −33 and −29‰; the measured values, however, were −34.2 to −17.5‰. The more enriched values are probably caused by reduced fractionation as a consequence of bicarbonate limitation. It has even been reported that water DIC and aquatic plants can have the same δ¹³C values (Higham et al. 2010).

Many aquatic plants can utilize both DIC species, CO₂ and HCO₃⁻, whereas others are specialized in only one of them (Osmond et al. 1981). CO₂ in equilibrium with HCO₃⁻ at 10 °C has ~9.5‰ lighter values than the HCO₃⁻ (Emrich et al. 1970; Romanek et al. 1992). The dominating DIC species in the rivers analyzed here is bicarbonate. Plants utilizing CO₂ experience thus a limited carbon pool and discriminate less against ¹³C than plants utilizing bicarbonate. The suitability of δ¹³C values of aquatic plants for identifying CO₂ or HCO₃⁻ assimilation is therefore limited, as low δ¹³C values can be caused by the assimilation of HCO₃⁻ involving large fractionation or by the assimilation of CO₂ without fractionation.

C/N ratios span a large range from 7.5 to 20.5, which corresponds to a range from typical values of algae to those of terrestrial matter (Meyers and Teranes 2001). Six out of 10 of the aquatic plants analyzed here, however, fall into a narrower range of C/N ratios between 10 and 11.5 (Table 4). Floating leaves tend to have higher carbon fractions than submerged plants or parts of plants. On average, floating leaves have a carbon fraction of 0.410 ± 0.022, and submerged plants and parts of plants, 0.295 ± 0.057. No other measured parameter distinguishes between submerged and floating plants.

Studies of reservoir ages and δ¹³C values of aquatic plants are highly important. Especially the carbon source of floating leaves should be explored. *Nymphaea alba*, for example, has been used for constructing an age model, just as terrestrial samples (Hammarlund et al. 2003). Furthermore, the rhizomes of both *Nuphar* and *Nymphaea* have been widely used as human food (Hutchinson 1975 and references therein). Freshwater reservoir effects in water lilies could thus introduce reservoir effects into age models and human nutrition.

**Aquatic Animals**

The average Rf of the animals from Alster and Trave is 1120 ± 620 ¹⁴C yr (excluding the mallard feather). For the Alster alone, the average Rf is 1060 ± 520, and for the Trave, Rf = 1150 ± 730. The large variability of ¹⁴C ages for fish and other freshwater animals (Table 4) is not surprising, regarding the large variability on the basis of the food web, including water DIC and aquatic plants. How-
ever, the average $R_f$ of the animals does not differ between Alster and Trave, while the average $R_f$ of aquatic plants from the 2 rivers differed substantially (see above). Organic samples (fish bone collagen, flesh, and whole animals) have $\delta^{13}C$ values between –22 and –28‰.

The bone collagen of $R. rutilus$ collected in 2007 had $\delta^{13}C$ values around –25‰, and experimental food crusts made of flesh from these fish had $\delta^{13}C$ values between –27 and –29‰. The raw flesh was not measured but is expected to have approximately the same $\delta^{13}C$ values as the food crust. The difference between flesh and bone collagen as measured here corresponds thus to the collagen-flesh differences in previous studies (Katzenberg et al. 1995; Lanting and van der Plicht 1998). However, the flesh of a $R. rutilus$ collected in 2008 had a $\delta^{13}C$ value of –22.3‰. Unfortunately, bone collagen from this fish was not measured. However, if the same flesh-bone collagen fractionation applies to this fish as well, its bone collagen would have $\delta^{13}C$ values between –20 and –18‰. The large difference between the fish from 2008 and those from 2007 cannot be explained with our current knowledge but is most probably a result of different food sources of the individual fish.

$\delta^{13}C$ values and $^{14}C$ ages of $R. rutilus$ bone collagen are of the same order of magnitude for both rivers, with $\delta^{13}C$ values around –25‰ and $^{14}C$ ages around 250 BP. This yields an estimated reservoir age between $R_f$ = 660 and 730 $^{14}C$ yr. The flesh from a roach caught 1 yr later, however, had a $\delta^{13}C$ value of –22.3‰ and a $^{14}C$ age of 70 BP ($R_f$ = 490 $^{14}C$ yr). The fish with the unusually high $\delta^{13}C$ value thus also had an unusually low $^{14}C$ age.

$\delta^{13}C$ values of inorganic samples (shells) are –13‰ and –15‰, while shell $\delta^{18}O$ values are –7.2 and –8.0‰. These $\delta^{13}C$ and $\delta^{18}O$ values of the shells from the Alster are comparable to values reported in the literature (e.g. Keith et al. 1964). The $\delta^{13}C$ values are furthermore comparable to water DIC $\delta^{13}C$ values from the same river and period (Table 2). The mallard feather has $\delta^{15}N$ = 4.8, whereas the fish samples have $\delta^{15}N$ values between 12 and 16‰. The high $\delta^{15}N$ values of the fish agree with the high $\delta^{15}N$ values of the aquatic plants (see above and Figure 3).

Figure 3 Measurements on modern aquatic plants and animals. Squares: Trave; Circles: Alster; Green solid: plants; Orange hatched: animals. Error bars in the left plot are of the same magnitude as symbols and therefore left out. Values in Table 4.

The $\delta^{13}C$ values and $^{14}C$ ages of modern aquatic organisms are correlated (cf. Figure 3). Without the 2 samples of shell carbonate, the Pearson correlation coefficient is $r = –0.52$ ($R^2 = 0.27$). When excluding an outlier value of an aquatic plant ($\delta^{13}C = –17.45$), the correlation coefficient is $r = –0.88$ ($R^2 = 0.79$). Interestingly, the correlation coefficients between $\delta^{13}C$ values and $^{14}C$ ages are positive for water DIC (see above and Figure 2).
Interestingly, the average reservoir ages $R_f$ of water DIC and aquatic plants are equal in the Alster, while they differ substantially in the Trave. The animals from the Alster, however, have significantly lower $R_f$ than the plants. In the Trave, on the other hand, aquatic plants and animals have similar average $R_f$ values. We have not yet been able to find a satisfactorily explanation for these similarities and differences.

**Archaeological Samples**

Two terrestrial samples, 1 fish bone, and 4 food crusts on pottery from the archaeological site Kayhude at Alster River were dated. The results are presented in Table 5 and Figure 4. The samples from Kayhude are believed to be contemporaneous, as they were found embedded in a stone layer. Still, the 2 terrestrial samples have very different $^{14}$C ages: 5450 and 9150 BP (Table 5, Figure 4). This bone must be an admixture from earlier layers, as it is not only older than the other terrestrial sample from Kayhude, but also older than the oldest finds of the entire Ertebølle culture. This exemplifies that the stone layer where we found our samples cannot be regarded as totally undisturbed. Direct $^{14}$C dating of the pottery is thus necessary, as we cannot be sure which terrestrial samples are clearly associated with the pottery.

<table>
<thead>
<tr>
<th>AAR-</th>
<th>Species/material</th>
<th>$^{14}$C age (uncal yr BP)</th>
<th>C/N ratio</th>
<th>$\delta^{13}$C (% VPDB)</th>
<th>$\delta^{15}$N (% AIR)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Kayhude, Alster</td>
<td></td>
<td></td>
<td></td>
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<td></td>
</tr>
<tr>
<td>11403 KAY8-432.01 FC pretreated</td>
<td>5695 ± 55</td>
<td>8.77 ± 0.38</td>
<td>−28.63 ± 0.05 (DI)</td>
<td>6.99 ± 0.43</td>
<td></td>
</tr>
<tr>
<td>11403 KAY8-432.01 FC not pretreated</td>
<td>—</td>
<td>8.93 ± 0.50</td>
<td>−29.01 ± 1.03 (EA)</td>
<td>6.76 ± 0.09</td>
<td></td>
</tr>
<tr>
<td>11404 KAY8-168.01 FC base-soluble</td>
<td>6740 ± 160</td>
<td>—</td>
<td>—</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11404 KAY8-168.01 FC pretreated</td>
<td>6090 ± 55</td>
<td>8.28 ± 0.91</td>
<td>−28.90 ± 0.05 (DI)</td>
<td>12.54 ± 0.21</td>
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</tr>
<tr>
<td>11404 KAY8-168.01 FC not pretreated</td>
<td>—</td>
<td>8.67 ± 0.43</td>
<td>−28.79 ± 0.17 (EA)</td>
<td>11.36 ± 0.21</td>
<td></td>
</tr>
<tr>
<td>11477 KAY8-815.0 BC terrestrial</td>
<td>9150 ± 110</td>
<td>—</td>
<td>—</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11479 KAY8-412.01 FC pretreated</td>
<td>5350 ± 110</td>
<td>17.77 ± 0.32</td>
<td>−26.53 ± 0.13 (EA)</td>
<td>6.38 ± 0.34</td>
<td></td>
</tr>
<tr>
<td>11479 KAY8-412.01 FC not pretreated</td>
<td>—</td>
<td>17.41 ± 1.06</td>
<td>−26.55 ± 0.10 (EA)</td>
<td>7.44 ± 1.00</td>
<td></td>
</tr>
<tr>
<td>11479 KAY8-412.01 FC base-soluble</td>
<td>6130 ± 60</td>
<td>—</td>
<td>—</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11480 KAY8-176 Charcoal</td>
<td>5438 ± 41</td>
<td>—</td>
<td>−24.83 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11695 Pike BC</td>
<td>8520 ± 80</td>
<td>—</td>
<td>−22.41 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>12412 KAY8-435 FC pretreated</td>
<td>5948 ± 35</td>
<td>15.46 ± 24.63</td>
<td>−26.72 ± 0.10 (EA)</td>
<td>—</td>
<td></td>
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<tr>
<td>Schlammersdorf, Trave</td>
<td></td>
<td></td>
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<td></td>
<td></td>
</tr>
<tr>
<td>11402 SLA5-10000 Wood</td>
<td>5638 ± 49</td>
<td>—</td>
<td>−26.49 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11405 SLA5-10001 Wood</td>
<td>5762 ± 48</td>
<td>—</td>
<td>−27.25 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11406 SLA5-10002 Wood</td>
<td>5818 ± 43</td>
<td>—</td>
<td>−28.78 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11407 SLA5-10003 Wood</td>
<td>5750 ± 90</td>
<td>—</td>
<td>−27.03 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11408 SLA5-10004 Wood</td>
<td>5642 ± 48</td>
<td>—</td>
<td>−27.47 ± 0.05 (DI)</td>
<td>—</td>
<td></td>
</tr>
<tr>
<td>11398 SLS-2784 BC Wild cat</td>
<td>5685 ± 60</td>
<td>3.25 ± 0.13</td>
<td>−19.27 ± 0.05 (DI)</td>
<td>6.49 ± 0.29</td>
<td></td>
</tr>
<tr>
<td>11399 SLS-2761 BC beaver</td>
<td>6480 ± 90</td>
<td>3.39 ± 0.11</td>
<td>−22.42 ± 0.05 (DI)</td>
<td>4.68 ± 0.89</td>
<td></td>
</tr>
<tr>
<td>11400 SLS-2883 BC wild boar</td>
<td>6015 ± 60</td>
<td>3.24 ± 0.01</td>
<td>−21.29 ± 0.05 (DI)</td>
<td>5.01 ± 0.13</td>
<td></td>
</tr>
<tr>
<td>11401 SLS-2913 BC fish; 5 vertebrae</td>
<td>—</td>
<td>3.27 ± 0.02</td>
<td>−21.40 ± 0.05 (EA)</td>
<td>4.75 ± 23</td>
<td></td>
</tr>
<tr>
<td>11401 SLS-2913 BC fish; rest of the bone fragments</td>
<td>—</td>
<td>3.25 ± 0.01</td>
<td>−26.91 ± 0.04 (EA)</td>
<td>7.91 ± 44</td>
<td></td>
</tr>
</tbody>
</table>
The pike bone collagen is ~3000 $^{14}$C yr older than the charcoal sample. Food crusts on pottery have the same or slightly larger $^{14}$C ages than the youngest terrestrial sample. None of the food crusts are as old as the fish bone. The base-soluble fraction of 3 food crusts has also been dated. It may consist of partly original material, e.g. fatty substances, but also partly of contamination from the soil such as humic acid. The base-soluble fraction is in all cases older than the food crusts (Figure 4), indicating contamination with an older soil substance.

![Figure 4](https://doi.org/10.1017/S0033822200048001) Archaeological samples from Kayhude/Alster. The plot was made with OxCal v 4.1 (Bronk Ramsey 2009), using a straight line $x=y$ as “calibration curve”; 2σ-intervals are indicated below the probability distributions of the $^{14}$C ages.

The pike bone collagen is ~3000 $^{14}$C yr older than the charcoal sample. Food crusts on pottery have the same or slightly larger $^{14}$C ages than the youngest terrestrial sample. None of the food crusts are as old as the fish bone. The base-soluble fraction of 3 food crusts has also been dated. It may consist of partly original material, e.g. fatty substances, but also partly of contamination from the soil such as humic acid. The base-soluble fraction is in all cases older than the food crusts (Figure 4), indicating contamination with an older soil substance.
If we assume that the charcoal AAR-11480 gives the correct age of the find layer, then the food crust AAR-11479 is not affected by reservoir effects. For the 3 other food crusts (AAR-11403, -11404, and -14212), reservoir ages of about $R_f = 300 - 600 \ 14C$ yr can be estimated. Compared to the average reservoir age of modern Alster animals, $R_f = 1060 \pm 520 \ 14C$ yr, this would indicate 30% to 60% aquatic ingredients in the food crust.

If we assume that the fish bone AAR-11695 is contemporaneous with the charcoal and food crust samples, then the reservoir effect in the Mesolithic at Kayhude would be about $R_f = 3000 \ 14C$ yr. In this case, the reservoir ages of the food crusts AAR-11403, -11404, and -14212 would indicate only 10% to 20% aquatic ingredients.

From Schlamersdorf at the Trave River, 9 terrestrial samples, 3 fish bones, and 5 food crusts on pottery were dated (Table 5, Figure 5). The age range of terrestrial samples is very broad, ~1000 yr (Figure 5). The food crusts have the same age as the terrestrial samples or are slightly older, and the fish bone collagen is significantly older than the terrestrial samples.

The terrestrial age range of Schlamersdorf complies with earlier charcoal datings from this site (Hartz 1993). The broad age range measured here (Figure 5) is unlikely to indicate an occupation period of 1000 yr. The site was probably occupied repeatedly for shorter periods, as archaeological analysis indicated that the site was a hunting or fishing station. The broad terrestrial age range reveals the necessity of direct pottery dating.
Three food crusts had previously been dated to ~5300 cal BC (Hartz 1996); their $\delta^{13}$C values between −28.6 and −31.9‰ indicate freshwater ingredients and thus the possibility of a freshwater reservoir effect. Two of the 4 food crusts we $^{14}$C dated from that site are from 5500–6000 BP (~4000–5000 cal BC), and 2 around 7000 BP (~5600–6000 cal BC; Figure 5). However, as the average $R_f$ in modern Trave animals is 1150 ± 730, the large ages of the 2 oldest potsherds could have been caused by a reservoir effect. In case fish or other aquatic resources had been prepared in these pots, their reservoir age $R_f$ could likely be ~1000 $^{14}$C yr. It is thus probable that the true ages of all the food crusts from Schlamersdorf are about the same, and lie within an interval of ~4000–5000 cal BC.

An interesting case is the potsherd AAR-11481 of which both the inner and outer crust have been dated. If one assumes that the outer crust is soot from the cooking fire, then it should give the date of cooking, or an older date in case old wood had been used. The reservoir effect would, in this case, be ~2000 yr. As this outer crust is younger than all the other terrestrial samples, it was suspected to be influenced by modern contamination. However, if it had been modern contamination from the burial environment, from the handling during the excavation, or later during storage in the archives, this contamination would be expected to have affected both sides of the sherd equally.

In one of the sherds, AAR-11483, we were lucky to find some plant remains that presumably had been incorporated into the clay during the forming of the pottery. The $^{14}$C age of these plant remains is 6000 BP. The calibrated 2σ age range is 4999–4766 cal BC (92.7%) and 4756–4729 cal BC (2.7%), calibrated with OxCal v 4.1.2 (Bronk Ramsey 2009) and the IntCal04 atmospheric curve (Reimer et al. 2004). The probability for the pottery being older than 5000 cal BC is thus <5%.

Unfortunately, the food crust sample of AAR-11483 was lost during dating. It would otherwise have helped to measure the reservoir effect in food crusts directly.

The hardwater effect at Schlamersdorf and Kayhude seems to be larger than the effect reported by Fischer and Heinemeier (2003), at least for the fish bones. In their study area, the Åmose in Zealand, Denmark, the fish was 100–500 $^{14}$C yr older than the archaeological context, while the food crusts were up to 300 $^{14}$C yr older.

CONCLUSION

The freshwater reservoir effect in the Alster and Trave rivers is large and very variable. In both rivers, $^{14}$C age and $\delta^{13}$C values of water DIC are correlated with each other and with the amount of precipitation in the week prior to sampling. After a period with small amounts of precipitation, the river water is dominated by hard groundwater with high $\delta^{13}$C values and $^{14}$C ages. Larger amounts of precipitation result in a larger contribution of terrestrial run-off, which takes up CO$_2$ with low $\delta^{13}$C values and low $^{14}$C ages from the root zone.

The high ages and large variability can also be found in aquatic plants. Short-term precipitation fluctuations are unlikely to be the reason, as the plants average $^{14}$C levels over the entire growing season. Aquatic plants can utilize a multitude of carbon species for photosynthesis with $^{14}$C ages varying between ancient (DIC from dissolved limestone), contemporaneous (atmospheric CO$_2$), and modern (organic material in the lake sediment that may be some decades old and thus contains bomb carbon). The large reservoir age of floating leaves clearly shows that the assimilation of atmospheric CO$_2$ is not the dominant carbon source for these specimens, despite the expectations. This result is certainly interesting for freshwater botanists. In the future, $^{14}$C dating may be used for identifying carbon sources in photosynthesis of aquatic plants in their natural environment, as long as this is an aquatic system with high water DIC ages.
It is striking that almost all modern river samples have large $^{14}$C ages. This indicates a substantial reservoir effect that would be even greater without the bomb effect. The $^{14}$C age of the contemporaneous atmosphere is around $-400$ BP. As can be seen from Tables 2 and 4, the estimated reservoir ages $R_f$, calculated considering the atmospheric $^{14}$C levels, are roughly 400 yr larger than the $^{14}$C ages BP. Furthermore, the organic matter in the rivers and their catchments may have a “true age” of several years to decades and thus contain bomb carbon. When CO$_2$ from this decaying organic matter is incorporated into the plant, $^{14}$C enrichment in the plants is the consequence. Due to the large variability, it is not possible to find a reservoir correction for any of the rivers.

In the archaeological material, the large spread of terrestrial ages emphasizes the importance of direct pottery dating. The age difference between terrestrial samples and fish bones indicates a substantial freshwater reservoir effect of $>1000$ yr in the Trave and $>3000$ yr in the Alster. Both for modern and archaeological samples, the reservoir age in the Alster is higher than in the Trave. Water hardness and cation ratios in the rivers would lead to the opposite expectation. This indicates that dissolved carbonate minerals are not the only source for the high reservoir ages. Mineralized organic matter or groundwater with high “real” ages could have increased the reservoir age of Alster water, while exchange with the atmosphere could have decreased the reservoir age in the Trave while it passed the shallow Wardersee.

In general, this study shows that the characterization of the reservoir effect in a freshwater system requires more than a few water, plant, or animal samples. It is shown that freshwater reservoir effects in rivers can be very high and strongly variable. Archaeologists should be made aware of this source of spurious ages when sending pottery or bones of omnivores like humans from inland sites for $^{14}$C dating. The surprisingly large ages of northern German pottery are in all likelihood caused by the freshwater reservoir effect. There is no longer reason to believe that inland potters were hundreds of years ahead of their colleagues at the coast.

ACKNOWLEDGMENTS

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