Erythrocytes, erythrocyte membranes, neutrophils and platelets as biopsy materials for the assessment of zinc status in humans

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During a controlled zinc depletion-repletion study, fifteen men aged 25.3 (SD 3.3) years were fed on a low-Zn diet with high phytate:Zn and phytate:x calcium: Zn molar ratios for 7 weeks, followed by a 2 week repletion period when 30 mg supplemental Zn/d was given. Changes in plasma, urine, and hair Zn concentrations, taste acuity, and cellular immune response confirmed the development of mild Zn deficiency. Zn concentrations in neutrophils, platelets, erythrocytes and erythrocyte membranes, mean platelet volume, and activities of alkaline phosphatase (EC 3.1.3.1) and a-D-mannosidase (EC 3.2.1.24) in neutrophils did not respond to changes in Zn status. In contrast, alkaline phosphatase activity in erythrocyte membranes showed a significant decline which was consistent in all subjects (nmol product formed/min per mg protein; baseline v. 7-week Zn depletion, 0.656 (SD 0.279) v. 0.506 (SD 0.230), at 7 weeks; P < 0.05); neutral phosphatase activity remained unchanged. Alkaline phosphatase activity in erythrocyte membranes may be a potential index of Zn status in humans.

Experimental zinc deficiency: Humans: Phytate: Neutrophils: Platelets

The importance of zinc for human nutrition has been recognized for about 25 years. During this time many important advances in the understanding of the role of Zn in human metabolism have been made. Nevertheless, the diagnosis of mild Zn deficiency in the population is still an unsolved problem (Solomons, 1979; Prasad, 1985a, b).

In adult humans less than 0.5% of the total body Zn content is in the blood, of which 12–22% is in plasma, approximately 3% in leukocytes and platelets, and 75–88% is in erythrocytes. The most widely used biopsy material for the assessment of Zn status is blood, of which plasma or serum is the most popular component (Solomons, 1979; Prasad, 1985b). Some (Meadows et al. 1981; Prasad & Cossack, 1982), but not all (Milne et al. 1985a, b), investigators have suggested that concentrations of Zn in mixed leukocytes or specific sub-populations of leukocytes (i.e. neutrophils and lymphocytes) are more reliable indices of Zn status than plasma Zn detecting ‘short-term’ changes in Zn status, because of their relatively short life-span (Prasad & Cossack, 1982; Prasad, 1985b). In contrast, erythrocytes have a slow turnover and hence their Zn concentration does not reflect recent Zn status; whether the Zn concentration of total erythrocytes, or of the erythrocyte membrane, or both, reflect ‘long-term’ Zn status remains unclear (Prasad et al. 1978; Rabbani et al. 1987). The activities of selected Zn metallo-enzymes in neutrophils and

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erythrocytes from subjects with some degree of Zn deficiency have also been investigated; results have also been equivocal (Ballester & Prasad, 1983; Prasad, 1985a; Milne et al. 1987).

Previously we reported changes in plasma, urine, and hair Zn concentrations, taste acuity and cellular immune response during an experimentally controlled Zn depletion–repletion study, which confirmed the development of mild Zn deficiency in fifteen young adult men (Ruz et al. 1991). In the present study we have evaluated the responses of selected variables in erythrocytes, neutrophils, and platelets as potential indices of Zn status during the experimentally induced mild Zn deficiency state, and during subsequent Zn repletion. The variables assessed were: Zn concentrations in erythrocytes, erythrocyte membranes, neutrophils, platelets, platelet count, mean platelet volume (MPV), and the activities of alkaline (EC 3.1.3.1) and neutral phosphatases in erythrocyte membranes, and α-D-mannosidase (EC 3.2.1.24) and alkaline phosphatase in neutrophils. This is the first time that α-D-mannosidase activity in neutrophils and changes in Zn-related erythrocyte membrane variables have been monitored during an experimentally controlled Zn depletion–repletion study in humans. The responses of the static indices (i.e. concentrations of Zn in neutrophils, platelets, and erythrocytes) were evaluated each week to provide information on weekly variations.

SUBJECTS AND METHODS

Experimental design

The study was approved by the Human Ethics Committee of the University of Guelph. A detailed description of the subjects, experimental design, and dietary manipulations of the Zn depletion–repletion study has been published earlier (Ruz et al. 1991). Briefly, fifteen young adult males (mean age 25.3 (SD 3.3) years) attending the University of Guelph participated in the 16-week study. The study had four phases: equilibration, Zn depletion, Zn repletion, and follow-up. The first week of the study (equilibration) comprised an adaptation period to the soya-bean-protein- and egg-albumin-based diets to be used throughout most of the depletion phase of the study. The following 7 weeks comprised the Zn-depletion phase. In the first week of this period subjects received a semi-purified liquid diet based on egg albumin containing 0.6 mg Zn/d, added phytic acid (1.27 g/d), and a calcium supplement (Caltrate; Lederle, Montreal, Quebec). As a result the average phytate:Zn and phytate × Ca:Zn molar ratios of the semi-purified liquid diets were 209 and 4116 respectively. During the rest of the 6 weeks of the depletion phase subjects received the egg-albumin- and soya-bean-protein-based diets used in the equilibration phase, modified to contain approximately 4 mg Zn/d, and average phytate:Zn and phytate × Ca:Zn molar ratios of 58 and 1853 respectively. For the 2-week Zn repletion phase subjects received these non-meat diets, but with the daily addition of a Zn supplement (30 mg Zn/d in the form of zinc gluconate). Total Zn intake during the 2-week repletion period was 35 mg/d. Vitamin and mineral supplements were also given to the subjects throughout the study period to ensure that their nutrient intakes (with the exception of Zn) met the recommended intakes for Canadians (Health and Welfare, 1990).

Throughout the first 10 weeks of the study all foods and beverages were supplied by the researchers, who maintained strict control of all items consumed. Subjects were informed that Zn concentrations in plasma and urine, and urinary creatinine would be assessed weekly to monitor compliance with the study protocol.

The final 6 weeks of the study comprised the follow-up phase in which the subjects consumed their usual self-selected diets plus an additional Zn supplement of 5 mg Zn/d as zinc gluconate.
INDICES OF ZINC STATUS IN HUMANS

Measurements

Fasting blood and 24 h urine samples were obtained at weekly intervals from week 1 to week 10, with an additional sample at week 16 for thirteen of the fifteen subjects, and analysed for selected indices of Zn status. Scalp hair samples taken at the end of the baseline (week 1), depletion (week 8), and follow-up (week 16) were also analysed for Zn. In addition, selected physiological functional tests of Zn status (i.e. cellular immune response and taste acuity) were evaluated. Results for some of these determinations have been presented in detail elsewhere (Ruz et al. 1991) and, where necessary, are summarized briefly here. The present paper focuses on the findings in platelets, neutrophils, erythrocytes and erythrocyte membranes.

Following an overnight fast, 30 ml fasting blood was taken with subjects in the sitting position in trace element-free evacuated tubes (Vacutainer; Becton Dickinson, Rutherford, NJ, USA) containing, where appropriate, either Zn-free and preservative-free heparin, EDTA, or ACD (25 g trisodium citrate/l, 14 g citric acid/l, 20 g dextrose/l) as an anticoagulant. Blood samples taken using heparin were used for tests described previously (Ruz et al. 1991). Blood samples taken with EDTA were used for routine haematology analyses, which included MPV, and platelet, leukocyte and lymphocyte counts performed using an automated method (Coulter counter; Coulter Electronics Ltd, Nortwell Drive, Luton, Beds, UK). All reagents were analysed for Zn content before use by atomic absorption spectrophotometry (AAS); they all contained negligible amounts of Zn.

Separation of platelets, neutrophils, erythrocytes and erythrocyte membranes

Platelets and neutrophils were isolated as follows: 7–10 ml blood were collected in a tube containing 1-5 ml ACD and centrifuged for 20 min at 300 g. Plasma was transferred carefully, to avoid contamination with erythrocytes, to a 15 ml centrifuge tube and centrifuged for 15 min at 1530 g. Platelet-poor-plasma (ppp; 2 ml) was saved for further use during the neutrophil separation. The pellet was visually inspected and, if erythrocytes were detected, 4 ml cold distilled deionized water was added for 30 s, followed by 4 ml sodium chloride (18 g/l) and centrifugation for 15 min at 1530 g. The pellet was suspended in 3 ml PBS buffer, pH 7.4 (containing, g/l: NaCl 7.1, disodium hydrogen phosphate 1.15, potassium dihydrogen phosphate 0.2, potassium chloride 0.2). A 20 µl portion was diluted in 25 ml Isotone II (Coulter Electronics of Canada, Burlington, Ontario) and platelets counted using a Coulter counter. The remaining platelet suspension was kept at –70º until further analysis.

Packed erythrocytes and buffy coats were transferred to a 15 ml tube and 1 ml Dextran T-500 reagent (60 ml/l) prepared in saline (9 g NaCl/l) was added (Pharmacia, Dorval, Quebec), followed by enough saline to produce a total volume of 10 ml. The mixture was then gently mixed and allowed to stand undisturbed for 45 min. The supernatant fraction was removed and centrifuged for 6 min at 280 g. The pellet was retained and resuspended in 8 ml ppp–saline reagent (prepared by mixing 2 ml ppp and 8 ml saline). This suspension was carefully loaded into a tube containing 3 ml Ficoll-paque (Pharmacia, Dorval, Quebec) and centrifuged for 25 min at 750 g. The pellet was retained. Cold distilled deionized water (4 ml) was added to lyse the remaining erythrocytes and, after no more than 30 s, 4 ml NaCl (18 g/l) was added to restore isotonicity. The mixture was centrifuged for 6 min at 280 g and the supernatant fraction was discarded. If, after visual inspection, erythrocytes still remained, the lysis procedure was repeated. The pellet was suspended in 2 ml PBS buffer, pH 7.4 and centrifuged for 6 min at 280 g. The supernatant fraction was discarded and the pellet re-suspended in 2 ml PBS buffer and centrifuged for 6 min at 280 g. The pellet was again suspended in 2 ml PBS buffer. A portion (50 µl) of the suspension was diluted in
25 ml Isotone II solution and the neutrophil counts were performed using a Coulter counter; the remaining neutrophil suspension was kept at −70° until further analysis.

Erythrocytes were obtained by centrifugation of whole blood for 15 min at 300 g. Plasma was separated. The packed erythrocytes were washed with isotonic phosphate buffer, pH 8·0 (145 mM-NaCl–5 mM-Na2HPO4), and centrifuged for 15 min at 300 g; this procedure was carried out three times.

Erythrocyte membranes were prepared according to the method of Steck et al. (1970), and frozen at −70° until analysed. For analysis, erythrocyte membranes were allowed to thaw and were sonicated for 15 s in a sonic dismembrator (Model 300; Fisher Scientific Co, Ottawa, Ontario). A 10 μl portion was used for protein determination according to the method of Markwell et al. (1978).

Determination of Zn concentrations in the isolated blood components
Before analyses, a wet ashing procedure using concentrated nitric acid (Ultrex, JT Baker Chemicals Co, Phillipsburg, NJ, USA) (Clegg et al. 1988) was employed to ash portions of platelets, neutrophils, erythrocytes, and erythrocyte membranes. Ash was then dissolved in nitric acid (30 ml/l) and Zn determined by flame AAS (Varian model SpectrAA-30; Varian Techtron Ltd, Georgetown, Ontario) using a micro-sampler. Accuracy and precision of the method were assessed by ashing and analysing five samples of bovine liver (National Institute of Standards and Technology, Reference Material no. 1577), certified for Zn. The mean for five determinations was 122·6 (SD 5·0) μg/g, compared with the certified value of 130 (SD 13) μg/g. Zn concentrations in platelets and neutrophils were expressed as nmol/1010 cells, and μmol/1010 cells respectively, as μmol Zn/g haemoglobin for erythrocytes (Nishi, 1980), and as μmol Zn/g protein for erythrocyte membranes.

Determination of enzyme activities in neutrophils
Neutrophils were analysed for the activities of alkaline phosphatase and acidic α-D-mannosidase. The neutrophil suspension was thawed and sonicated two to four times for 15 s in a Fisher sonic dismembrator Model 300. Enzyme activities were determined immediately because of reported loss of activity during the pre-incubation period (De Chatelet & Cooper, 1970).

The activity of neutrophil alkaline phosphatase was measured by a modification of the method by De Chatelet & Cooper (1970). Briefly, 50 μl sonicate was incubated for 60 min in a reaction mixture containing buffer (0·1 M-2-amino-2-methyl-1-propanol, pH 10·5) 500 μl p-nitrophenylphosphate (4 mg/ml), 125 μl, distilled deionized water 125 μl; magnesium (from magnesium chloride) was present at a final concentration of 4 mM. Reaction was stopped by the addition of 500 μl 0·5 M-sodium hydroxide. Tubes were centrifuged for 15 min at 300 g. The supernatant fraction was carefully removed and absorbance read at 405 nm against a reactive blank in a spectrophotometer.

The activity of neutrophil acidic α-D-mannosidase was measured by an adaptation of the method of Faber & Glew (1983). A portion (50 μl) of neutrophil sonicate was incubated for 90 min in a reaction mixture containing: 0·5 M-acetate buffer (pH 4·5) 500 μl, 10 mM-4-nitrophenyl-α-D-mannopyranoside 200 μl, distilled deionized water 150 μl. The reaction was stopped by adding 500 μl 0·13 M-glycine–NaOH buffer, pH 11. Tubes were centrifuged for 15 min at 300 g. The supernatant fraction was removed and absorbance read against a reactive blank at 405 nm in a spectrophotometer (model DU-50; Beckman Instruments Inc, Jamboree, Irvine, CA, USA). Precision of the methods was assessed by analysing six samples of a pool of neutrophils divided into individual portions. The means for six
determinations were 402.9 (SD 20.5) nmol product formed/h per mg protein for alkaline phosphatase, and 184.8 (SD 8.8) nmol/h per mg protein for acidic α-D-mannosidase. Protein was determined by the method of Markwell et al. (1978).

**Determination of enzyme activities in erythrocyte membranes**

Total alkaline phosphatase activity in erythrocyte membranes (AP-EM) was determined as follows: 100 μl membrane suspension was incubated for 90 min at 37° in a reaction mixture containing: 0.1 M 2-amino-2-methyl-1-propanol buffer (pH 10.5) 500 μl, p-nitrophenylphosphate (4 mg/ml; Sigma Chemical Co., St Louis, MO 63178, USA) 125 μl, distilled deionized water 75 μl; magnesium (from MgCl₂) was present at a final concentration in the cuvette of 4 mM. Reaction was stopped by the addition of 500 μl 0.5 M NaOH. Tubes were centrifuged for 15 min at 300 g to remove the turbidity that develops after the addition of NaOH. The supernatant fraction was carefully removed and absorbance due to the formation of p-nitrophenol was read at 405 nm against a reactive blank in a spectrophotometer (Beckman model DU-50). The blank contained the same components described previously except in this case the NaOH solution was added before the sample. Neutral phosphatase activity in erythrocyte membranes was determined using a 50 μl sample incubated for 90 min at 37° in a mixture containing: 2-hydroxymethyl-1,3-propanediol (Tris) buffer (pH 7.4) 500 μl, p-nitrophenylphosphate (4 mg/ml) 125 μl, distilled deionized water 125 μl; Mg was also present at 4 mM. The rest of the procedure was performed in a manner similar to that described for alkaline phosphatase activity. Precision of the methods was evaluated by analysing a pooled sample (saved frozen in several portions on six occasions. Means (nmol product formed/min per mg protein) for the six determinations were 0.523 (SD 0.018) for alkaline phosphatase activity (coefficient of variation (CV) 3.4%) and 1.605 (SD 0.088) for neutral phosphatase activity (CV 5.5%).

**Statistical analysis**

All variables were tested for normality by the Kolmogorov-Smirnov test before any statistical comparisons were computed. Analysis of variance of repeated measures was performed to assess the significance of observed changes during the study. Contrasts were used to identify specific differences over time. Pearson’s product-moment correlation coefficients were computed to assess relationships among selected variables (Snedecor & Cochran, 1980). P ≤ 0.05 was considered to be significant. All statistical analyses were carried out using the Statistical Package for the Social Sciences, SPSSx (SPSS Inc., 1986). All results are expressed as means and standard deviations.

**RESULTS**

The compliance during the study was very satisfactory; fourteen of the fifteen subjects completed the 7-week Zn-depletion phase. The remaining individual received the low-Zn diets for 6 weeks, at which point measurements were considered to be ‘at the end of Zn depletion’. This same individual successfully completed the 2 weeks of Zn repletion. Thirteen subjects volunteered to continue in the study during the 6 weeks of the follow-up phase.

Significant (P < 0.05) changes were observed in mean plasma (μmol/l) and urinary Zn (μmol/d): baseline 97.0 (SD 10.9) and 8.0 (SD 2.7), depletion 80.1 (SD 13.4) and 4.3 (SD 2.3), repletion 100.8 (SD 13.6) and 8.2 (SD 3.1) respectively, and in taste acuity (0.05 < P < 0.10) and cellular immune responses (P < 0.05), emphasizing that marginal Zn deficiency had
Table 1. Neutrophil zinc concentration (μmol/10^10 cells) and activities (nmol product/h per mg protein) at 37°C of α-D-mannosidase (EC 3.2.1.24) and alkaline phosphatase (EC 3.1.3.1) in neutrophils during experimental Zn depletion and repletion in young adult male subjects.

(Mean values with their standard deviations)

<table>
<thead>
<tr>
<th>Experimental period</th>
<th>Week of study</th>
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<th>α-D-mannosidase</th>
<th>Alkaline phosphatase</th>
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<tr>
<td></td>
<td></td>
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<td>Mean</td>
<td>Mean</td>
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<td>7</td>
<td>1.15</td>
<td>212.6</td>
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<td>212.6</td>
<td></td>
</tr>
<tr>
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<td>1.37</td>
<td>212.6</td>
<td></td>
</tr>
<tr>
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<td>1.31</td>
<td>212.6</td>
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</tbody>
</table>

* For details of procedures, see pp. 516-519.

been successfully induced by the dietary regimen of the present study. However, the activities of alkaline phosphatase, angiotensin-1-converting enzyme (EC 3.4.15.1) and acidic α-D-mannosidase in plasma did not change. The responses of these biochemical and physiological functional indices of Zn status have been presented in detail elsewhere (Ruz et al. 1991).

In contrast, the concentrations of Zn in neutrophils (Table 1) did not show any significant trends during the study. Similarly, no significant responses to the dietary manipulations were noted for the activities of α-D-mannosidase and alkaline phosphatase in neutrophils (Table 1). The mean platelet Zn, total circulating platelet count, and MPV were also unaffected by the dietary treatments in the present study (Table 2), as were both the total leukocyte and lymphocyte counts reported earlier (Ruz et al. 1991).

The high dispersion of values observed at the different weeks of the study for some of the variables, particularly those related to neutrophils, is noteworthy. A visual inspection of values can lead to inaccuracies in the evaluation of the biological variation of the results, and CV are a more objective method of comparison. CV (%) were: MPV 13.1, platelet Zn concentration 25.7, neutrophil Zn concentration 37.3, neutrophil alkaline phosphatase activity 50.0, neutrophil α-D-mannosidase activity 40.4.

With few exceptions, no significant associations were observed among the variables analysed at baseline. These exceptions were neutrophil α-D-mannosidase activity v. MPV (r = 0.60, P < 0.05), and neutrophil Zn concentration v. total leukocyte count (r = 0.78, P < 0.05). None of the variables studied in platlets and leukocytes was correlated with initial plasma Zn concentrations.

Table 3 presents the Zn concentrations in erythrocytes and erythrocyte membranes observed during the study. No significant changes were evident, although Zn concentrations in erythrocyte membranes tended to decrease as the experimental period progressed (P = 0.06). The results for the weekly Zn concentrations in erythrocytes emphasize the
Table 2. Platelet zinc concentration (nmol/10⁹ cells), mean platelet volume (fl) and platelet count (× 10⁹/l) during experimental Zn depletion and repletion in young adult male subjects* (Mean values with their standard deviations)

<table>
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<tr>
<th>Experimental</th>
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<th>Platelet Zn</th>
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<th>Mean platelet volume</th>
<th>Mean</th>
<th>sd</th>
<th>Platelet count</th>
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<td>sd</td>
<td>Mean</td>
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<td>13.3</td>
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* For details of procedures, see pp. 516–519.

Table 3. Concentration of zinc in erythrocytes (µmol/g haemoglobin) and erythrocyte membranes (µmol/g protein) during experimental Zn depletion and repletion in young adult male subjects* (Mean values with their standard deviations)

<table>
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* For details of procedures, see pp. 516–519.

consistency of the values throughout the study, unlike those observed for neutrophils and platelets (Table 1).

In Table 4 the activities of the two enzymes assayed in erythrocyte membranes are presented. Total neutral phosphatase activity did not change significantly during the study. In contrast total AP-EM significantly decreased during the Zn-depletion phase (P < 0.01), followed by a slight (non-significant) increase during the Zn-repletion period. No further changes during the follow-up phase were noted. Fig. 1 presents the total change (baseline
Table 4. Total alkaline (EC 3.1.3.1) and neutral phosphatase activities (nmol product/min per mg protein) at 37° in erythrocyte membranes during experimental Zn depletion and repletion in young adult male subjects*

(Mean values with their standard deviations)

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<td>Mean  ± SD</td>
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<td>1.84 ± 0.24</td>
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<td>7</td>
<td>0.506 ± 0.230</td>
<td>1.83 ± 0.20</td>
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<td>2</td>
<td>0.529 ± 0.246</td>
<td>1.82 ± 0.23</td>
</tr>
<tr>
<td></td>
<td>Follow-up</td>
<td>0.523 ± 0.215</td>
<td>1.84 ± 0.23</td>
</tr>
</tbody>
</table>

*a, b, c* Mean with unlike superscript letters were significantly different: \( P < 0.05 \).

* For details of procedures, see pp. 516–519.

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![Graph showing changes in total alkaline phosphatase activity](https://www.cambridge.org/core)

Fig. 1. Changes in total alkaline phosphatase (EC 3.1.3.1) activity in erythrocyte membranes (expressed as percentage of the initial value) during experimental zinc depletion and repletion in young adult male subjects. For details of procedures, see pp. 516–519.

—end of depletion) of AP-EM expressed as a percent of the initial value. The decrease in AP-EM was consistent in all fifteen subjects, averaging 23%, and ranging from 3-7 to 44.9%. The total decrease in AP-EM correlated significantly with the activity of this enzyme at baseline \( r = 0.64; P = 0.01 \), and negatively with initial urinary Zn concentration \( r = -0.68; P < 0.05 \). No significant associations were observed when the total change of AP-EM was correlated with: total change in plasma Zn, total change in urinary Zn, total
change in erythrocyte membrane Zn, initial plasma Zn, initial erythrocyte membrane Zn and initial total erythrocyte Zn concentration. The addition of Zn in vitro (5–100 μM final concentration in the cuvette) did not significantly affect the activity of AP-EM.

**DISCUSSION**

*Indices of Zn status in neutrophils*

The conventional approach when using mixed leukocytes, or specific sub-populations of leukocytes, to assess Zn status has been to determine the concentration of Zn in these cells (Prasad & Cossack, 1982; Milne *et al.* 1985a, b; Prasad, 1985a, b). This approach, however, has produced equivocal results.

There are important technical considerations regarding the separation of blood cells for mineral analysis. Mean Zn concentrations in neutrophils and lymphocytes (separated simultaneously using Histopaque–Ficoll-hypaque gradients) in studies conducted in 1980–83 were between 104 and 116 μg/10⁶ cells (Prasad & Cossack, 1982; Whitehouse *et al.* 1982; Ballester & Prasad 1983), whereas results provided by the same laboratory 5–8 years later were about 50% lower for both cellular types (Pai & Prasad, 1988). This discrepancy is partly the result of the observations by Milne *et al.* (1985a), who (using Percoll gradients) reported that the separation procedure used earlier had yielded lymphocyte preparations contaminated with platelets which artificially increased the Zn content. The effect of the separation procedure on the neutrophil fraction is less clear. In a density-gradient system typically used to separate platelets, lymphocytes and neutrophils, cells become separated in this same sequence from the top to the bottom of the test-tube (Milne *et al.* 1985a). As a result, the risk of contamination with platelets is high for the lymphocyte fraction but low for neutrophils, making it difficult to understand the claim by some investigators (Pai & Prasad, 1988) that improved isolation procedures were responsible for the reductions in the ‘normal values’ for both lymphocytes and neutrophils. In our study, we initially attempted to separate the cellular types using Percoll gradients, but we observed that this reagent contained Zn as a contaminant (0.32 μg/ml). Consequently, we used a Ficoll–Hypaque gradient, but did not attempt to isolate the lymphocyte fraction to avoid contaminating the other cell types with platelets.

The effects of Zn status on concentrations of Zn in neutrophils are equivocal (Prasad *et al.* 1978; Prasad & Cossack, 1982; Ballester & Prasad, 1983; Milne *et al.* 1987; Rabbani *et al.* 1987). The formation of neutrophils from their earliest stages of development until their release into the bloodstream takes approximately 8–9 d. Their half-life in the circulation pool is about 7 h (Simmons, 1989). Cells are removed from this pool randomly rather than according to their age as is the case of erythrocytes and platelets (Wintrobe *et al.* 1981). Consequently, if it is assumed that neutrophils are sensitive to alterations in Zn status, a short period of 2–4 weeks should, theoretically, be sufficient to detect changes in neutrophil variables. Such changes in neutrophil Zn were not observed during the 7-week Zn-depletion period of the present study, despite the development of mild Zn deficiency, as indicated by the changes observed in plasma, urinary and hair Zn concentrations, immune function and taste acuity (Ruz *et al.* 1991).

As neutrophil Zn concentrations were analysed on a weekly basis in the present study, changes in their responses to the experimental diets could be monitored very carefully. For example, when all the eleven weekly measurements were included in the ANOVA of repeated measures, no significant differences were apparent. In contrast, if only the baseline and the end of depletion points were compared by using a paired *t* test, the resulting *P* value was 0.04. Such a simplistic approach fails to recognize some inconsistencies during the
study; for instance, the lowest mean value observed (seventh week of depletion) was preceded by a value closer to that at the baseline rather than that obtained at the end of depletion.

In the present study we also assessed the response of two Zn metallo-enzymes in neutrophils (α-d-mannosidase and alkaline phosphatase) as potential indices of Zn status in humans. To our knowledge, this is the first study to assess the response of α-d-mannosidase in neutrophils during experimentally induced Zn depletion–repletion in humans. Several species of α-mannosidase exist in mammalian tissues and fluids, and are classified according to their optimum pH (Snaith, 1977). Only acid α-mannosidase is said to be Zn-dependent (Snaith, 1977). It is located in the lysosomal fraction and has an optimum pH in the range of 4.0–4.6. Our findings (Table 1) indicate that the activity of this enzyme in neutrophils (and plasma) is not a useful index of Zn status. In addition, no significant changes were observed in the activity of neutrophil alkaline phosphatase in our study. Some (Shrader & Hurley, 1972; Baer et al. 1985), but not all (Prasad & Cossack, 1984; Schiliro et al. 1987) investigators have also failed to show that the activity of alkaline phosphatase in neutrophils or leukocytes, or both, is a valid index of Zn status.

Indices of Zn status in platelets

The use of platelets as a biopsy tissue for the assessment of Zn status has been advocated by some investigators (Pai & Prasad, 1988). Others, however, have failed to document any significant changes in platelet Zn concentrations during experimental Zn depletion in human (Milne et al. 1987) or animal (Gordon & O'Dell, 1980; Milne et al. 1985b) studies. Our results support these latter observations and emphasize the insensitivity of platelet Zn concentrations to changes in Zn status. Values for platelet Zn concentrations reported here (Table 2) are within the range for healthy adults (29.5–73.3 nmol Zn/10¹⁰ cells (Milne et al. 1985a, 1987; Pai & Prasad, 1988).

Some functional defects in platelets have been described in Zn-deficient animals (Gordon & O'Dell, 1980) and humans (Gordon et al., 1982), some of which (i.e. impaired platelet aggregation) have been related to a reduction in MPV (Ralston & Milne, 1988). In our study, MPV in whole blood did not show any significant change in response to a 7-week period of Zn depletion. No explanation can be given at present for the significant correlations observed at baseline between MPV and neutrophil α-d-mannosidase, and between neutrophil Zn and total circulating leukocyte count. More research is necessary to elucidate the nature and significance (if any) of these associations.

Indices of Zn status in erythrocytes

Erythrocyte Zn concentrations do not reflect short-term changes in Zn status; evidence for their use as a valid index of Zn status in medium- to long-term studies of controlled Zn deficiency (i.e. 6 or more weeks) is controversial. Earlier observations indicated some degree of sensitivity of erythrocyte Zn to fluctuations in Zn status (Buerk et al. 1973). More recent reports of human studies, based on results obtained after feeding low-Zn diets for periods ranging from 4 to 28 weeks (Prasad et al. 1978; Baer & King, 1984; Milne et al. 1987; Rabbani et al. 1987), indicate that erythrocyte Zn is unchanged during controlled Zn depletion. This finding is consistent with that noted in animals with severe Zn deficiency (Milne et al. 1985b). Consequently, the lack of response of erythrocyte Zn concentration as an index of Zn status reported here is not unexpected. In the present study, inter-subject variation in initial erythrocyte Zn concentrations was high but intra-subject variation was low.

The erythrocyte membrane was also investigated as a potential tissue for the assessment of Zn status in the present study, utilizing both a static (i.e. Zn concentration) and a
functional (i.e. alkaline phosphatase) index of Zn status. The response of erythrocyte membrane Zn concentration to Zn deficiency has been investigated in rats (Bettger & Taylor, 1986; Johanning et al. 1989) and pigs (Johanning et al. 1989), but not in humans. In our study, erythrocyte membrane Zn concentrations were slightly higher than levels reported by McWilliams et al. (1983) in patients with taste and smell dysfunction. Moreover, although erythrocyte membrane Zn concentrations tended (not significantly) to decrease with time ($P = 0.06$), the fall was only 6% after 7 weeks of the low-Zn diets, whereas at the end of the follow-up phase, erythrocyte membrane Zn concentrations had fallen by 20%. This slow response in erythrocyte membrane Zn concentration limits the use of erythrocyte membrane Zn concentration as an index of Zn status.

The activity of several membrane-bound enzymes is apparently affected by Zn deficiency in a variety of tissues, although information on the erythrocyte membrane is limited. For instance, the activity of 5'-nucleotidase (EC 3.1.3.5), an enzyme suggested to be a Zn-metallo-enzyme (Johanning et al. 1988), is significantly reduced in erythrocyte membranes in pigs and rats consuming low-Zn diets (Johanning et al. 1988). Important differences in the activity of 5'-nucleotidase in erythrocyte membranes among species exist, for instance, the activity of this enzyme in erythrocyte membranes from Zn-adequate pigs can be up to thirty times higher than that in erythrocyte membranes from normal rats (Johanning et al. 1988). Furthermore, the activity of Ca, Mg-ATPase in erythrocyte membranes was reduced in pigs consuming low-Zn diets, but not in Zn-deficient rats (Johanning et al. 1988). In our study, 5'-nucleotidase activity was not detected in erythrocyte membranes, an observation consistent with that of Delaunay et al. (1978).

In the present study we also developed a functional test to evaluate Zn status based on the simultaneous determination of the total alkaline phosphatase and neutral phosphatase activities in erythrocyte membranes. The activity of alkaline (Galdes & Hill, 1979), but not neutral phosphatase (W. J. Bettger, personal communication) is a Zn-dependent function. Therefore, different patterns should be observed during experimental Zn deficiency if Zn status is truly reflected by the Zn-dependent portion of the phosphatase activity. In our study, AP-EM, but not the plasma AP activity (Ruz et al. 1991), showed a consistent decrease during the Zn-depletion phase; neutral phosphatase activity was unaffected. Furthermore, the change in AP-EM activity was significant after only 4 weeks of low-Zn diets, and the magnitude of change was positively correlated with baseline activity. These findings suggest that AP-EM may have potential as an index for diagnosing mild deficiency in humans. During the Zn-repletion phase, AP-EM was not significantly increased, probably because a 2-week period was not sufficient to restore the activity. Nevertheless, it is noteworthy that eleven of the fifteen subjects had increased AP-EM during the 2 weeks of the Zn-repletion phase (Fig. 1).

The reasons for the results for the activity of AP-EM at week 16 (follow-up) are unclear. During this period the subjects were consuming their own self-selected diets. Furthermore, the mechanisms involved in the observed changes in total AP-EM during the Zn-depletion and Zn-repletion periods remain to be investigated. The enzyme responsible for these changes must be purified to confirm that it is truly a Zn metallo-enzyme. It we make this assumption, two possible mechanisms may explain the changes observed. In the first, alkaline phosphatase may lose Zn to counteract the decrease in plasma Zn, followed by degradation of the apo-enzyme by proteinases. Consequently, a return to the original activity must await de novo synthesis. The second possibility is that Zn deficiency decreases the amount of the holo-enzyme anchored to the membrane. If the enzyme is an endo-enzyme, it would move to the cytosol where it remains or is degraded; if the enzyme is an ecto-enzyme, then it would be released into the plasma. In both cases, a return of the activity to the original levels must await de novo synthesis.
In summary, we have documented significant changes in AP-EM during an experimentally-controlled Zn-depletion study. Comparable changes were not observed for the activity of erythrocyte membrane neutral phosphatase and for alkaline phosphatase and α-d-mannosidase activities in neutrophils. The mechanisms underlying the observed changes in AP-EM, however, warrant further investigation before confirming its use as a functional index of Zn status in humans. In the present study we have also monitored the D-mannosidase activities in neutrophils. The mechanisms underlying the observed changes in hair Zn concentrations, taste acuity and cellular immune response. Hence, they are not during experimentally-controlled Zn depletion and repletion. Results show that none of these indices detected the alterations in Zn status indicated by changes in plasma, urine, hair Zn concentrations, taste acuity and cellular immune response. Therefore, they are not sensitive indices of Zn status. Similarly, the concentration of Zn in erythrocyte membranes is not sensitive to short-term changes in Zn status; its usefulness as an index of long-term Zn status remains to be determined.

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REFERENCES
INDICES OF ZINC STATUS IN HUMANS


