#### Neutralization of Osmium Tetroxide in Case of Accidental Spillage and for Disposal

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In October 1979, a technician in the electron microscopy laboratory of one of the teaching hospitals in Toronto spilled a bottle containing approximately 500 ml of 2% osmium tetroxide. He asked for help, and four other technicians were involved in the cleanup. They all were exposed to osmium tetroxide vapours for long enough to show signs of acute intoxication. They developed redness and swelling of the conjunctiva, impaired vision, chest pains, breathing difficulties and skin irritation and were taken into Emergency Care at the hospital. The occular symptoms disappeared after two days, but the chest pains and breathing difficulty persisted for approximately one week.

This accident could have had more serious consequences. It has been reported that exposure to osmium vapour could lead to blindness and serious respiratory trouble<sup>1,2</sup>. At McMaster University Medical Centre, the safety committee gives precise procedures to be followed in case of accidents with dangerous chemicals. All the occupants of the laboratory have to leave immediately and phone the safety and security team. The use of a self-contained breathing apparatus is advocated to re-enter the premises. Had thispolicy been followed in the Toronto hospital, the technicians would not have suffered the acute intoxication.

Recently the Government of Ontario issued an ordinance concerning disposal of dangerous chemicals. Osmium Tetroxide is amongst those, and can no longer be discarded thorough the sewage system. The ordinance, however, does not indicate how to dispose of it. For the time being, used osmium tetroxide is kept in the laboratory. This policy constitutes a hazard for the personnel. The accident in Toronto, and the new Ontario ordinance, prompted us to study ways of rapidly and effectively neutralizating osmium tetroxide.

Osmium tetroxide reacts particularly with unsaturated lipid. The oxidation of the lipid double bond leads to the formation of cyclic osmic acid monoester which is hydrolized to yield diol and osmic acid. The monoester osmic acid, being unstable, reacts with the diol producing a stable diester, crosslinking two double bonds from two unsaturated aliphatic chains<sup>3</sup>.

Consequently, the feasibility of effectively neutralizing large amounts of osmium tetroxide with unsaturated lipids was studied, and the stability of the resulting diester was ascertained. Other ways of neutralizing osmium tetroxide, for example with alcohols, and with sodium hydroxide, were also investigated.

#### Materials and Methods

Among easily available oils, corn oil has a high percentage of unsaturated bonds (54% of polyunsaturated lipids in the case of Mazola Corn Oil). The determination of the amount of oil necessary to neutralize a 2% aqueous solution of osmium tetroxide was made in test tubes. Osmium tetroxide and different quantities of oil were mixed together and the presence of vapour detected with either an oil-coated glass cover slip or filter paper suspended over the solution. Blackening of the oil indicated the presence of vapor. Complete neutralization was assumed when no blackening occurred.

The following different ways of applying oil to spilled osmium tetroxide were tested:

- 1. Spreading ouil over the pool.
- 2. Applying an oil-soaked cloth.
- 3. Applying oil-soaked absorbent granules (cat litter).

 Spraying oil over the pool to attempt to neutralize the vapour. In addition, a commercial sweeping compound (Dustbane) was tested and compared to the above procedure.

Finally, 2% aqueous osmium tetroxide was treated separately in test tubes with ethyl alcohol and sodium hydroxide.

Osmium tetroxide that had been treated with oil was mixed with each of the following chemicals to test the stability of the diester: 1N hydrochloric acid, 10N hydrochloric acid, 1N sulfuric acid, concentrated sulfuric acid, 50% chromic acid and hydrogen peroxide.

All tests were performed in a fume hood, taking all the necessary precautions.

#### Results

The amount of oil needed to fully neutralize a 2% aqueous solution of osmium tetroxide, was found to be twice the volume of the osmium solution, and this ratio was used in all the tests. Complete neutralization was indicated by the oil turning black in the test tube, and by the absence of vapour as shown by the lack of oxidised black oil on the glass cover slip or filter paper above the solution.

The spreading of oil over the spilled osmium tetroxide neutralizes the fixative almost instantly, preventing the formation of vapour. This method, however, is not very convenient because of the difficulty involved in cleaning it up. Very absorbent tissue or fabric has to be used to soak up the osmium/oil mixture. Likewise, applying an oil-soaked cloth, or spraying oil over the pool, effectively neutralized the osmium tetroxide, but was messy to clean up and actually increased the spreading of the spill

The most effective method of application was found to be the oil-soaked absorbent granules. 100 gm of granules absorbed 50 ml of corn oil, and 300 gm of this compound neutralizes very rapidly and effectively 50 ml of 2% osmium tetroxide. The mixture is easy to remove, and the area easy to clean. Absorbent granules soaked in oil were kept in tightly sealed bags for more than a month, and still retained their effectiveness.

The commercial sweeping compound was ineffective at neutralizing the osmium tetroxide and is not recommended. Sodium hydroxide gave only temporary neutralization, and ethyl alcohol had no apparent effect. Neither the acids nor the hydrogen peroxide reduced the stability of the diester.

#### Conclusion

The present study shows that unsaturated oil is effective in neutralizing osmium tetroxide. An aqueous 2% osmium tetroxide solution is neutralized by twice its volume of oil. The diester formed by the oxidation is very stable and not destroyed by acid. Consequently, it should be easier to find a safe method of disposal for osmium tetroxide in this form. One possibility is sanitary landfill, but this is yet to be confirmed.

Used osmium tetroxide contained in a bottle can be neutralized directly with oil. Spilled osmium tetroxide should, however, be neutralized with a compound consisting of oil-soaked absorbent granules. A 2% osmium tetroxide solution will be neutralized by six times its volume of this compound.

The following recommendations are made for handling osmium tetroxide:

1. For normal use, one should never prepare more than 50 ml of osmium tetroxide at a time.

The glass bottle containing the osmium tetroxide should be kept inside a padded plastic container.

3. Osmium tetroxide should always be handled in a fume hood.

 All personnel in a laboratory where osmium tetroxide is used should be aware of the danger involved in handling this chemical

 Enough oil-soaked absorbent granules should always be stored where osmium tetroxide is used.

6. In case of a major spill of osmium tetroxide, the personnel should immediately leave the premises and use a self-contained breathing apparatus to reenter the laboratory, and neutralize the osmium tetroxide.

 Used osmium tetroxide should always be poured into an oil-containing bottle in order to be neutralized.

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