Cubic boron nitride (c-BN) is a promising host material in various optoelectronic applications such as photoemitters, scintillators and light-emitting diodes in the UV or visible range. Owing to its excellent hardness (next to diamond) and strong covalent bonding nature, it has been extremely difficult to effectively dope functional impurities. Recently, crystal growth by using reactive flux with the temperature gradient method at high pressure and high temperature (5.5 GPa, 1773 K) has achieved stable doping of luminous rare-earth (RE) atoms into c-BN host lattice, and we have firstly produced c-BN:RE single crystals with visible light luminescence (blue-colored luminescence in Fig. 1(a)) [1,2]. To understand the doping mechanism and emergence of luminous properties, the fundamental question remains as to how RE dopants with a huge size mismatch can be accommodated within high-density c-BN crystal lattices. Here we show the direct determination of the exact atomic sites and valence states of individual Ce dopants buried inside c-BN single crystals using atomic-resolution scanning transmission electron microscopy (STEM) and electron energy-loss spectroscopy (EELS) [2].

Figure 1(b) shows a typical annular dark-field (ADF) STEM image of c-BN:Ce viewed along the [110] direction. Randomly distributed bright spots highlight the Ce containing atomic columns, which clearly reveals that the Ce atoms are not form Ce-clusters but are distributed in the c-BN lattices as single atoms. In the higher magnified image (Fig. 1(c), (d)), the sub-angstrom BN dumbbells (0.9 Å) are clearly resolved, and moreover it is found that all the cationic Ce dopants substitute for the exact N anti-site (CeN), not for the cationic B site (Fig. (e)). The valence state of those single Ce atoms was investigated using column-by-column EELS (Fig. 1(f)). By comparison to the EELS spectra of Ce3+ in Ce2O3 and Ce4+ in CeO2, the valence state of the Ce atoms is determined to be 3+, which is doubly checked by $M_4/M_5$ ratio with second derivative spectrum.

To determine the stable Ce atom configurations in c-BN lattice, we consider all the possible defect structures and evaluated their relaxed structures, charge states and the formation energies using hybrid Hartree-Fock density functional calculations. Consequently, we found that the Ce dopants are incorporated by introducing the complex point defect structure of (CeN-4VB)6−: Ce atom is located at anti-site, surrounded with four B vacancies and the defect is negatively charged. Our findings indicate that even extremely large size mismatch can be stably doped into host lattice by means of the formation of complex point defect structures.

References
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Figure 1. (a) Cathodoluminescence optical image of c-BN:Ce single crystal. (b) Typical and (c) atomic-resolution ADF STEM image of c-BN:Ce viewed along the [110] direction and simulated image (inset). (d) Enlarged ADF STEM image with Ce dopants. (e) Intensity profiles obtained along X-X' and Y-Y'. (f) Core-loss spectrum obtained from a single nitrogen atom column containing one Ce atom.