A glacio-chemical characterization of the new EPICA deep-drilling site on Amundsenisen, Dronning Maud Land, Antarctica

Fidan GÖKTAS,1 Hubertus FISCHER,1 Hans OERTER,2 Rolf WELLER,1 Stefan SOMMER,2 Heinz MILLER1
1Alfred Wegener Institute for Polar and Marine Research, Columbusstrasse, P.O. Box 120161, D-27515 Bremerhaven, Germany E-mail: fgoktas@awi-bremerhaven.de
2Climate and Environmental Physics, Physics Institute, University of Bern, Sidlerstrasse 5, CH-3012 Bern, Switzerland

ABSTRACT. In the framework of the European Project for Ice Coring in Antarctica (EPICA) a glacio-chemical pre-site survey was carried out in Dronning Maud Land (DML), Antarctica, to investigate seasonal and spatial variations. All ion species show pronounced seasonal cycles with the exception of nitrate, which is subject to post-depositional alterations. Sea salt reaches maximum concentrations in late winter/spring, while sulphate, being mainly of marine biogenic origin, shows a double peak with high concentrations both in autumn and in late spring/summer. Methanesulphonate (MSA) also shows a strong autumn peak but only slight indications of a second peak in late spring/summer, as seen for sulphate. Due to post-depositional changes, the seasonal cycle of MSA vanishes further down in the firm. These changes are also reflected in the spatial distribution of MSA. While surface MSA concentrations decline with altitude and higher accumulation rates, concentrations of aged snow show a strong increase with higher accumulation rates in our ice cores. Non-sea-salt sulphate shows a 40% decrease with an increase in snow accumulation of about 80% in recent and aged snow. While the geographical variation is negligible for average nitrate concentrations, sea salt shows an exponential decline with altitude. The outcome of this study confirms that the data of the new EPICA deep drilling site in DML (75°00.10’S, 0°04.07’E) will be representative for this region, and high-resolution analytical methods will allow accurate stratigraphic dating of a deep ice core.

INTRODUCTION

Of the various natural climate archives (such as tree rings, corals, sea and lake sediments) ice cores play a unique role since they store information not only about temperature and precipitation but also about atmospheric composition and transport (Stauffer, 1999). To obtain such long-term records, several deep ice-core drilling projects have been carried out in Greenland and Antarctica. Two new deep cores will be drilled in the framework of the European Project for Ice Coring in Antarctica (EPICA). One of those drillings will take place in Dronning Maud Land (DML), in the Atlantic sector of the East Antarctic plateau (Fig. 1a). This region is mainly influenced by air masses originating from the Southern Ocean (Noone and others, 1999; Reijmer and others, in press). Accordingly, the DML deep core is expected to provide the first South Atlantic counterpart to the Greenland ice-core records over the last glacial cycle and may give crucial information about the climate coupling of both hemispheres across the Atlantic.

In order to study the glacio-chemical and glacio-meteorological characteristics of DML and to corroborate the interpretation of the coming deep ice-core record, a comprehensive pre-site survey has been carried out in this still unexplored area (Oerter and others, 1999, 2000). Up to now, chemical analyses have been performed on samples of four intermediate deep ice cores and fourteen snow pits taken during the 1997/98 and 1999/2000 field campaigns (Fig. 1b).

Fig. 1. Map of DML: (a) overview of Antarctica from 90°W to 90°E, indicating the study area; (b) enlargement of the study area showing ice-core drilling (■) and snow-pit (●) locations (Oerter and others, 1999, 2000).
Intermediate deep ice cores were drilled and snow-pit samples taken during the pre-site surveys in DML (Fig. 1b). The cores at DML03, DML05, DML07 and DML17 were sampled for chemical studies in 2 cm resolution (referred to below as high-resolution samples). With respect to the annual snow accumulation at the individual sites this corresponds to a resolution of approximately 5–9 samples per year. The high-resolution samples cover the time period 1945–90. The deeper parts of the cores were subdivided into one sample per year (referred to below as low-resolution samples) according to the records measured by continuous flow analysis (CFA), which show distinct annual cycles (Sommer and others, 2000a).

All ice-core samples were thoroughly decontaminated under clean-room conditions using a contamination-free electromechanical plane (Fischer and others, 1998). The decontaminated samples were packed in pre-cleaned PE bags, which had been rinsed with ultra-pure water. To check contamination during handling, ultra-pure water samples (referred to below as process blanks) were prepared, frozen and processed like the cores in the processing routine. Additionally, IC vials were filled with ultra-pure water to quantify the remnant ion concentrations of the ultra-pure water itself and the contamination introduced during the analysis (referred to below as vial blanks). After decontamination, all samples were melted under clean-room conditions and stored in pre-cleaned PE vials for analysis and as archive material.

### METHODS

#### Sampling

Intermediate deep ice cores were drilled and snow-pit samples taken during the pre-site surveys in DML (Fig. 1b). The cores investigated here were drilled at sites DML03, DML05, DML07 and DML17 and are 150, 115 and 130 m long, respectively, covering 1600–2000 years. At DML03 a firm core 41 m long was drilled, covering >200 years. Because of the poor core quality of the uppermost metres of the cores, data for these depth intervals were taken from snow pits excavated at the drill sites. Unfortunately, the snow pit at DML03 was too shallow, causing a gap of 4 years (1990–93). Additionally, one snow pit (SS9908) located at the new EPICA deep drilling site in DML was sampled with high resolution in the 1999/2000 field season.

Snow pits were sampled using 60 mL polyethylene (PE) beakers. All beakers were rinsed with ultra-pure water (σ > 18 MΩ cm) until the electrical conductivity of water stored in the beakers dropped to values lower than 0.5 μS cm⁻¹. After cleaning they were sealed in PE bags and only opened directly in the snow pit. The beakers, having a diameter of 44 mm, were pushed into the snow-pit wall at depth intervals of 30–40 mm, thus slightly overlapping each other. For snow pit SS9908 snow in defined 2 cm increments was collected in pre-cleaned 250 mL PE beakers using a pre-cleaned spatula. The beakers were closed very carefully and sealed again in PE bags for transportation back to Germany. The snow pits from the 1997/98 campaign cover a time period of 4–14 years back in time starting from December 1997; the two from 1999/2000 cover 10 (SS9908) and 11 years (SS9901), starting from December 1999.

Ice cores were packed in PE bags in the field and transported to Neumayer station. All core material and pit samples were shipped in freezer facilities to Bremerhaven, Germany. In our cold laboratory the upper 7–10 m of the cores at DML03, DML05, DML07 and DML17 were sub-sampled for chemical studies in 2 cm resolution (referred to below as high-resolution samples). With respect to the annual snow accumulation at the individual sites this corresponds to a resolution of approximately 5–9 samples per year. The high-resolution samples cover the time period 1945–90. The deeper parts of the cores were subdivided into one sample per year (referred to below as low-resolution samples) according to the records measured by continuous flow analysis (CFA), which show distinct annual cycles (Sommer and others, 2000a).

### Analysis

During the 1997/98 field season, CFA measurements were carried out on the core samples from DML03, DML05, DML07 and DML17 for Na⁺, Ca²⁺, NH₃⁻, H₂O₂ and electrolytical conductivity at Neumayer station (Sommer and others, 2000a, b). Additionally, IC analyses for methanesulphonate (MSA), Cl⁻, NO₃⁻, SO₂⁻ were performed at Neumayer station on samples taken from several of the snow pits (SS9903, SS9907, SS9905, SS9900). All other snow-pit and core samples were performed on Dionex 500 systems using an isotropic method for cations and a gradient method for anions equipped with a Dionex CS21 and a Dionex AS11 separator column, respectively. The cation system ran in the auto-suppression mode, while the anion system used an external ultra-pure water supply for the regeneration chamber of the suppression unit.

Average blank concentrations for 159 process blank samples and 449 vial blanks are listed in Table 2. Laboratory and process blank concentrations were below the lowest cali-
bration levels and distinctively below the sample concentrations for all anion species. Comparison of process and vial blanks shows that any additional contamination introduced by the decontamination routine in the cold laboratory is negligible. Because of these low blank values, no corresponding blank correction was necessary for the presented ion concentrations. The analytical error of the IC measurements is better than 10% for all sample concentrations which are well above the blank level. However, despite the extremely low blank levels, the lower end of the sample-concentration range overlaps with the process blank values for NH$_4^+$, K$^+$, Mg$^{2+}$ and Ca$^{2+}$, significantly increasing the analytical error of such samples. In the following, only anion and Na$^+$ concentrations are discussed.

The sodium concentrations measured in high resolution by IC and CFA agree very well with each other (see Fig. 2). Average sample concentrations of both methods are equal on the 95% confidence level. Slight differences in the depth axes between the two methods can be attributed to variability in the flow velocity of the CFA and/or small losses of ice during the cutting of discrete samples. At certain depths the CFA shows gaps (Fig. 2), where data suspected of contamination had been removed from the CFA dataset. However, remeasuring these intervals in high resolution using IC proved these high values to be uncontaminated ion concentrations.

### Dating

A preliminary dating by dielectric profiling was carried out, showing that the volcanic eruption of Tambora, Indonesia, in 1815 could be clearly detected in all cores. Accordingly, dating based on radioactive horizons of nuclear weapon tests in the 1950s and 60s, as used in an earlier study in this area (Oerter and others, 1999), was not performed here. Stratigraphic dating was accomplished by Sommer and others (2000a) using a combination of annual-layer counting in the CFA records and identification of the most prominent historic volcanic horizons in the electrolytical conductivity profile. The latter are mainly related to high concentrations of sulphuric acid in the respective snow layers. In general, this dating was adopted for this study. Only slight corrections were made according to our high-resolution snow-chemistry profiles: for the high-resolution and snow-pit samples annual markers were set at the falling flank of Na$^+$ and the rising flank of non-sea-salt (nss) sulphate (calculated by subtracting the sea-salt contribution from the total sulphate concentration according to $[\text{nssSO}_4^{2-}] = [\text{SO}_4^{2-}] - 0.252[\text{Na}^+]$, indicating the spring season (Fig. 3). Stable-isotope profiles were not considered because of resolution constraints and diffusional smoothing of the isotope record. Snow pit SS9908, which was sampled for isotope and chemistry analyses in high resolution (2 cm), was the only one dated by marking the summer maximum in $\delta^13C$ for each year (Fig. 4b).

Dating of the cores using the high-resolution CFA records was reliable within ±5 years (Sommer and others, 2000a).
over the last millennium. Dating of the DML05 core, however, had to be modified at one point. Sommer and others (2000a) assigned a peak in the conductivity record at 6.12 m depth to the eruption of Mount Agung, Indonesia, in 1963, which was also observed at other Antarctic sites (Delmas and others, 1985, 1992; Isaksson, 1994). However, the coherent increase of MSA and nss-sulphate at this depth implies that the assigned peak in the DML05 core is mainly caused by biogenic sulphate input and not by a volcanic eruption. Additionally, an utmost prominent double peak in the sea-salt components was found in all cores, as indicated in Figure 3. This feature could only be aligned for all cores within the dating accuracy by shifting the eruption of Mount Agung in DML05 to another peak in the sulphate and conductivity profiles at 6.77 m depth.

RESULTS AND DISCUSSION
Seasonal variation

The high-resolution records of the chemical components in snow and ice show consistent seasonal variations in all three

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**Fig. 3.** (a) Concentration of sodium, (b) Cl/Na⁺ ratio, (c) concentrations of nitrate, (d) nss-sulphate and (e) MSA vs depth for the snow pit (SS9805) and core (B32) at DML05. The snow-pit data cover the top 2.38 m, the high-resolution sequence of core DML05 the interval from 2.38 down to 8.52 m. The arrows indicate the prominent double peak in sea-salt concentration occurring in all three cores. The vertical lines mark the time between the falling flank of Na⁺ and the rising flank of nssSO₄²⁻, indicating the spring season.
cores and snow pits. An example of these seasonal cycles is shown in Figure 4 for the high-resolution snow pit SS9908 and in Figure 3 for snow pit SS9805 (depth 0–2.58 m) and core B32 (2.58–9 m) at DML05. Figure 4b also shows the results for the isotopic signature of the ice acting as proxy thermometer. Automatic weather stations (AWS) deployed in central DML (Reijmer, 2001) reveal a strong temperature maximum in December/January, and very cold conditions ranging from May to October. AWS snow-height measurements indicate that a few major precipitation events were responsible for most of the annual snow accumulation. For instance, in 1998 four snowfall events contributed about 80% of the annual snow accumulation (Reijmer and Van den Broeke, 2001). In principle, this is supported by modeling studies (Noone and others, 1999; Reijmer, 2001). Despite these event-like precipitation characteristics, δD shows a well-developed seasonal cycle, with maxima corresponding to the temperature maxima in summer and minima in winter. Due to the diffusional smoothing of the isotope record in the firm, however, single precipitation events cannot be distinguished in the δD record, and dating of the seasonal cycles in ion concentrations relative to the isotope record is only accurate to a few months. In the following, we distinguish only between the four major austral seasons, not referring to specific months.

In Figure 4a, sodium, which together with chloride is mainly derived from sea-salt aerosol, shows maximum ion concentrations on the rising flank of the δD record, indicating a sea-salt maximum in late winter/spring. This points to enhanced cyclonic activity over the South Atlantic region connected to higher transport of sea-salt aerosol onto DML during late winter/spring, despite the larger sea-ice coverage at this time of year. Higher storminess in this season is also indicated by faster air-mass transport in trajectory studies by Reijmer and Van den Broeke (2001).

MSA and nss-sulphate, which in Antarctica are essentially derived from the marine biogenic production of dimethyl-sulphide (DMS) (Legrand, 1995; Stenberg and others, 1998), show a prominent maximum on the declining flank of the δD (Fig 4d and c) record equivalent to maximum concentrations in autumn. This is significantly later than the distinct biogenic sulphur maximum observed at the coastal Antarctic station Neumayer (Minikin and others, 1998) in summer when sea ice retreats for a short time. Such a phase shift may be attributed to the different source areas of biogenic sulphur relevant for DML compared to Neumayer (NM). While NM is more influenced by DMS production in high-latitude ocean areas relatively close to NM, trajectory studies for DML show that air masses originate as far away as 35°S 5 days back in time (Reijmer and others, in press). In this area, biogenic productivity is extended to March/April (Minikin and others, 1998) in line with our MSA and nss-sulphate peak at that time of the year.

However, nss-sulphate also shows a second peak during late spring/summer in the high-resolution snow-pit record (Fig. 4c) which is only occasionally connected to elevated MSA levels. Due to the lower resolution of the core data compared to data from snow pit SS9908, the double peak in nss-sulphate concentrations is not always resolved in the cores. Based on aerosol measurements of the cosmogenic radio-isotopes 10Be and 7Be performed at NM (Wagenbach and others, 1998), a stratospheric input of sulphate during summer seems very unlikely. Therefore, only biogenic sources appear to be able to explain such a sulphate input. A post-depositional migration of the MSA peak away from high-acidity layers (e.g. high sulphuric acid concentrations), which has been observed in ice cores from other Antarctic sites (Ivey and others, 1986; Minikin and others 1994; Pasteur and Mulvaney, 2000), cannot be clearly seen in Figure 4d. However, post-depositional smoothing of the MSA record is observed below a few metres depth in Figure 3e.

Post-depositional effects were also observed for nitrate which originates from lightning-induced NO formation and intrusions of stratospheric air masses (Wolff, 1995; Wagenbach and others, 1998). In contrast to results from firn cores retrieved at high-accumulation sites showing maximum nitrate concentrations in summer (Minikin and others, 1994; Hou and others, 1999), it was not possible to detect any seasonality in our measured nitrate profiles. This can be attributed to a substantial post-depositional nitrate loss, which degrades any initially existing seasonal signal. The nitrate loss is up to 40%, which is estimated from the 40% lower mean nitrate concentration in core B32 compared with the mean concentration in snow pit SS9805 (upper 2.58 m).

Spatial distribution

The dependence of average ion concentrations on the major geographical and glacio-meteorological parameters in the
DML region is shown in Figure 5. To distinguish surface snow and older firn strata, which might have been subject to post-depositional alterations, and to improve the spatial resolution, average ion concentrations both in the snow pits (for the common time period 1997–98) and in the ice cores (over the time period 1950–1865) are plotted in Figure 5. Data for the same time period (1950–1865) from a DML ice core investigated by Isaksson (1994) are also plotted in Figure 5.

All our drill sites are located on the Antarctic plateau and cover an altitude range between 2669 m (DML07) and 3606 m (DML17) (Fig. 1b). At the same time the average annual snow accumulation covers a range from 89 kg m$^{-2}$ a$^{-1}$ (DML03) to 47 kg m$^{-2}$ a$^{-1}$ (DML17) (Table 1), with generally lower accumulation rates found at higher altitudes. Lower accumulation rates were also found on the ice side of the main ice divide (DML07 in Fig. 1b). Note that the snow pits cover an altitude range between 2630 m (SS8902) and 3606 m (SS8910) (Fig. 1b) but are limited in the time-span shared by all snow pits to only 4 years. Thus, the temporal representativeness of the pit averages (Fig. 5) is limited.

The long-term core averages in nss-sulphate show no clear relationship with altitude but a strong decline with higher accumulation rates (Fig. 5c). The recent snow-pit data essentially show the same feature, but with a somewhat higher concentration level reflecting the different time periods covered.

Such a spatial decrease of sulphate concentration with higher snow accumulation can be expected from the dilution of dry-deposited sulphate aerosol by higher precipitation rates. Using the linear dependence of sulphate concentration on inverse snow accumulation as given by Fischer and others (1998), a total dry-deposition flux of approximately 240 ng cm$^{-2}$ a$^{-1}$ can be deduced. This implies that about three-quarters of the total sulphate flux is due to dry deposition at the future EPICA drill site in DML (75°00.10′ S, 00°04.07′ E) with a long-term accumulation rate of 62 kg m$^{-2}$ a$^{-1}$.

In Figure 5d and f, MSA shows an exponential decrease with higher altitude and higher snow accumulation for the snow-pit data. However, in Figure 5f for the long-term core averages reflecting older firn the opposite relationship with snow accumulation is found. While the snow-pit data of MSA and nss-sulphate show a similar dependence on a change in accumulation rate, the opposite relationship was observed in the core data. Considering that the source of both MSA and the main part of nss-sulphate is the photo-oxidation of DMS (Legrand, 1995; Saltzman, 1995), similar transport and deposition mechanisms, and thus a similar geographical distribution, are expected. One reason for a difference in the spatial distribution of MSA and sulphate in the ice-core averages could be related to the different atmospheric residence times of MSA and nss-sulphate with their gaseous precursors in the atmosphere. The photo-oxidation mechanism of DMS is highly complex (Yin and others, 1990). In short, MSA is produced via short-lived transient intermediates, while formation of sulphate occurs via the more stable SO$$_2$$, which considerably enlarges the effective atmospheric residence time of this branch of DMS oxidation. Thus, MSA concentrations may be subject to stronger depletion along an air-mass trajectory over the DML region. However, a 50% decline in the MSA ice-core averages appears to be far too large to be accounted for by this effect. A more important consideration seems to be that a difference in atmospheric residence time cannot explain the different spatial distribution of snow-pit and ice-core MSA data.

In view of the inconsistency between snow-pit and ice-core averages and considering the very large size of the spatial MSA decrease in the ice-core data in this geographically rather uniform area, only an accumulation-rate-dependent post-depositional loss of MSA could readily explain a net decrease of up to 50%. Such a loss of MSA has been reported from the low-accumulation site Vostok, East Antarctica, by Wagnon and others (1999), who, however, pointed out that a re-evaporation of particulate MSA from the snowpack is difficult to achieve. Additionally, such a loss cannot be unambiguously detected in our pit profiles. There are somewhat higher MSA concentrations in the uppermost metre, from 1992 to 1997 (see Figs 4d and 5c). However, higher MSA concentrations in these years may also be attributed to extraordinarily high DMS concentrations, possibly connected to El Niño events as proposed by Legrand and Feniet-Saigne (1991) for South Pole snow and by Isaksson and others (2001) for Amundsenisen. Taking into account that intervals with higher MSA concentrations are also repeatedly found in deeper layers, the question of the existence of a post-depositional net loss of MSA remains open.

The sea-salt-derived components chloride (Fig. 5a) and sodium show an exponential decline with altitude with an e-folding height of approximately 1800 and 1300 m, respectively. The larger scaling height of chloride compared to sodium may reflect the additional transport of gaseous HCl onto the...
ice sheet, produced by an acid-induced release of HCl from sea-salt aerosol (Legrand and Delmas, 1988). Such a HCl con-
tribution is also supported by the existence of significant nss-
chloride concentrations in the upper snow layers on the
plateau. This effect is most prominent during summer, when
biogenic sulphur species cause a substantial acidification of
the atmospheric aerosol as reflected in summer peaks in the
Cl/Na+ ratio in our high-resolution records (Fig. 4a). The
geographical differences in sea-salt aerosol essentially levels
out on top of the DML region (Fig. 5a). Sea-salt concentra-
tions in a core investigated by Isaksson (1994), covering the
same time period as the cores studied here, are somewhat
lower than our ice-core data. Note, however, that the con-
centrations of all other ion species agree well between the
DML cores studied here and the one studied by Isaksson
(1994). The lower sea-salt concentrations in the core studied
by Isaksson (1994) can most likely be explained by the
removal of very large concentration values in that core
which were suspected to be contaminated (personal com-
munication from E. Isaksson, 2001).

Nitrate shows a rather uniform spatial distribution in Fig-
ure 5b, with significantly higher nitrate concentrations in
recent snow pits compared to the older ice-core data. We attri-
bute this finding to post-depositional nitrate loss (Legrand
and Delmas, 1983; Wolff, 1995; Mulvaney and others, 1998),
which levels the ion concentrations. The post-depositional
loss in nitrate is discussed in detail by Rothlisberger and oth-
ers (2002). A post-depositional nitrate loss of the size observed
by Rothlisberger and others (2002) at ultra-low-accumulation
sites like Dome Concordia (from 1000 ppm at the surface
down to 15 ppm), however, is not observed in our cores. Nitrate
concentrations in surface snow samples taken during the
2000/01 season at DML05 reach 263 ppm (personal communi-
cation from C. Piel, 2001), whereas average concentrations of
about 56 ppm (at DML03), 47 ppm (at DML05), 53 ppm (at
DML07) and 45 ppm (DML17) were found in the DML cores.
The average nitrate concentrations at Dome C (Legrand and
Delmas, 1986) after post-depositional alteration are less than
a third of the values found in DML. Even taking into account a
higher post-depositional loss of nitrate at Dome C, where the
accumulation rate is approximately half of that in DML, the
size of this decrease is extremely large. Accordingly, factors
other than snow-accumulation rate, such as temperature, snow
formation and recrystallization in the snowpack or wind speed,
may additionally affect the post-depositional nitrate loss.

CONCLUSION

Seasonal and spatial variations of chemical snow para-
eters in DML have been discussed based on ion chromato-
graphic analyses of four intermediate deep ice cores
covering the last 140 years and 14 snow pits covering the
time period 1994–97.

The seasonal and spatial variations show that the area
under investigation in DML is well suited for the reconstruc-
tion of long-term chemical records from the new EPICA ice
core. The snow-accumulation rate is large enough to ensure
seasonal information while at the same time making it pos-
sible to reach far into the past. The seasonal signal is well
archived for sulphate and sea-salt components, allowing
for an accurate stratigraphic dating of the new deep core,
the minimum for the Holocene period.

The variation in the spatial distribution of most chemical

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