

RADIOCARBON DATING OF CERAMIC MATERIALS: PROGRESS AND PROSPECTS

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ABSTRACT. Although initial studies in the 1960's established the feasibility of applying the radiocarbon technique directly to ceramics, subsequent analyses have revealed this category of materials to be more complex than previously believed. Consideration of the points at which various types of organic carbon are introduced into ceramic materials by natural or human agencies allows more effective screening of potential sherd samples. It also suggest ways to overcome some of the limiting factors involved in dating ceramics.

INTRODUCTION

The paucity of charcoal or other readily datable material in many archaeological sites often confounds the archaeologist's attempts to obtain absolute dates for his studies. In addition, even in cases where datable material can be found, its association with the cultural activity of interest is often suspect. This situation has prompted several researchers to turn their attention to other types of materials, such as ceramic artifacts. The advantages of such material are several. First, it provides direct dates on cultural material. Second, the association of dated material with the primary depositional context or the cultural activity of interest can be more readily evaluated. Third, ceramic materials, particularly utilitarian wares, are usually preserved in the form of sherds in many archaeological contexts, and are often expendable in quantities sufficient to produce a reliable radiocarbon date.

Initial experiments

Initial studies in the 1960's established the feasibility of applying the radiocarbon technique directly to ceramic materials. The operating assumption is that organic material present in the sherds represents only cultural activity, and that it was introduced in two ways: 1) organic tempering material might be added to the clay during the manufacture of the vessel, and 2) soot and food residue might be deposited in coarse wares during the use of the vessel. The method generally used to obtain a date involves crushing several kilograms of sherds to a fine size and treating them with hydrochloric acid to remove inorganic carbon. The residue is then burned and purified for CO₂ counting.

The first attempts utilized sherds tempered with various organic materials such as carbonized bark and charcoal (Ralph, 1959; Evans and Meggers, 1962; Stuckenrath, 1963). Of those samples that could be related to known cultures and/or associated dated material, only the sherds with extremely small amounts of organic carbon (<.6 percent) produced questionable dates. However, the use of organic tempering is not prevalent everywhere, and even when it is used, the organic material often burns away completely in firing. Consequently, later studies were undertaken

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to test the validity of ^{14}C determinations based on finely dispersed organic material in low-fired ceramics due to soot from cooking fires or to other organic matter not burned out of the clay during firing (Engstrand, 1965; Taylor and Berger, 1968; Taylor, Swain, and Berger, 1968). The correspondence between ^{14}C dates on ceramics and other independently dated associated material was very good, and investigators were able to conclude that ceramic materials could be reliably dated if ordinary precautions were exercised in selecting and pretreating sherds. The major drawback was the small organic carbon content of most sherds and the need to process large quantities of material.

Subsequent analyses performed on ceramics from the Pacific and southwestern United States revealed them to be a more complex category of materials than previously believed. Correspondence of sherd dates and associated archaeological dates was the exception rather than the rule.

An extensive series of sherds from Palau in the Philippine Islands was submitted to the UCLA laboratory by Douglas Osborne, along with a set of charcoal samples. Although these were not paired samples, they were expected to date events which could be placed relatively within the same temporal framework. The sherd dates were evaluated by cross-dating and stratigraphy, and by extrapolation from the charcoal dates (see table 1). While the charcoal dates fell within the expected range, the sherd samples were all far too old (Osborne, in press). The temporal range for the charcoal samples was from 285 ± 80 (UCLA-1762K) to 1800 ± 80 years BP (UCLA-1762I), while the sherd dates ranged between 3320 ± 80 (UCLA-1855H) to 8150 ± 80 years BP (UCLA-1855FF). This is a rather significant discrepancy for materials from the same cultural horizons.

The southwestern United States material was less consistent in terms of the results obtained (see table 2). As with the Palau sherds, the dates were based on the finely dispersed organic material present in the ceramics. While three sherd dates corresponded with the associated tree-ring dates and pottery seriations, the rest ranged from older than expected to more recent than expected, and even dated to >100 percent modern, indicating the incorporation of bomb carbon. This lack of consistency in the direction of error in the dates suggested that carbon from non-cultural sources may also be present in ceramic materials, and it may effectively dilute the age or otherwise contaminate the cultural sample.

Consideration of the steps involved in the manufacture and use of pottery can help to delimit the points at which non-cultural carbon of varying age may be incorporated into the ceramic material, and thereby, account for the discrepancies as well as the acceptable dates obtained. Knowing the sources of this carbon and the points at which it is introduced should facilitate more effective screening of potential sherd samples. It should also suggest ways to overcome some of the limiting factors involved in dating.

Sources of carbon in ceramic materials

A survey of ethnographic and ethnohistoric studies on primitive ceramic technology indicates that there are four points at which organic

carbon can be introduced that may not represent the time of manufacture or use of a vessel. Briefly, these are in the selection of clay, the selection of temper, the use of the vessel during its "lifetime", and in the post-use or depositional environment. Each of these points will be examined for their effect on the apparent age of the pottery.

Clay

Primitive potters use a wide range of clays, including many impure, low-grade varieties with high organic matter content. Some of these highly carbonaceous clays require long oxidation periods in firing, and fairly high temperatures to burn out the organic material. Consequently, oxidation is not complete in many low-fired wares, and a "carbon core" remains, bearing the finely dispersed organic matter in the clay source. Since the types of clays used by prehistoric potters vary geologically from recently formed to older shale deposits that are ground for use, the inherent age of the organic material in the carbon core may affect the age of the ceramics. Furthermore, the effect may vary according to the extent of the core throughout the ceramic body; vagaries in firing may result in sets of vessels which have been oxidized to different degrees.

Residual or primary clays in contact with igneous rock from which they are formed usually have the fewest organic impurities. Nonetheless, such clays are variable in this respect. In surface clays of recent origin, the organics may be decaying vegetal matter (Shepard, 1965; Lawrence, 1972). Although larger materials, such as leaves, would probably be picked out and discarded by the potter, finely divided humus from disintegration of root systems would remain (Eyre, 1968). Because these materials are roughly contemporaneous with the selection of clay and subsequent manufacture of the pot, it is unlikely that their presence in a carbon core would be problematic.

On the other hand, sedimentary clays may be extremely high in organics. The amount varies from about 1 percent in glacial and alluvial clays to about 10 percent in shales (Shepard, 1965; Lawrence, 1972). Since these clays have been transported and redeposited, along with many different types of impurities, the source and age of the constituent organics is highly suspect. Clays derived from shale deposits contain peat, lignite, and various forms of coal that would significantly "dilute" the age of the ceramics, making them seem older than expected. In this case, it is necessary to assess the range of clay sources potentially available to prehistoric potters and obtain samples to use as correction factors.

Some of the anomalous dates in tables 1 and 2 can be explained to some extent by this factor. In the southwestern series, UCLA-1931A comes from an area in northern Arizona where the clays used in ceramic manufacture are derived from shales (Hack, 1942). The Palau samples are also made of clay with older organic materials. Two types of environments, volcanic and limestone islands (Osborne, 1966) are represented by these samples. Although there is no assurance that only one clay source was used, the clay is probably entirely from the volcanic islands

(Osborne, in press; Hidikata, 1973). Osborne provided a raw clay sample from Palau, UCLA-1906, which gave a date of 1890 ± 400 years. When this date was used as a rough correction factor for the sherds dates, UCLA-1855H, J, K, S, and BB fell much closer to the expected dates. The lack of correspondence by the other samples may indicate the use of a different clay source. This should be more systematically examined.

Temper

As a category of material, organic tempers are the most suitable for dating because they usually consist of straw, grasses, ash, or other materials whose apparent age is contemporaneous with the manufacture of the pottery. However, fresh-water and marine shells have also been used in prehistoric pottery, but these would not necessarily represent the time of manufacture. In addition to the question of initial ^{14}C content of the shells, such carbonate tempers may be susceptible to CO_2 substitution

TABLE I
Palau ceramic dates

Lab no.	Sample description	Date BP	Associated date*
UCLA-1855H	Aulong, Au 1, Wall test St II, level 3. 0.43% organic C	3320 ± 80	Au 1 is a reef (limestone) island. Date should be UCLA-1762E (charcoal): 1480 ± 80
UCLA-1855I	Aulong, Au 1, Wall test, St III, level 4, 5. 1.10% organic C	6530 ± 80	Expect slightly older than UCLA-1762E
UCLA-1855J	Aulong, Au 1, Wall test St IV, level 6, 7, 8. 0.99% organic C	4970 ± 80	Should be oldest of the 3 samples. Older than expected, but may be OK
UCLA-1855K	Aulong, Au 1, F-E test, bottom, level 7, 8. 0.98% organic C	3780 ± 80	Older than expected
UCLA-1855Q	Babeldaob, B 37, Coconut Grove test St II, level 3. 0.70% organic C	4475 ± 80	Volcanic island, coastal site, UCLA-1762G (B37 Coconut Grove St II, levels 12-36", charcoal): 1055 ± 80
UCLA-1855S	Babeldaob, B 37, North Face test, level 2. 0.64% organic C	3800 ± 80	Volcanic terrace site. Should date in same range as UCLA-1762E: 1480 ± 80
UCLA-1855X	Babeldaob, B 19, level 2, 3. 1.03% organic C	7610 ± 80	Volcanic; ca 6500 years too old
UCLA-1855BB	Babeldaob, B 18, Altar test St I. 1.01% organic C	3760 ± 80	Volcanic. Should be younger than UCLA-1762I: 1800 ± 80
UCLA-1855FF	Babeldaob, B 18, Platform I, St II. 0.98% organic C	8150 ± 80	Volcanic. Should be close to UCLA-1855BB; ca 5000 years too old
UCLA-1906	Raw clay sample from Palau	1890 ± 400	

Comments: If 1890 ± 400 is subtracted from BP dates, UCLA-1855H, J, K, S, BB are close to the expected dates. There is greater coincidence for dates from limestone islands than from volcanic islands. All the clay is probably from the volcanic islands, but it is not certain that all these sherds were made from the same clay source.

Refs: Osborne, 1966; in press.

* Evaluation of dates and comments provided by Douglas Osborne.

after manufacture. When fired in the 650° to 898°C range, they can decompose, leaving CaO. Through time, the very fine CaO particles reunite with CO₂ from the environment, thereby introducing carbon of unknown origin, atmospheric or geologic (Shepard, 1965). Given the lack of control in firing, calcination is often incomplete. Therefore, such tempers may contain carbon from the original shell, from the prehistoric atmosphere, and from the depositional environment in unknowable proportions.

Use of the vessel

Most material that would be incorporated into the pores of ceramics during use would be foodstuffs or soot which would provide a good basis for obtaining an accurate result. However, there are exceptions. For example, post-firing treatment to render a vessel impervious may include coatings of asphaltum. Thus, care should be taken to exclude such treated vessels from the sample. Another exception pertains to the type of fuel used. In northern Arizona, coal has been documented as a prehistoric and ethnohistoric source of fuel (Brew, 1937; Hodge, 1904; Hack, 1942), and of course, in this case, soot is unreliable. UCLA-1931A may have been covered with coal soot.

Depositional contamination

Because low-fired ceramics can be especially susceptible to weathering and alteration, environments that promote such disintegration will facil-

TABLE 2
U S southwestern ceramic dates

Lab no.	Sample description	Date BP	Associated date
UCLA-1931A	Moenkopi corrugated ware, N Arizona	5600 ± 80	Tree-ring dates for ware: ca AD 1100-1300*
UCLA-1931B	Homolovi corrugated ware, Chevelon Ruin, N Arizona	455 ± 90	Expected date based on pottery seriation, ca AD 1400**
UCLA-1931F	Corrugated ware, Chevelon Ruin, N Arizona	690 ± 110	Expect ca AD 1440, based on pottery seriation**
UCLA-1931G	Cibola corrugated ware, Scribe S site, New Mexico	575 ± 60	Tree-ring dates for site average AD 1265. Occupation range, AD 1225-1280***
UCLA-1948B	Plain ware, site HS-1, Hidalgo Co, New Mexico	245 ± 80	UCLA-2122A: charcoal, site HS-1:490 ± 60 UCLA-2122B: charcoal, site HS-1:910 ± 60
UCLA-1948D	Plain ware, Pendelton Ruin, Hidalgo Co, New Mexico	>100% modern	Expect AD 1300-1400, based on ceramic seriation & obsidian hydration values
UCLA-1948F	Plain ware, HS-15, Hidalgo Co, New Mexico	>100% modern	UCLA-2122G: charcoal, HS-15: 910 ± 80

* Mc Gregor, 1965.

** Martin and Plog, 1973.

*** S Le Blanc, 1978, personal commun.

itate incorporation of non-cultural carbon. In tropical environments, sherds may become impregnated with more recent organic matter (Shepard, 1965). However, this problem is not universal because none of the Amazon dates reported by Ralph (1959) and Stuckenrath (1963) were so affected. In the Palau series, sherds from volcanic contexts were extremely altered while those from the limestone islands were not (Osborne, in press). Table 1 shows that of the corrected dates, those from the limestone islands showed greater correspondence with expected ages (3 of 4) than did those from the volcanic islands (2 of 5). This suggests that in areas of volcanic activity, ceramics may be more susceptible to incorporation of dead CO₂ than in other areas. Finally, in the southwestern series (table 2), the dates that show too recent ages or contamination by bomb carbon were from very shallow sites, and had long been subjected to severe mechanical alteration and surface flooding from annual rains. In general, weathered sherds are unsuitable for dating.

CONCLUSION

Because obtaining reliable ¹⁴C dates on ceramic material is not always straightforward, systematic screening of potential samples is in order. Consideration of the patterns of ceramic manufacture and use for the samples involved, and of the environment in which sherds are found will substantially aid in this process.

Sherds with carbon cores are quite prevalent archaeologically, and it is not always possible to obtain a sample of the raw clay from which the original vessels were made. An alternative process is temperature-controlled burning. A short firing at 500 to 800°C should be sufficient to clear loose carbon from the pores of the ceramic. Original carbon in the clay will require longer burning at a higher temperature (Shepard, 1965; Rubin, personal communication). Collecting CO₂ fractions liberated at different temperature increments should provide a means of assessing the dilution by non-cultural carbon, much as analyzing humic acid fractions on regular ¹⁴C samples does.

The final problem is sample size. The percentage of organic carbon in sherd samples is usually very low. Although accurate dates can usually be obtained with about 1 percent carbon, the quantity of sherds needed (often up to several kilograms) can be prohibitive. Fortunately, developments in accelerator dating should soon reduce the importance of this limiting factor, and routine, reliable, direct dating of ceramic materials appears to be more practical than ever before.

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